

# **RF Active Compositions for Use in Adhesion, Bonding and Coating**

## ***Cross Reference to Related Application***

5 The present application is a continuation-in-part of application no. 09/404,200, filed September 23, 1999, which is a continuation-in-part of application no. 09/270,505, filed March 17, 1999, which claims the benefit of U.S. provisional application no. 60/078,282, the contents of each of which are fully incorporated by reference herein.

10 This patent application is related to the following co-pending U.S. utility patent application: "Radio Frequency Heating System," application no. 09/270,507, filed March 17, 1999, the contents of which are incorporated by reference herein.

## ***Background of the Invention***

### ***Field of the Invention***

15 This invention relates generally to the use of media containing ionic compounds and/or nonionic compounds with high dipole moments as a radio frequency (RF) susceptors in RF activated systems.

### ***Related Art***

20 Radio frequency (RF) heating is a well established non-contact precision heating method that is used to generate heat directly within RF susceptors, and indirectly within materials that are in thermally conductive contact with RF susceptors. RF susceptors are materials that have the ability to couple and convert RF energy into heat energy within the material.

25 Conventional adhesives are not suitable RF susceptors that can be directly heated and activated by RF heating. Rather, these conventional adhesives are

typically heated indirectly through thermally conductive contact with an RF susceptor material. FIG. 1 illustrates two conventional methods that are currently used in industry for indirect RF heating of conventional adhesives: The first method is illustrated in FIG. 1A, where susceptor material 102 exists as a bulk macroscopic layer. RF susceptor material 102 is directly heated by RF energy, and adhesive layer 104 is indirectly heated through thermally conductive contact with RF susceptor material 102. For example, adhesive layer 104 may be applied to a continuous surface of susceptor material 102, such as steel or aluminum. The second method is illustrated in FIG. 1B, where susceptor material 112 consists of discrete macroscopic particles. Adhesive layer 114 is loaded with macroscopic particles of a RF susceptor material 112, such as macroscopic particles or flakes of metal oxides, metallic alloys, or aluminum. With this conventional method, each RF susceptor particle 112 acts as a discrete RF susceptor, generating heat throughout adhesive layer 114.

An example of a conventional RF energy activated composition, such as that shown in FIG. 1B, is described in U.S. Patent No. 5,378,879, issued to Monovoukas ("Monovoukas"). Monovoukas utilizes macroscopic "loading particles" as discrete RF susceptors. The particles are heated by RF energy and in turn conduct heat to the surroundings. These macroscopic loading particles are thin flakes (i.e. in thin disk-like configuration) that are designed to be admixed to relatively thick extruded materials. However, these flakes are not well suited for use as susceptors in thin film bonding applications in which physical distortions, discolorations in the surface, or opacity of the bonded films would result from the flakes.

Another example of a conventional inductively activated adhesive is described in U.S. Patent No. 3,574,031, issued to Heller *et al.* ("Heller"). Heller describes a method of heat welding thermoplastic bodies using an adhesive layer that contains uniformly dispersed macroscopic RF susceptors, typically iron oxide particles. These discrete RF susceptor particles are ferromagnetic in nature. A disadvantage of this type of method is that a tradeoff must be made between the size of the particle employed versus the power level and duration of the inductive

heating process. For example, if susceptor particles are kept small in size, the mechanical strength of the bond tends to increase. However, as the size of these discrete susceptors is reduced, the power levels and dwell times required to heat the RF susceptor material and achieve acceptable bonds tend to increase. Another disadvantage of this type of method is the high levels of loading of the medium with RF susceptor particles that is required for efficient activation. Such high loading levels detract from the physical properties and rheology of the adhesive composition. Still another disadvantage is the dark color and opacity of the composition, which renders the composition undesirable for many applications.

An example of adhesive activated by a dielectric process is described in U.S. Patent No. 5,661,201, issued to Degrand ("Degrand"). Degrand describes a thermoplastic film including at least one ethylene copolymer and a sufficient quantity of N,N-ethylene-bisstearamide that is capable of being sealed utilizing a current at a frequency of about 27.12 megahertz (MHz). A disadvantage of this type of film and sealing process is the inherent tendency to also heat the adherand.

U.S. Patent No. 5,182,134, issued to Sato, discloses methods of curing a thermoset composition by applying an RF signal having a frequency of about 1 to 100 MHz to a composition comprising a major portion of a thermoset and a receptor. The receptor is described as being one of the alkali or alkaline earth metal sulfate salts (e.g. calcium sulfate), aluminum trihydrate, quaternary ammonium salts, phosphonate compounds, phosphate compounds, polystyrene sulfonate sodium salts or mixtures thereof. According to this patent, all of the exemplified compositions took longer than one second to heat.

U.S. Patent No. 5,328,539, issued to Sato, discloses methods of heating thermoplastic susceptor compositions by applying an RF signal having a frequency of about 1 to 100 MHz. The susceptors are described as being one of the alkali or alkaline earth metal sulfate salts (e.g. calcium sulfate), aluminum trihydrate, quaternary ammonium salts, phosphonate compounds, phosphate compounds, polystyrene sulfonate sodium salts or mixtures thereof. According to this patent, all of the exemplified compositions took longer than one second to heat.

U.S. Patent No. 4,360,607, issued to Thorsrud, discloses a composition suitable for sensitizing thermoplastic compositions to the heating effects of microwave energy comprising (1) an alcohol amine or derivative thereof, (2) a simple or polymeric alkylene glycol or derivative thereof, (3) silica and, optionally,  
5 (4) a plasticizer.

U.S. Patent No. 5,098,962, issued to Bozich, discloses a water dispersible hot melt adhesive composition comprising:

- (a) from about 40% to 95% by weight of a water dispersible ionically substituted polyester resin having a molecular weight from about  
10 10,000 to about 20,000 daltons;
- (b) from about 60% to about 5% by weight of one or more compatible plasticizers; and
- (c) from about 0.1% to about 1.5% of one or more compatible stabilizers of the anti-oxidant type.

15 Examples of plasticizers that may be used according to this patent include one or more low molecular weight polyethylene glycols, one or more low molecular weight glycol ethers, glycerin, butyl benzyl phthalate and mixtures thereof.

U.S. Patent No. 5,750,605, issued to Blumenthal *et al.*, discloses a hot  
20 melt adhesive composition comprising:

- (i) 10 to 90% by weight of a sulfonated polyester condensation polymer;
- (ii) 0 to 80% by weight of a compatible tackifier;
- (iii) 0 to 40% by weight of a compatible plasticizer;
- 25 (iv) 5 to 40% by weight of a compatible wax diluent with a molecular weight below 500 g/mole containing at least one polar functional group, said group being present at a concentration greater than  $3 \times 10^{-3}$  equivalents per gram;
- (v) 0 to 60% by weight of a compatible crystalline thermoplastic polymer; and
- 30 (vi) 0 to 3% by weight of a stabilizer.

What is needed is a composition (e.g. adhesive composition or coating) containing either dissolved or finely dispersed susceptor constituents that are preferably colorless or of low color. Further, the composition should be transparent or translucent throughout an adhesive matrix or plastic layer. This type of RF susceptor will result in more direct and uniform heating throughout an adhesive matrix or plastic layer. Further, it is desirable that such a composition will allow bonding with no physical distortion or discoloration in the bonded region of thin films. Still another desirable feature is activation of the RF susceptors at frequencies, e.g. of about 15 MHz or below, most preferably about 13.5 MHz, which are more economical to generate than higher frequencies and do not substantially heat dielectric substrates. A further desirable feature is that the composition can be activated or melted in less than one second and that it exhibit acceptable shear strength. It is also desirable to have a formulation which may be optimized for a particular application, such as cutting, coating, or bonding substrates.

### ***Summary of the Invention***

The present invention generally relates to the creation and use of a composition (also referred to as a "susceptor composition") that can bond two or more layers or substrates to one another and that can be used to coat or cut a substrate. The susceptor composition is activated in the presence of radio frequency (RF) energy.

In one embodiment, the susceptor composition of the present invention comprises a susceptor and a carrier. The carrier and susceptor are blended with one another and form a mixture, preferably a substantially uniform mixture. The susceptor is present in an amount effective to allow the susceptor composition to be heated by RF energy. In a preferred embodiment, the susceptor also functions as an adhesive or coating.

In another embodiment of the present invention, the susceptor composition further comprises an adhesive compound. The adhesive compound, susceptor,

and carrier are blended with one another to form a mixture that is activated in the presence of RF energy. Preferably, the mixture is substantially uniform.

In another embodiment of the present invention, the susceptor composition further comprises at least one of a thermoplastic polymer, thermoset resin, elastomer, plasticizer, filler or other material. The additive, susceptor, and carrier are blended with one another to form a mixture that is activated in the presence of RF energy.

In yet another embodiment of the present invention, the composition can further comprise a second carrier that is an insoluble porous carrier that is saturated with the composition.

The susceptor is an ionic or polar compound and acts as either a charge-carrying or an oscillating/vibrating component of the susceptor composition. The susceptor generates thermal energy in the presence of an RF electromagnetic or electrical field (hereafter RF field). According to the present invention, the susceptor can be an inorganic salt (or its respective hydrate), such as stannous chloride ( $\text{SnCl}_2$ ), zinc chloride ( $\text{ZnCl}_2$ ) or other zinc salt, or lithium perchlorate ( $\text{LiClO}_4$ ), or an organic salt, such as lithium acetate ( $\text{LiC}_2\text{H}_3\text{O}_2$ ). The susceptor can be a non-ferromagnetic ionic salt. The susceptor can also be a polymeric ionic compound ("ionomer") which preferably also functions as an adhesive or coating. Under RF power levels of about .05 kilowatt (kW) to 1 kW, and frequencies of about 1 to 100 MHz, the susceptor composition of the present invention facilitates (a) the bonding of single layers of polymeric materials such as polyolefins, non-polyolefins, and non-polymeric materials, as well as multilayer stacks of these materials, and (b) coating on a substrate such as a printed pattern on plastic films, metallic foils, etc.

Surprisingly, it has been discovered that when an ionomer is combined with a polar carrier, much more heating occurs when exposed to RF energy than when the ionomer or carrier is exposed separately to RF energy. Also surprisingly, it has been discovered that when the polar carrier is present at about 13-30% weight percent, more preferably, about 15-25 weight percent, most

preferably, about 20-23 weight percent, very short heating times are possible while retaining acceptable shear strength of the bond.

According to another embodiment of the present invention, a method of bonding a first material or substrate to a second material or substrate comprises interposing a composition according to the invention between the first and second materials and applying RF energy to the composition to heat the composition, thereby causing the first and second materials to become bonded. In one embodiment, the composition comprises a susceptor and a carrier that are distributed in one another to form a mixture, preferably, a substantially uniform mixture. Optionally, the composition may further comprise other compounds and additives as described herein. The susceptor is present in the composition in an amount effective to allow the composition to be heated by RF energy.

According to another embodiment of the present invention, a method of bonding or adhering a first substrate to a second substrate includes: applying a first composition onto the first substrate; applying a second composition onto the second substrate; contacting the first composition with the second composition; applying RF energy to the first and second compositions to heat the compositions, thereby causing the first and second substrates to become adhered or bonded; wherein one of the compositions comprises a susceptor and the other of the susceptors is a polar carrier, and the susceptor and/or the carrier are present in amounts effective to allow the first and second compositions to be heated by RF energy.

According to yet another embodiment of the present invention, a method of bonding or adhering a first substrate to a second substrate includes: applying a first composition onto the first substrate; applying a second composition onto the first composition; contacting the second substrate with the second composition; and applying RF energy to the first and second compositions to heat the compositions, thereby causing the first and second substrates to become adhered or bonded, wherein one of the compositions comprises a susceptor and the other of the compositions is a polar carrier, and the susceptor and/or the

carrier are present in amounts effective to allow the first and second compositions to be heated by RF energy.

According to another embodiment of the present invention, a method of making a susceptor composition comprises admixing a susceptor and a carrier, wherein, preferably, the carrier and susceptor are substantially uniformly dispersed in one another and form a uniform mixture. The susceptor and/or carrier are present in the composition in an amount effective to allow the susceptor composition to be heated by RF energy.

According to a further embodiment of the present invention, an adhered or a bonded composition can be obtained according to the disclosed methods.

According to a further embodiment of the present invention, a kit for bonding a first material to a second material comprises one or more containers, wherein a first container contains a composition comprising a susceptor and a carrier that are dispersed in one another and form a mixture. The kit may also contain an adhesive or elastomeric compound or other additives as disclosed herein. The susceptor and/or carrier are present in an amount effective to allow the composition to be heated by radio frequency energy.

According to a further embodiment of the present invention, a kit for adhering or bonding a first substrate to a second substrate, comprises at least two containers, wherein one of the containers comprises a susceptor and another of the containers comprises a polar carrier, wherein when the susceptor and the carrier are applied to substrates and the susceptor and carrier are interfaced, a composition is formed that is heatable by RF energy.

The invention also relates to a composition comprising an ionomeric polymer and a polar carrier.

The invention also relates to a method of curing a thermoset resin, comprising combining the thermoset resin with a polar carrier to give a mixture and exposing the mixture to RF energy.

The invention relates to an apparatus, having: a first portion having a first mating surface; a second portion, having a second mating surface; a composition disposed between the first mating surface and the second mating surface, wherein



the composition comprises a susceptor and a polar carrier wherein the susceptor and/or the polar carrier are present in amounts effective to allow the composition to be heated by RF energy, and wherein the composition adheres the first mating surface to the second mating surface such that application of a force to separate the first mating surface and the second mating surface results in breakage of the apparatus unless the composition is in a melted state.

The invention also relates to a method of applying a protective film or printed image/ink on a substrate.

The invention also relates to a method for dynamically bonding a first adherand to a second adherand. The method includes: (1) creating an article of manufacture comprising the first adherand, the second adherand, and a composition, the composition being between the first adherand and the second adherand, wherein the composition can be activated in the presence of an RF field; (2) moving the article of manufacture along a predetermined path; (3) generating along a portion of the predetermined path an RF field having sufficient energy to activate the composition, wherein the composition is activated by its less than one second exposure to the RF field.

The invention also relates to a method for applying a susceptor composition to a substrate. In one embodiment, the method includes: (1) formulating the susceptor composition as a liquid dispersion; (2) applying the liquid dispersion of the susceptor composition to the substrate; (3) drying the susceptor composition, wherein the drying step includes the step of applying RF energy across the composition, thereby generating heat within the liquid dispersion. In a preferred embodiment, one may roll up the substrate after the susceptor composition has dried.

The invention also relates to a method for cutting a substrate. The method includes: (1) applying a composition to a portion of the substrate, wherein the composition comprises a susceptor and polar carrier wherein the susceptor and/or said polar carrier are present in amounts effective to allow the composition to be heated by RF energy, and wherein the portion of the substrate defines a first section of the substrate and a second section of the substrate; (2) melting the

portion of the substrate by heating the composition via RF energy; and (3) after the portion of the substrate has begun to melt, applying a force to the substrate to separate the first section from the second section.

5 The method also relates to a method of dynamically bonding a first substrate to a second substrate. The method including: applying a composition onto the first substrate; after applying the composition onto the first substrate, forming a roll of the first substrate; storing the roll; unrolling the roll; and while unrolling the roll: joining an unrolled portion of the first substrate with a portion of the second substrate such that the portion of the second substrate is in contact with a portion of the composition applied onto the first substrate; and applying RF energy to the portion of the composition, wherein the portion of the composition heats and melts as a result of the RF energy being applied thereto.

10 Further features and advantages of the present invention, as well as the structure and operation of various embodiments of the present invention, are described in detail below with reference to the accompanying drawings.

### ***Brief Description of the Drawings***

The present invention is described with reference to the accompanying drawings. In the drawings, like reference numbers indicate identical or functionally similar elements. Additionally, the left-most digit(s) of a reference number identifies the drawing in which the reference number first appears.

20 FIGS. 1A and 1B illustrate conventional schemes for inductively heating adhesives.

FIG. 2 shows an RF active composition according to the present invention.

FIG. 3 shows a susceptor composition placed between two polyolefin layers to be attached according to the present invention.

25 FIG. 4 illustrates a block diagram of an RF heating system according to a first embodiment.

FIG. 5 illustrates a block diagram of a heating system according to a second embodiment.

FIG. 6 illustrates a two probe heating system.

FIGS. 7A and 7B further illustrate the two probe heating system.

FIG. 7C illustrates a probe having a curled end to reduce corona effects.

FIG. 8 illustrates one embodiment of an alternating voltage supply.

5        FIG. 9 is a flow chart illustrating a process for heating a composition according to the present invention.

FIG. 10A further illustrates one embodiment of an impedance matching circuit.

10        FIG. 10B further illustrates another embodiment of an impedance matching circuit.

FIG. 11 shows a method of bonding adherents using a composition that is activated in the presence of RF energy.

FIGS. 12 to 17 illustrate additional embodiments of probes 602 and 604.

15        FIG. 18 illustrates one embodiment of an application system for applying a composition according to the present invention to a substrate.

FIG. 19 illustrates one embodiment of a system for bonding or adhering various adherents.

FIGS. 20A and 20B illustrates a static bonding system for bonding adherents.

20        FIG. 20C illustrates an electrically insulating block for housing probes.

FIG. 21 illustrates an in-line bonding system.

FIG. 22 further illustrates one embodiment of the in-line bonding system illustrated in FIG. 21.

25        FIGS. 23-27 illustrate alternative designs of the in-line bonding system illustrated in FIG. 21.

FIGS. 28A and 28B illustrate one embodiment of a system for the manufacture of flexible packaging material.

FIG. 29 further illustrates film 2815.

FIG. 30 illustrates one embodiment of film 2870.

30        FIG. 31 illustrates an alternative system for manufacturing an RF activated adhesive film for use in the flexible packaging industry.

FIG. 32 illustrates a conventional aseptic package material construction.

FIG. 33 illustrates an aseptic package material according to one embodiment that does not include metallic foil.

FIG. 34 illustrates another embodiment of an aseptic packaging material construction that does not use metallic foils.

FIG. 35 illustrates a conventional cap sealing construction.

FIG. 36 illustrates a seal, according to one embodiment, for sealing a bottle.

FIG. 37 illustrates a design for adhering a flexible bag to an outer box.

FIG. 38 illustrates a step and repeat manufacturing system.

FIG. 39 illustrates an index table bonding system.

FIG. 40 shows an example experimental set-up utilized to test compositions according to the present invention.

FIG. 41 illustrates another experimental set-up for testing compositions according to the present invention.

FIG. 42 illustrates test probes.

FIG. 43 illustrates a process for assembling a book, magazine, or periodical, or the like.

FIG. 44 illustrates a paper substrate coated with a susceptor composition.

FIG. 45 illustrates a stack of coated paper substrates.

FIGS. 46 and 47 illustrates one embodiment of an envelope or mailer according to the present invention.

FIG. 48 illustrates a cross-section of a container sealed with a susceptor composition of the present invention.

FIG. 49 illustrates another example of a device sealed or otherwise joined together with a composition of the present invention.

FIG. 50 shows another example of a device sealed or otherwise joined together with a composition of the present invention.

FIG. 51 illustrates still another example of a cross-section of a container 5100 that has been sealed with the adhesive of the present invention.

FIG. 52 illustrates a system for bonding two substrates.

FIG. 53 illustrates another embodiment of a system for bonding two substrates.

Fig. 54 depicts a graph showing RF activation time vs. % Glycerin for a composition comprising AQ55S.

5 Fig. 55 depicts a graph showing shear holding time vs. % glycerin for a composition comprising AQ55S.

Fig. 56 depicts a graph showing RF activation time vs. % glycerin for a composition comprising AQ35S.

10 Fig. 57 depicts a graph showing shear holding time vs. % glycerin for a composition comprising AQ35S.

Fig. 58 depicts a family of curves showing RF activation time vs. % various polar carriers.

Fig. 59 depicts a graph showing RF activation time vs. % PARICIN 220 in a composition comprising 80% AQ55S/20% glycerin.

15 Fig. 60 depicts a graph showing brookfield viscosity vs. % PARICIN 220 in a composition comprising 80% AQ55S/20% glycerin.

FIG. 61 depicts a graph showing RF activation time vs. % glycerin in a composition comprising the sodium salt of an ethylene acrylic acid copolymer (MICHEM Prime 48525P).

20 FIG. 62 illustrates a seam sealing system according to one embodiment.

### ***Detailed Description of the Preferred Embodiments***

#### ***I. Overview and Discussion of the Invention***

#### ***II. Terminology***

##### ***A. Sulfonated Polymers***

##### ***B. Acrylic Acid and Maleic Anhydride Polymers and Copolymers***

##### ***C. Starch/Polysaccharide Derivatives***

##### ***D. Proteins***

##### ***E. Others***

#### ***III. The Polar Carrier***

**IV. Further Additives to the Susceptor Compositions**

- A. Adhesive/Thermoplastic Additives**
- B. Adhesive/Coating Thermoset Additives**
- C. Surfactant Additives**
- D. Plasticizer Additives**
- E. Tackifiers**
- F. Fillers**
- G. Stabilizers and Antioxidants**
- H. Other Additives**

**V. Applying the Susceptor Compositions to Substrates**

**VI. Apparatus For Activating the Various Compositions of the Present Invention**

**VII. Method of Bonding Substrates**

**VIII. Additional Probe Embodiments**

**IX. Applicator System for Applying a Composition of the Present Invention to a Substrate/Adherand**

**X. Systems for Adhering or Bonding two Adherands.**

**XI. Exemplary Specific Applications of the Present Invention**

- A. Manufacture of Flexible Packaging**
- B. Food Packaging and Cap Sealing**
- C. Printing Applications**
- D. Bookbinding and Mailers**
- E. Security Devices**
- F. Thermal Destruction**
- G. Seam Sealing**

**XII. Kits**

**XIII. Experimental Set-up**

**XIV. Examples**

## ***I. Overview and Discussion of the Invention***

The present invention is directed towards an RF susceptor composition and methods and systems of bonding, cutting, and/or coating substrates and surfaces using the susceptor composition. The susceptor composition is a mixture  
5 of RF susceptors and/or adhesive/coating compounds and/or other additives dissolved or finely dispersed in a matrix. Preferably, the RF susceptors and/or adhesive compounds and/or other additives are uniformly dissolved or finely dispersed in the matrix. The susceptor composition is capable of coupling efficiently in an RF field having a frequency of about 15MHz or below. In order  
10 to be useful in industry and commercial products, a susceptor composition preferably has the following characteristics: (1) an activation time in the presence of a low power RF field on the order of 1 second or less, (2) adequate bond or adhesive strength for the intended use, (3) transparency or translucency and only slight coloration (if any), (4) minimal distortion of the substrates being attached,  
15 and (5) on demand bonding of preapplied adhesive. Further, it is desirable that the susceptor composition have coupling ability in the absence of volatile solvents, although the presence of nonvolatile liquids (such as plasticizers) may be desirable. These characteristics are important in providing sufficient heat transfer to the substrates or layers to be bonded to one another, or for adhesion to take place at  
20 the interface. Additionally, the susceptor composition should not interfere with the thermal bonding or inherent adhesive properties of the substrates or layers to be bonded or adhered to one another.

According to the present invention, a susceptor composition used to bond or adhere substrates or layers can be directly heated by exposure to an RF field  
25 having frequencies ranging from 1 - 100 MHz. The susceptor composition comprises a susceptor, and a carrier blended with one another to form a mixture. In addition, the susceptor composition can further comprise one or more adhesive compounds blended with the susceptor and carrier to form the mixture.

Susceptors are either ionic or polar compounds introduced as a component  
30 of a composition, such that RF heating of the resulting susceptor composition

occurs. An ionic susceptor is an ionic compound introduced as a sufficiently charge-carrying or oscillating component of the composition. A polar susceptor is a polar compound which has sufficiently high dipole moment that molecular oscillations or vibrations of the compound occur when exposed to an RF field. As shown in FIG. 2, a susceptor composition 202 comprises a continuous mixture of susceptors such as microscopic, ionic salts or polymeric ionic compounds or dipoles 204, which generate thermal energy in the presence of the RF field. It has been discovered that acceptable bonding results occur with inorganic salts such as stannous chloride ( $\text{SnCl}_2$ ); zinc salts such as chloride ( $\text{ZnCl}_2$ ), bromide ( $\text{ZnBr}_2$ ) and the like; and lithium perchlorate ( $\text{LiClO}_4$ ), and organic salts such as lithium acetate ( $\text{LiC}_2\text{H}_3\text{O}_2$ ). These salts or combination of salts, when distributed in the mixture, create an ionic and/or polar medium capable of being heated by RF energy.

## *II. Terminology*

"RF Energy" means an alternating electromagnetic field having a frequency within the radio frequency spectrum.

A "susceptor composition" comprises a susceptor and a carrier interfaced with one another and/or mixed or blended together. Preferably, the susceptor and carrier are mixed together. More preferably, the susceptor and carrier are substantially uniformly mixed together. In another embodiment, the susceptor and carrier are interfaced together by disposing a layer of the susceptor onto a layer of the carrier or visa versa. In this embodiment, the susceptor may be coated onto a first substrate and the carrier, with or without added ingredients such as a wax or other additives that prevent the carrier from evaporating substantially, may be coated onto a second substrate. The first and second substrates containing the susceptor and carrier layers, respectively, may then be brought into contact or interfaced and activated then or at a later time.

The susceptor compositions of the invention may further comprise one or more adhesive compounds or other additives mixed, preferably substantially



uniformly mixed, together with the susceptor and the carrier. The susceptor composition is activated in the presence of radio frequency (RF) energy. The susceptor composition can be used to bond two or more layers or substrates to one another, can be used as a coating, and can be used to thermally cut substrates.

5 A "carrier" provides the mobile medium in which the susceptors are dissolved, distributed, or dispersed. Preferably, the carrier is a polar carrier as defined below which enhances the activation of the compositions. Carriers (also referred to as mobile media) can be liquids, such as solvents and plasticizers, or polymers that are utilized for their polar functionality and for their ability to be heated by RF energy.

10 An "adhesive compound" refers to polymers, copolymers and/or ionomers as described herein that are blended into the susceptor composition to enhance its adhesive properties.

15 "Bonding" is defined as the joining of one substrate to another substrate to cause a physical joining process to occur.

"Adhesion" is an interaction between two adherands at their interface such that they become attached or joined.

A "substantially transparent" mixture refers to a mixture that transmits greater than about 50% of incident visible light.

20 "Thermal bonding" or "welding" is defined as the reflowing of one substrate into another substrate to cause a physical joining process to occur.

"Mechanical bonding" occurs between adherands when a susceptor composition holds the adherands together by a mechanical interlocking action.

25 An RF "susceptor" converts coupled RF energy into heat energy in the susceptor composition. According to the present invention, the susceptor, as described above, is either the charge carrying or oscillating ionic compound or the oscillating polar compound having a sufficiently high dipole moment comprising a composition to generate thermal energy in the presence of an RF field. Generally, the susceptor can be a salt. For example, the susceptor can be an inorganic salt or its respective hydrate(s), such as stannous chloride ( $\text{SnCl}_2$ ),  
30 stannous chloride dihydrate ( $\text{SnCl}_2 \times 2\text{H}_2\text{O}$ ), lithium perchlorate ( $\text{LiClO}_4$ ), lithium

perchlorate trihydrate ( $\text{LiClO}_4 \times 3\text{H}_2\text{O}$ ) or an organic salt, such as an alkali metal salt of a  $\text{C}_{1-4}$  alkanolic acid such as lithium acetate ( $\text{LiC}_2\text{H}_3\text{O}_2$ ), lithium acetate dihydrate ( $\text{LiC}_2\text{H}_3\text{O}_2 \times 2\text{H}_2\text{O}$ ), or sodium acetate and the like; alkali metal salts of arylcarboxylic acids such as lithium benzoate, sodium benzoate, and the like; alkali metal salts of alkyl and aryl sulfonates such as sodium methylsulfonate and sodium *p*-toluenesulfonate and the like. Other types of salts and their respective hydrates include, but are not limited to, magnesium acetate, magnesium nitrate, sodium-based salts (such as sodium chloride, sodium bromide and the like), lithium-based salts (such as lithium bromide, lithium carbonate, lithium chloride, etc.) and potassium-based salts. Many of these salts are commercially available from Aldrich Chemical Company, Milwaukee, WI. See the Aldrich Catalog Handbook of Fine Chemicals 1996-1997. It is not intended that this list of salts is an exclusive or comprehensive list. These salts are disclosed as typical examples. The present invention is not restricted to the listed salts, as would be apparent to those of skill in the art.

The susceptor can also be an ionomer. Preferably, the ionomer also functions as an adhesive and/or coating. Examples of such ionomers include without limitation styrenated ethylene-acrylic acid copolymer or its salts, sulfonated polyesters and their salts, sulfonated polystyrene and its salts and copolymers, polyacrylic acid and its salts and copolymers, hydroxy/carboxylated vinylacetate-ethylene terpolymers, functionalized acrylics, polyesters, urethanes, epoxies, alkyds, latex, gelatin, soy protein, casein and other proteins, alginate, carrageenan, starch derivatives, ionic polysaccharides, and the like. An example of an ionomer that does not function as an adhesive is sodium polystyrenesulfonate.

Examples of ionomer adhesives are described in more detail below.

#### **A. Sulfonated Polymers**

Sulfonated polyesters and copolymers thereof are described in U.S. Patent Nos. 5,750,605, 5,552,495, 5,543,488, 5,527,655, 5,523,344, 5,281,630,

4,598,142, 4,037,777, 3,033,827, 3,033,826, 3,033,822, 3,075,952, 2,901,466,  
2,465,319, 5,098,962, 4,990,593, 4,973,656, 4,910,292, 4,525,524, 4,408,532,  
4,304,901, 4,257,928, 4,233,196, 4,110,284, 4,052,368, 3,879,450, and  
3,018,272. The invention relates to compositions comprising sulfonated  
polyesters and copolymers thereof, e.g. as described in these patents, together  
with a polar carrier as described herein as well as the adhesive compositions  
described in these patents (comprising the sulfonated polyesters and copolymers  
thereof) together with the polar carrier. Such sulfonated polyesters and  
copolymers thereof are one preferred embodiment of the present invention, as  
such materials function both as an ionomeric susceptor and as an adhesive.

In a preferred embodiment, the sulfonated polyester is a higher Tg (about  
48 °C to about 55 °C or higher) linear polyester which shows improved heat  
resistance compared to lower Tg (about 35 °C) linear or branched sulfonated  
polyesters. Once blended with the polar carrier, the Tg of the resulting  
composition should be higher than the temperature at the intended use, e.g. higher  
than body temperature for diaper adhesives. For example, a linear sulfonated  
polyester with a Tg of 55 °C (e.g. AQ55S) blended with a sufficient amount  
(greater than 10%) of polar carrier (e.g. glycerin) to achieve RF activity will result  
in a Tg higher than body temperature if the polar carrier is no more than about  
35% of the composition.

In another embodiment, a salt comprising a sulfonated polyester and a  
cationic dye as described in U.S. Patent No. 5,240,780, are employed. Such salts  
provide a colored susceptor composition that may be used, e.g. in printing.

Sulfonated polyesters may be prepared by the polycondensation of the  
following reactants:

- (a) at least one dicarboxylic acid;
- (b) at least one glycol;
- (c) at least one difunctional sulfomonomer containing at least one  
metal sulfonate group attached to an aromatic nucleus wherein the  
functional groups may be hydroxy, carboxyl, or amino groups.

The dicarboxylic acid component of the sulfonated polyesters comprises aliphatic dicarboxylic acids, alicyclic dicarboxylic acids, aromatic dicarboxylic acids, or mixtures of two or more of these acids. Examples of such dicarboxylic acids include oxalic; malonic; dimethylmalonic; succinic; glutaric; adipic; trimethyladipic; pimelic; 2,2-dimethylglutaric; azelaic; sebacic; fumaric; maleic; itaconic; 1,3-cyclopentanedicarboxylic; 1,2-cyclohexanedicarboxylic; 1,3-cyclohexanedicarboxylic; 1,4-cyclohexanedicarboxylic; phthalic; terephthalic; isophthalic; 2,5-norbornanedicarboxylic; 1,4-naphthalic; diphenic; 4,4'-oxydibenzoic; diglycolic; thiodpropionic; 4,4'-sulfonyldibenzoic; and 2,5-naphthalenedicarboxylic acids. If terephthalic acid is used as the dicarboxylic acid component of the polyester, at least 5 mole percent of one of the other acids listed above may also be used.

It should be understood that use of the corresponding acid anhydrides, esters, and acid chlorides of these acids is included in the term "dicarboxylic acid." Examples of these esters include dimethyl 1,4-cyclohexanedicarboxylate; dimethyl 2,5-naphthalenedicarboxylate; dibutyl, 4,4'-sulfonyldibenzoate; dimethyl isophthalate; dimethyl terephthalate; and diphenyl terephthalate. Copolyesters may be prepared from two or more of the above dicarboxylic acids or derivatives thereof.

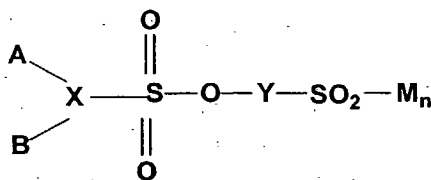
Examples of suitable glycols include poly(ethylene glycols) such as diethylene glycol, triethylene glycol, tetraethylene glycol, and pentaethylene, hexaethylene, heptaethylene, octaethylene, nonaethylene, and decaethylene glycols, and mixtures thereof. Preferably the poly(ethylene glycol) employed in the present invention is diethylene glycol or triethylene glycol or mixtures thereof.

The remaining portion of the glycol component may consist of aliphatic, alicyclic, and aralkyl glycols. Examples of these glycols include ethylene glycol; propylene glycol; 1,3-propanediol; 2,4-dimethyl-2-ethylhexane-1,3,diol; 2,2-dimethyl-1,3-propanediol; 2-ethyl-2-butyl-1,3-propanediol; 2-ethyl-2-isobutyl-1,3-propanediol; 1,3-butanediol; 1,4-butanediol; 1,5-pentanediol; 1,6-hexanediol; 2,2,4-trimethyl-1,6-hexanediol; thiodiethanol; 1,2-cyclohexanedimethanol; 1,3-cyclohexanedimethanol; 1,4-cyclohexanedimethanol; 2,2,4,4-tetramethyl-1,3-

cyclobutanediol; p-xylylenediol. Copolymers may be prepared from two or more of the above glycols.

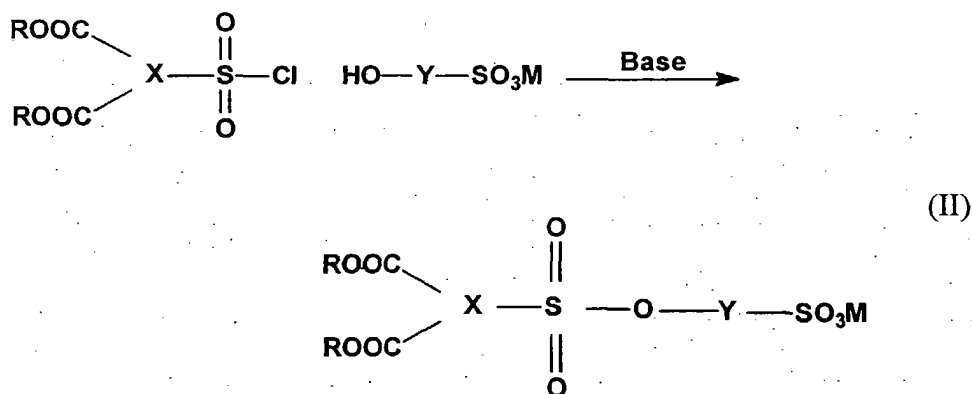
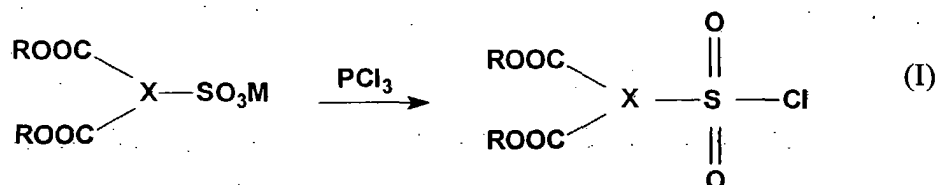
The difunctional sulfo-monomer component of the sulfonated polyester may advantageously be a dicarboxylic acid or an ester thereof containing a metal sulfonate group or a glycol containing a metal sulfonate group or a hydroxy acid containing metal sulfonate group.

Advantageous difunctional sulfo-monomer components are those wherein the sulfonate salt group is attached to an aromatic acid nucleus such as benzene, naphthalene, diphenyl, oxydiphenyl, sulfonyldiphenyl, or methylenediphenyl nucleus. Particular examples include sulfophthalic acid, sulfoterephthalic acid, sulfoisophthalic acid, 4-sulfonaphthalene-2,7-dicarboxylic acid, and their esters; metalosulfoaryl sulfonate having the general formula.



wherein X is a trivalent aromatic radical derived from a substituted or unsubstituted aromatic hydrocarbon, Y is a divalent aromatic radical derived from a substituted or unsubstituted aromatic hydrocarbon, A and B are carboalkoxy groups containing 1 to 4 carbon atoms in the alkyl portion or a carboxy group, the metal ion M is  $\text{Li}^+$ ,  $\text{Na}^+$ ,  $\text{K}^+$ ,  $\text{Mg}^{++}$ ,  $\text{Ca}^{++}$ ,  $\text{Ba}^{++}$ ,  $\text{Cu}^{++}$ ,  $\text{Fe}^{++}$ ,  $\text{Fe}^{+++}$ , and n is 1 for monovalent M or 2 for divalent M or 3 for trivalent M. When a monovalent alkali metal ion is used, the resulting sulfonated polyesters are less readily dissipated by cold water and more rapidly dissipated by hot water. When a divalent or a trivalent metal ion is used, the resulting sulfonated polyesters are not ordinarily easily dissipated by cold water, but are more readily dissipated in hot water. Depending on the end use of the polymer, either of the different sets of properties may be desirable. It is possible to prepare the sulfonated polyester using, for example, a sodium sulfonate salt and later by ion-exchange replace this ion with a different ion, for example, calcium, and thus alter the characteristics of the

polymer. In general, this procedure is superior to preparing the polymer with divalent metal salt inasmuch as the sodium salts may be more soluble in the polymer manufacturing components than are the divalent metal salts. Polymers containing divalent or trivalent metal ions are less elastic and rubber-like than polymers containing monovalent ions. One such metallosulfoaryl sulfonate component may be prepared as shown by the following general reactions:



and other chlorinating agents (e.g., thionyl chloride, phosphorus trichloride, phosphorous oxychloride) may be used. In addition, the reaction between the sulfonyl chloride and the sulfophenol may be carried out in water or an inert organic solvent, and the base used may be an alkali metal hydroxide or a tertiary amine. Such suitable compounds are disclosed in U.S. Patent No. 3,734,874.

Optionally, the polycondensation reaction may be carried out in the presence of one or more of the following:

- (d) an unsaturated mono- or dicarboxylic acid; and,
- (e) a difunctional hydroxycarboxylic acid having one  $-\text{CH}_2-\text{OH}$  group, an aminocarboxylic acid having one  $-\text{NRH}$  group, an amino alcohol having one  $-\text{CR}_2-\text{CH}$  and one  $-\text{NRH}$  group, a diamine having two  $-\text{NRH}$  groups, or a mixture thereof, wherein each R is hydrogen or a  $\text{C}_{1-4}$  alkyl group.

The  $\alpha,\beta$ -unsaturated acids (d) are described by the following structure:



wherein R is H, alkylcarboxy, or arylcarboxy and R<sup>1</sup> is carboxy or arylcarboxy. Polymers derived from the above components can be used in combination with polymers derived from other components and/or in combination with other ethylenically unsaturated comonomers (e.g., acrylic acid, acrylamide, butyl acrylate, diacetone acrylamide). The comonomers can be from 1-75 parts by weight, preferably 5-25 parts by weight  $\alpha,\beta$ -unsaturated acids.

Advantageous difunctional components which are aminoalcohols include aromatic, aliphatic, heterocyclic and other types as in regard to component (e). Specific examples include 5-aminopentanol-1, 4-aminomethylcyclohexanemethanol, 5-amino-2-ethyl-pentanol-1, 2-(4- $\beta$ -hydroxyethoxyphenyl)-1-aminoethane, 3-amino-2,2-dimethylpropanol, hydroxyethylamine, etc. Generally these aminoalcohols contain from 2 to 20 carbon atoms, one-NRH group and one -CR<sub>2</sub>-OH group.

Such difunctional monomer components which are aminocarboxylic acids include aromatic, aliphatic, heterocyclic, and other types as in regard to component (c) and include lactams. Specific examples include 6-aminocaproic acid, its lactam known as caprolactam, omegaaminoundecanoic acid, 3-amino-2-dimethylpropionic acid, 4-( $\beta$ -aminoethyl)benzoic acid, 2-( $\beta$ -aminopropoxy)benzoic acid, 4-aminomethylcyclohexanecarboxylic acid, 2-( $\beta$ -aminopropoxy)cyclohexanecarboxylic acid, etc. Generally these compounds contain from 2 to 20 carbon atoms.

Examples of such difunctional monomer component (e) which are diamines include ethylenediamine; hexamethylenediamine; 2,2,4-trimethylhexamethylenediamine; 4-oxaheptane-1,7-diamine; 4,7-dioxadecane-1,10-diamine; 1,4-cyclohexanebismethylamine; 1,3-cycloheptamethylene-diamine; dodecamethylenediamine, etc.

Greater dissipatability is achieved when the difunctional sulfo-monomer constitutes from about 6 mole percent to about 25 mole percent out of a total of 200 mole percent of (a), (b), (c), (d), and any (e) components of the polyester or

polyesteramide. The total of 200 mole percent can also be referred to as 200 mole parts.

Any of the above-identified difunctional monomers generally contain hydrocarbon moieties having from 1 to about 40 carbon atoms in addition to their two functional groups, but they may in general also contain up to six non-functional groups such as -O-, -S-, -SO<sub>2</sub>-, -SO<sub>2</sub>-O-, etc. For example, the poly(ethylene glycol) monomer used may contain from 1 to about 19 oxy groups, such as -O- groups.

In a preferred embodiment, the ionomer is one of the sulfonated polyesters sold by Eastman Chemical Company, Kingsport, TN (hereafter "Eastman"), which are water dispersible, linear or branched polyesters formed by the polycondensation of glycols with dicarboxylic acids, some of which contain sodiosulfo groups. Sulfopolyester hybrids may also be employed which are formed by the in situ polymerization of vinyl and/or acrylic monomers in water dispersions of SULFOPOLYESTER. Such Eastman sulfonated polyesters may be purchased from Eastman under nos. AQ1045, AQ1350, AQ1950, AQ14000, AQ35S, AQ38S, AQ55S and EASTEK 1300.

The sulfonated polyesters and copolymers thereof may range from about 10 to about 90 weight percent, more preferably, about 60 to 80 weight percent, most preferably about 70 weight percent of the total composition. The polar carrier may range from about 10 to about 90 weight percent, more preferably, about 20 to about 40 weight percent, most preferably, about 30 weight percent of the total composition. The remainder of the composition may comprise one or more of the other additives described herein.

Compositions comprising branched sulfonated polyesters tend to give clear, tacky and flexible films. Compositions comprising linear polyesters tend to give clear or white, tack-free, flexible films.

Other sulfonated polymers that can be used in the practice of the invention include polystyrene sulfonate, acrylamino propane sulfonate (AMPS) based polymers (e.g. 2-acrylamido-2-methylpropanesulfonic acid and its sodium salt available from Lubrizol Process Chemicals). In addition, urethane ionomers can



be prepared by reacting a diisocyanate with a diol that has sulfonate functionality (e.g. butane diol sulfonate).

***B. Acrylic Acid and Maleic Anhydride Polymers and Copolymers***

Other ionomers include acrylic acid polymers and copolymers and salts thereof. Such polymers and copolymers are described in U.S. Patent Nos. 5,821,294, 5,717,015, 5,719,244, 5,670,566, 5,618,876, 5,532,300, 5,530,056, 5,519,072, 5,371,133, 5,319,020, 5,037,700, 4,713,263, 4,696,951, 4,692,366, 4,617,343, 4,948,822, and 4,278,578.

The invention relates to compositions comprising the acrylic acid polymers and copolymers thereof described in these patents together with a polar carrier as described herein as well as the adhesive compositions described in these patents (comprising the acrylic acid polymers and copolymers thereof) together with the polar carrier.

Specific examples of such acrylic acid copolymers include ethylene acrylic acid copolymer and the ammonium (MICHEM 4983P) and sodium (MICHEM 48525P) salts thereof available from Michelman Incorporated, Cincinnati, OH. A further example is vinyl acetate acrylic copolymers (e.g. ROVACE HP3442) available from Rohm and Hass, Philadelphia, PA.

The acrylic acid polymers and copolymers may range from about 10 to about 90 weight percent, more preferably, about 40 to 80 weight percent, most preferably about 50-70 weight percent of the total composition. The polar carrier may range from about 10 to about 90 weight percent, more preferably, about 10 to about 40 weight percent, most preferably, about 30 weight percent of the total composition. The remainder of the composition may comprise one or more of the other additives described herein.

Compositions comprising ethylene acrylic acid copolymers and a polar carrier tend to give clear, colorless, tack-free films with very good adhesion that heat in well under one second when exposed to RF. Vinyl acetate acrylic

copolymer compositions tend to give clear, colorless, flexible but very tacky films with very good adhesion that heat in well under one second when exposed to RF.

In a preferred embodiment, compositions comprising acrylic acid polymers or copolymers are applied as liquid dispersions and dried into an RF susceptible coating.

Alternatively, maleic anhydride based copolymers such styrene maleic anhydride, ethylene maleic anhydride, and polyethylene maleic anhydride (available from Eastman Chemicals) may be employed as an ionomer. Such compositions are preferably applied as an aqueous dispersion at room temperature and dried into an RF susceptible coating.

### *C. Starch/Polysaccharide Derivatives*

Other ionomers include starch and polysaccharide derivatives such as polysulfonated or polysulfated derivatives, including dextran sulfate, pentosan polysulfate, heparin, heparan sulfate, dermatan sulfate, chondroitin sulfate, a proteoglycan and the like. Dextran sulfate is available from Sigma Chemical Corporation, St. Louis, MO, with molecular weights of 10,000, 8,000 and 5,000. Examples of other ionic polysaccharides include carrageenan, chitosan, xanthan gum, etc.

Phosphorylated starch as disclosed in U.S. 5,329,004 may be employed as a susceptor.

The starch/polysaccharide derivatives may range from about 10 to about 90 weight percent, more preferably, about 60 to 80 weight percent, most preferably about 70 weight percent of the total composition. The polar carrier may range from about 10 to about 90 weight percent, more preferably, about 20 to about 40 weight percent, most preferably, about 30 weight percent of the total composition. The remainder of the composition may comprise one or more of the other additives described herein.

#### **D. Proteins**

Other ionomers include proteins such as gelatin, soy protein, casein, etc. Gelatin is the purified protein derived from the selective hydrolysis of collagen. Collagen is the principal organic component of the bones and skin of mammals. Common raw materials include bones, cattle hides and pigskins. Gelatins are classified as either acid type (A type) or limed (B type) according to the process by which they are made. Particular examples of gelatins include KNOX gelatin as well as types P, D, D-I, LB, LM and K, available from PB Gelatins. See also the gelatin described in U.S. Patent 5,877,287. In a preferred embodiment, the gelatin is 45Y56-853-3V0-6CS, available from Eastman Gelatin, Peabody, MA. Alternatively, a gelatin-modified polyurethane as disclosed in U.S. Patent 5,948,857 may be used.

In a preferred embodiment, the pH of the gelatin is raised or lowered in order to enhance the ionomeric character of the gelatin. The pH may be raised by the addition of aqueous base to an aqueous solution or suspension of the gelatin. Examples of suitable bases include alkali metal hydroxides, alkali metal carbonates and bicarbonates, alkali metal acetates, ammonia, amino compounds such as methylamine, dimethylamine, trimethylamine, triethylamine, and the like. Alternatively, a basic buffer solution may be added, e.g. a solution comprising 2-amino-2-methyl-1-propanol; or a glycine buffer at pH 9.4 and 10.4; each of which is available from Sigma Chemical Corporation, St. Louis, MO. Other buffers include 0.01 borax (pH 9.2), TRIS (pH 7-9.1 depending on concentration), 0.05 M carbonate (pH 9.93), and 0.05 M trisodium phosphate (pH 12). See "The Chemist's Companion," A.J. Gordon and R.A. Ford, John Wiley & Sons, New York, N.Y., 1972. The pH may be lowered by the addition of an acid such as HCl, HBr, H<sub>2</sub>SO<sub>4</sub>, H<sub>3</sub>PO<sub>4</sub>, or an organic acid such as C<sub>1-4</sub> alcanoic acid (e.g. acetic acid, propionic acid or butyric acid), an arylcarboxylic acid (e.g. benzoic acid), or arylsulfonic acid (e.g. *p*-toluenesulfonic acid). Alternatively, an acidic buffer may be added, e.g. acetate buffer at pH 4.5, 4.9 and 5.0; citrate buffer at pH 4.8; or a phosphate-citrate buffer at pH 5.0; each of which is available from Sigma

Chemical Corporation. Other buffers include 0.005 M potassium tetraoxalate (pH 1.7), saturated potassium tartrate (pH 3.6), 0.05 M potassium phthalate (pH 4.0), and 0.05 M sodium succinate (pH 5.3). See "The Chemist's Companion," A.J. Gordon and R.A. Ford, John Wiley & Sons, New York, NY, 1972. As discussed in the Examples, it has been discovered unexpectedly that when the pH of the gelatin composition is shifted into the acidic or basic range, the composition exhibits enhanced heating in an RF field compared to the untreated gelatin. The best heating occurs when the pH is low. Such gelatin compositions give flexible films that attach well to substrates and heat in under one second.

In a preferred embodiment, the pH of the gelatin may range from about 8 to about 12. In a most preferred embodiment, the pH of the gelatin is about 10. In another preferred embodiment, the pH of the gelatin may range from about 1 to about 6. In a most preferred embodiment, the pH of the gelatin is about 2.

The gelatin may range from about 10 to about 90 weight percent, more preferably, about 60 to 80 weight percent, most preferably about 70 weight percent of the total composition. The polar carrier may range from about 10 to about 90 weight percent, more preferably, about 20 to about 40 weight percent, most preferably, about 30 weight percent of the total composition. The remainder of the composition may comprise one or more of the other additives described herein.

#### *E. Others*

Other ionomers that may be used in the practice of the invention include sulfonated novolak resins obtained by a process comprising reacting an aromatic compound with a sulfonated agent to form a sulfonated aromatic compound, condensing the sulfonated aromatic compound with a non-sulfonated phenolic compound and an aldehyde or aldehyde precursor to form a sulfonated condensate, and reacting the condensate with a monovalent or divalent metal oxide, hydroxide, carbonic acid, boronic acid or carboxylic acid. See U.S. Patent No. 5,098,774. Other ionomers that can be used are lignosulfonates and their

sodium salts which are available with different molecular weights and levels of sulfonation from Westvaco, North Charleston, SC.

In addition, urethane ionomers can be prepared by reacting a diisocyanate with a diol that has carboxy functionality (e.g. dimethylol).

### 5      *III. The Polar Carrier*

10      In a preferred embodiment, the ionomer is combined with a carrier that is a flowable polar compound, such as a polar solvent, having a high dielectric constant, e.g.  $\epsilon$  (20°C)  $\geq$  about 10, more preferably,  $\geq$  about 20. A preferred dielectric constant range is about 13-63 (25°C), more preferably, about 17-43 (25°C). It has been unexpectedly discovered that compositions comprising an ionomer and such a carrier heat much more rapidly when exposed to RF energy, even at low levels, compared to when the ionomer or carrier are exposed separately. Without being bound by any particular theory, it is believed that upon exposure to RF energy, the polar carrier allows for the migration and/or vibration  
15      of protons or metal ions from the ionomer, resulting in the generation of heat.

20      Such polar carriers include, but are not limited to, water, dimethylformamide (DMF), dimethylacetamide (DMAC), dimethylsulfoxide (DMSO), tetrahydrofuran (THF), polypropylene carbonates, ketones (such as acetone, acetyl acetone, cyclohexanone, diacetone alcohol, and isophorone),  
25      alcohols (such as ethanol, propanol, 2-methyl-1-propanol, and the like) amino alcohols (such as ethanolamine), oxazolidines, polyols, organic acids (such as formic, acetic, propionic, butyric and dimethylol butyric acid and the like), anhydrides (such as acetic anhydride and maleic anhydride), amides (such as formamide, acetamide and propionamide), nitriles (such as acetonitrile and propionitrile), and nitro compounds (such as nitrobenzene, nitroaniline, nitrotoluene, nitroglycerine and any of the nitroparaffins). Any polar carrier that can weaken, to some degree, the ionic interaction between the anion and cation of the ionic susceptor, even if the susceptor component is a non-ionic compound, may be utilized in the present invention. Preferred polar carriers are humectants

(e.g., glycerin, 1,2-propanediol and polyethyleneglycol), i.e., they retain at least a low level of moisture after application. It is believed that the low level of residual moisture enhances the RF activation of the compositions.

5 Examples of polyols that may be used as a polar carrier include glycols such as diethylene glycol, triethylene glycol, tetraethylene glycol, dipropylene glycol, thioethylene glycol, and pentaethylene, hexaethylene, heptaethylene, octaethylene, nonaethylene, and decaethylene glycols, and mixtures thereof, as well as aliphatic, alicyclic, and aralkyl glycols. Particular examples of these glycols include ethylene glycol; 1,2-propylene glycol; 1,3-propanediol; 2,4-dimethyl-2-ethylhexane-1,3,diol; 2,2-dimethyl-1,3-propanediol; 2-ethyl-2-butyl-1,3-  
10 propanediol; 2-ethyl-2-isobutyl-1,3-propanediol; 1,3-butanediol; 1,4-butanediol; 1,5-pentanediol; 1,6-hexanediol; 2,2,4-trimethyl-1,6-hexanediol; thiodiethanol; 1,2-cyclohexanedimethanol; 1,3-cyclohexanedimethanol; 1,4-cyclohexanedimethanol; 2,2,4,4-tetramethyl-1,3-cyclobutanediol; p-xylylenediol.

15 Also included are polyethylene glycols, e.g. having weight average molecular weights ranging from about 400 to about 2,000; mixed poly(ethylene)-poly(propylene) glycols having weight average molecular weights ranging up to about 6,000 and containing from about 30 to about 90 weight percent ethylene oxide; the monomethyl, monoethyl and monobutyl ethers of ethylene glycol, propylene glycol and diethylene glycol, the monomethyl and monoethyl ethers of triethylene glycol; the dimethyl and diethyl ethers of diethylene glycol, dipropylene glycol and trimethylene glycol. Examples of polyols containing three or more hydroxy groups include glycerin and derivatives of glycerin such as glycerol mono-, di-, and triacetate, or monomethacrylate. Also included is  
20 polyvinylalcohol, which also functions as an adhesive compound. Polyvinylalcohols of molecular weights 89,000-98,000, 85,000-146,000, 124,000-186,000, 31,000-50,000, 85,000-146,000, 124,000-186,000, 13,000-23,000, 50,000-85,000, with various levels of hydrolysis, are available from Aldrich Chemical Company.

30 The polar carrier may also be an alkanolamine and substituted alkanolamine based on ethanol and isopropanol such as mono-, di- and

triethanolamine, mono-, di- and triisopropanolamine, methylethanolamine, dibutylethanolamine, phenyldiethanolamine, di-(2-ethylhexyl)ethanolamine, dimethylisopropanolamine, dibutylisopropanolamine, and the like as well as mixtures thereof.

5 N-Alkyl sulfonamides are also useful carriers.

The present invention is not restricted to the listed carriers, and mixtures of carriers may be utilized, as would be apparent to those of skill in the art. Such polar carriers may comprise about 10 to 90 weight percent of the composition. In a preferred embodiment, the polar carrier comprises about 30 weight percent  
10 of the total composition. In a more preferred embodiment, the polar carrier comprises about 13-30% weight percent, more preferably, about 15-25 weight percent, most preferably, about 20-23 weight percent. At these percentages, very short heating times are possible while retaining acceptable shear strength of the bond.

15 Preferable high dielectric constant carriers are those that can generate heat without being highly volatile, in order to preserve RF susceptor mobility in the composition. Preferred carriers are glycols such as glycerine and N-methyl pyrrolidone (NMP). NMP has a high dipole moment of 4.09 Debye, which produces a dielectric constant, K, of 32.2 at 25°C. NMP is noncorrosive, biodegradable, and almost odorless. NMP has a low order of oral toxicity and is  
20 neither a skin irritant nor a sensitizer. NMP is also an excellent solvent both for a wide range of organic compounds and polymers, as well as for some inorganic salts. In short, it is a very useful medium for dissolving or dispersing susceptors and film formers that are employed in the bonding or adhering of substrates or layers according to the present invention.

25 A further preferred high dielectric constant carrier is glycerine. Glycerine has a dielectric constant of 42.5 at 25°C, is noncorrosive, biodegradable, and odorless. Glycerine is nontoxic and is neither a skin irritant nor a sensitizer. Thus, glycerine is a preferred carrier for consumer products containing adhesives and coatings. Glycerine is also an excellent solvent both for a wide range of organic  
30 compounds and polymers, as well as for some inorganic salts.

A suitable susceptor composition according to the present invention comprises a susceptor present in a concentration of from about 10 % to about 50 % and a carrier present in a concentration of from about 1% to about 75%. Additionally, another suitable susceptor composition further comprises an adhesive compound or other additive as described herein present in a concentration of from about 10 % to about 35%. The susceptor composition can be used to bond or adhere substrates or layers to one another. The substrates can include single layers of polyolefins and non-polyolefins, as well as multilayer stacks. Such stacks may comprise 2, 3, 4, 5 or more layers. One or more susceptor compositions, which may be the same or different, may be between 2 or more layers of the multilayer stacks. All composition concentrations described herein correspond to weight-weight percentages, unless indicated otherwise.

#### *IV. Further Additives to the Susceptor Compositions*

A number of different additives may be added to the susceptor compositions of the present invention including the carrier or mobile medium. In order to provide uniform heating of a susceptor composition, the susceptors are dissolved, distributed, or dispersed, preferably substantially uniformly, in a carrier containing either various polymers and/or solvents or plastisizers. Some carriers, such as solvents, plastisizers, or polymers, are utilized for their polar functionality and for their ability to enhance the heating process.

##### *A. Adhesive/Thermoplastic Additives*

The adhesive properties of the susceptor composition of the present invention are enhanced by the presence of one or more thermoplastic or adhesive compounds, such as polymers or copolymers, that are blended in the susceptor composition. Some of the thermoplastic or adhesive compounds utilized in the present invention include, but are not limited to, polyesters such as a thermoplastic methylol polyester prepared from the reaction of at least one dicarboxylic acid



with a diglycidyl ether, a diglycidyl ester or combination thereof (see U.S. 5,583,187) or a cyanoacrylate/polyester adhesive composition (see U.S. 5,340,873); polyamides; polyurethanes (see U.S. 5,391,602); polysiloxanes; elastomers; polyvinylpyrrolidone; ethylene vinyl acetate copolymers (see U.S. 4,460,728), vinylpyrrolidone vinyl acetate copolymers; vinyl ether copolymers (e.g. polyvinyl methyl ether); polyvinyl alcohol; partially hydrolyzed polyvinyl acetate; copolymers comprising a starch ester (see U.S. 5,498,224) and starch hydrolysates (see U.S. 5,827,553); graft copolymer prepared from a vinyl monomer and a polyalkylene oxide polymer, and a hydroxy-containing ester or acid wax (see U.S. 5,852,080); copolymers comprising a graft copolymer prepared from a vinyl monomer, at least one polyalkylene oxide polymer, a polar wax and other optional ingredients (see U.S. 5,453,144); thermoplastic block copolymers comprising an aromatic vinyl copolymer block, a diene polymer or hydrogenated derivative thereof and other additives (see U.S. 5,723,222); vinyl chloride copolymers; vinylidene chloride copolymers; vinylidene fluoride copolymers; vinyl pyrrolidone homo- and copolymers; vinyl pyridine homo- and copolymers; hydrolyzed polyvinyl alcohol and compositions thereof (see U.S. 5,434,216); cellulose esters (e.g. cellulose acetate and starch acetate, see U.S. 5,360,845) and ethers (e.g. hydroxypropyl cellulose, methyl cellulose, ethyl cellulose, propyl cellulose and the like; see U.S. 5,575,840, 5,456,936 and 5,356,963); modified starch ester containing adhesives (see U.S. 5,360,845); high amylose starch containing adhesive (see U.S. 5,405,437); poly-alpha olefins; propylene homo- and copolymers; ethylene homo- and copolymers (especially those of vinyl acetate, vinyl alcohol, ethyl- and butyl- acrylate, carbon monoxide, acrylic and methacrylic acid, crotonic acid, and maleic anhydride), an alkyl acrylate hot melt adhesive (see U.S. 4,588,767), a hot melt adhesive comprising an alkyl acrylate and an alpha-olefin (see U.S. 4,535,140), a hot melt adhesive comprising an ethylene n-butyl acrylate copolymer (see U.S. 5,331,033), a hot melt adhesive comprising a graft copolymer comprising at least one vinyl monomer and at least one polyalkylene oxide polymer (see U.S. 5,217,798), a vinyl acetate copolymer copolymerized with a cyclic ureido compound (see U.S. 5,208,285), a hydrophilic

polycarbodiimide (see U.S. 5,100,994), a photopolymerized, pressure sensitive adhesive comprising an alkyl acrylate, a monethylenically unsaturated polar copolymerizable monomer, ethylene vinylacetate copolymer and a photo initiator (see U.S. 5,079,047), a hot melt adhesive comprising tackifying resins, oil diluent, and a substantially radial styrene-butadiene block copolymer (U.S. 4,944,993), an adhesive prepared from the vinyl ester of an alkenoic acid, ethylene, a dialkyl maleate, an N-methylol comonomer, and an ethylenically unsaturated mono- or dicarboxylic acid (see U.S. 4,911,960), an adhesive prepared from the vinyl ester of an alkenoic acid, ethylene, a dialkyl maleate, and a monocarboxylic acid (see U.S. 4,892,917), a hot melt adhesive consisting essentially of an ethylene n-butyl acrylate copolymer (U.S. 4,874,804), hot melt adhesive compositions prepared from styrene-ethylene-butylene-styrene tri-block and/or styrene-ethylene-butylene di-block copolymers that are tackified (U.S. 4,822,653), a hot melt packaging adhesive comprising a ethylene n-butyl acrylate copolymer with n-butyl acrylate (U.S. 4,816,306), polysaccharide esters containing acetal and aldehyde groups (U.S. 4,801,699), polysaccharide aldehyde derivatives (U.S. 4,788,280), an alkaline adhesive comprising a latex polymer or a halohydrin quaternary ammonium monomer and starch (U.S. 4,775,706), polymeric fatty acid polyamides (U.S. 4,419,494), hot melt adhesives comprising resins containing 2-methylstyrene, styrene and a phenol (U.S. 4,412,030). The present invention is not restricted to the listed adhesive compounds and compositions, as would be apparent to those of skill in the art.

Such adhesive additives may comprise about 1 to 50 weight percent of the composition, more preferably, about 25 weight percent.

#### ***B. Adhesive/Coating Thermoset Additives***

It is also possible to add a thermoset resin to the susceptor compositions of the present invention. Such thermosets are capable of being cross-linked or cured through heat and/or catalysts and include those described in U.S. Patent No. 5,182,134, e.g. epoxies, polyurethanes, curable polyesters, hybrid thermosets, and

curable acrylics. Others include bismaleimides, silicones, phenolics, polyamids and polysulfides among others. Further examples include maleate resins formed by the reaction of various polyols with maleic anhydride. Orthophthalic resins may be used which are formed by the reaction of phthalic anhydride and maleic anhydride or fumaric anhydride as the dibasic acid. Isophthalic resins may also be used which may be formed by reacting isophthalic acid and maleic anhydride. Others include the bis-phenol fumarides, chlorendic polyester resins, vinyl esters, dicyclopentadiene resins, orthotolyl biguanine, the diglycidyl ether formed from bis-phenol A and epichlorohydrin, triglycidyl isocyanurate thermosetting compositions, bis-phenol A-epichlorohydrin diglycidyl ether cured with phenolic cross-linking agents, aliphatic urethane thermosetting compositions such as an unblocked isofuron diisocyanate-E-caprolactam, BTDA thermosetting compositions which are generally the reaction product of 3,3',4,4'-benzophenone tetracarboxylic dianhydride and a bis-phenol A-epichlorohydrin diglycidyl ether, hybrid thermosetting compositions which are the reaction product of a carboxylated saturated polyester curing agents and bis-phenol A-epichlorohydrin diglycidyl ether, standard bis-phenol A-epichlorohydrin diglycidyl thermosets such as those which are cured from 2-methylimidazole, and standard bis-phenol A-epichlorohydrin diglycidyl ether thermosets which are cured with 2-methylimidazole and dicyandiamide thermosetting compositions. See U.S. Patent Nos. 5,182,134, 5,387,623

Other thermosets and adhesives/coatings that may be added to the susceptor compositions of the invention include a reactive polyurethane prepolymer and 2,2'-dimorpholinoethyl ether or di(2,6-dimethylmorpholinylethyl) ether catalyst (see U.S. 5,550,191), a free radical polymerizable acrylic monomer, diazonium salt/activator composition (see U.S. 4,602,073), a diphenylmethane diisocyanate, a caprolactone triol, a neopentyl adipate ester diol, and, optionally, at least one polypropylene diol together with a catalyst (U.S. 5,057,568), an aqueous polyurethane dispersion comprising an isocyanate-terminated polyurethane prepolymer containing carboxylic acid salt groups, and an active hydrogen containing chain extender (U.S. 4,801,644).

The susceptor compositions of the present invention may also be combined with a shelf stable thermosetting resin as described in U.S. 5,739,184, which is then activated by RF energy to give coatings, e.g. for wood or paper products. This thermosetting resin comprises an epoxy resin, a rosin and an organometallic compound in an amount effective to provide improved adhesion to wood or paper substrates.

Curing agents may also be combined together with the susceptor/thermoset compositions of the invention, including melamines such as dialkyl melamines, amides such as dicyandiamide, adipamide, isophthallyl diamide, ureas such as ethylene thiourea or guanylurea, azides such as thiosemicarbazide, azoles such as guanazole or 3-amino-1,2,4-triazole, and anilines such as dialkylanilines such as dimethyl aniline and diethyl aniline.

Such thermoset additives may comprise about 1 to 50 weight percent of the composition, more preferably, about 25 weight percent.

It has also been discovered that thermoset compositions may be activated with only the polar carrier and without a susceptor. Thus, the invention also relates to compositions comprising a thermoset and a polar carrier. The thermoset may comprise about 60 to 95 weight percent of such a composition. The polar carrier may comprise about 5 to 40 weight percent. The invention relates as well to methods of bonding, adhering or coating substrates with such thermoset/polar carrier compositions.

### *C. Surfactant Additives*

According to another embodiment of the present invention, surfactant additives can be added to the susceptor composition to enhance the ability to draw down the susceptor composition of the present invention onto the layers or substrates to be bonded, adhered or coated. Depending on the types of materials that are to be joined or coated, surfactant additives, such as SURFYNOL 104PA (available from Air Products Corporation) and SURFADONE LP 300 (N-dodecyl-2-pyrrolidone, available from International Specialty Products), can be

used to wet a variety of substrates such as Mylar and polyethylene (PE). A further plasticizer is *p*-toluenesulfonamide, a good plasticizer that also dissolves stannous chloride. The present invention is not restricted to the listed surfactant additives, as would be apparent to those of skill in the art. Such surfactants may  
5 comprise about 0.1 to 5 weight percent of the composition.

#### ***D. Plasticizer Additives***

The susceptor compositions of the present invention may further comprise a plasticizer to modify the flexibility of the adhesive or coating. Examples of such plasticizers include, but are not limited to acetyl tributyl citrate, butyl benzyl  
10 phthalate, butyl phthalyl butyl glycolate, dibutyl phthalate, dibutyl sebacate, diethyl phthalate, diethylene glycol dibenzoate, dipropylene glycol, dipropylene glycol dibenzoate, ethyl phthalyl ethyl glycolate, ethyl-*p*-toluene sulfonamide, hexylene glycol, methyl phthalyl ethyl glycolate, polyoxyethylene aryl ester, tributoxyethyl phthalate, triethylene glycol polyester of benzoic acid and phthalic acid, glycerin,  
15 or mixtures thereof. Other plasticizers that may be used include N-methyl-2-pyrrolidone (NMP), and substituted toluene sulfonamides (e.g. *p*-toluenesulfonamide, RIT-CIZER #8<sup>TM</sup> and RIT-O-LITE MHP<sup>TM</sup> from Rit-Chem Co., Inc., Pleasantville, NY). Such plasticizers may comprise about 1 to 40 weight percent of the composition.

#### ***E. Tackifiers***

The tackiness of the compositions of the invention may be increased by the addition of a suitable tackifier, e.g. one or more of hydrogenated aromatic petroleum resins, hydrogenated aliphatic petroleum resins, and hydrogenated terpene resins (see U.S. 5,418,052), coumarone-indene, ester gum, gum rosin,  
25 hydrogenated rosin, phenolic modified hydrocarbon resins, rosin esters, tall oil rosins, terpene phenolic, terpene resins, toluene sulfonamide-formaldehyde resin, wood rosin (see U.S. 5,442,001), distilled rosin, dimerized rosin, maleated rosin,

polymerized rosin (see U.S. 5,532,306). Other tackifiers and modifiers, include (but are not limited to) styrene and alpha methyl styrene resins, glycerol and pentaerythritol esters, etc. Particular tackifiers include WINGTACK 95 from Goodyear, Herculin D and PICCOLYTE C from Hercules, EASTOTACK H100 from Eastman, and ECR 149B or ECR 179A from Exxon Chemical (see U.S. 5,559,165). Other tackifiers include rosin and its derivatives available from Reichold Chemicals, Manila Copal (softening point 81-90 °C acid No. 110-141), Pontianac (softening point 99-135 °C acid No. 1120129) and Sanarec (softening point 100-130 °C, acid no. 117-155). Such tackifiers may comprise about 1 to 25 weight percent of the composition.

#### ***F. Fillers***

A number of different fillers may be added to the susceptor compositions of the invention, including, but not limited to cellulose, bentonite, calcium carbonate, calcium silicate, clay, mica silica, talc, alumina, glass beads, fibers and the like. Such fillers may comprise about 0 to 40 weight percent of the composition.

#### ***G. Stabilizers and Antioxidants***

Stabilizers and antioxidants may be added to the susceptor compositions of the invention in amounts effective to achieve the intended result. Included among such stabilizers include high molecular weight hindered phenols and multifunctional phenols such as sulfur and phosphorous-containing phenols. Representative hindered phenols include 1,3,5-trimethyl-2,4,6-tris(3,5-di-tert-butyl-4-hydroxybenzyl)benzene, pentaerythritol tetrakis-3-(3,5-di-tert-butyl-4-hydroxypropionate, n-octadecyl-3,5-di-tert-butyl-4-hydroxyphenyl)propionate, 4,4'-methylenebis(2,6-di-tert-butylphenol), 4,4'-thiobis(6-tert-butyl-o-cresol), 2,6-di-tert-butylphenol, 6-(4-hydroxyphenoxy)-2,4-bis(n-octylthio)-1,3,5-triazine, di-n-octadecyl-3,5-di-tert-butyl-4-hydroxybenzylphosphonate, 2-(n-octylthio)ethyl-

3,5-di-tert-butyl-4-hydroxybenzoate, and sorbitol hexa[3-(3,5-di-tert-butyl-4-hydroxyphenyl)propionate (see U.S. 5,574,076) . Such stabilizers and antioxidants may comprise about 0.01 to 5 weight percent of the composition.

#### **H. Other Additives**

According to another embodiment of the present invention, other types of additives to the susceptor composition may include flow aids, heat and UV stabilizers, coupling agents, waxes, pigments and other organic compounds. For example, in some instances, waxes can facilitate lower melt temperatures. Waxes that can be utilized include, but are not limited to, Bees wax (SYNCHROWAX BB4), Candelilla wax, CARBOWAX 3350 (available from Union Carbide Corporation), Carnauba wax, and CASTORWAX NF. Other waxes include N-(2-hydroxyethyl)-2,2'-ethylene-bis-stearamide, stearamide, 12-hydroxystearamide wax, hydrogenated castor oil, oxidized synthetic waxes, poly(ethylene oxide) having a molar average molecular weight of above about 1000, and functionalized synthetic waxes such as carbonyl containing ESCOMER H101 from Exxon (see U.S. 5,532,306). Preferably, the wax polar as described in U.S. Patent 5,750,605. A preferred polar wax is Peracin 200. Also preferably, the polar wax is present at no more than about 25%, more preferably, no more than 17% of the composition, most preferably, no more than 10% of the composition.

Other additives include elastomers such as those described in U.S. 5,506,298, 5,739,184, 5,169,890, 5,039,744, 4,761,198 may be used, including styrene butadiene rubber, polybutadiene rubber, rubber, nitrile rubbers, butyl rubber and halogenated butyl rubber.

When the compositions are applied and activated as coatings, they may further comprise one or more additives to impart color to the composition. Such additive include, without limitation, titanium dioxide, iron oxide pigments, carbon black and organic pigments such as isoindoline yellow.

The present invention is not restricted to the listed additives, as would be apparent to those of skill in the art. Such other additives may comprise about 1 to 25 weight percent of the composition.

***V. Applying the Susceptor Compositions to Substrates***

5           The compositions of the invention may be formulated to be applied as a liquid at room temperature, hot melt, or powder. Liquid compositions may be solvent borne or water-borne. The liquid applied compositions may be applied as a liquid at room temperature and dried down to give the desired coating. The liquid applied coating may be applied to a substrate by any conventional method  
10           including spraying, ink-jet, brushing, rolling, gravure printing, dripping and the like. Methods of actively drying down liquid compositions include but are not limited to conventional oven drying, forced air, heat lamps, microwave heating, RF heating or various combinations of these or other methods. When a liquid composition is dried down, it loses some or all of its volatiles. RF drying of a  
15           liquid applied composition may be accomplished by applying RF energy across the composition in order to generate sufficient heat within the liquid to facilitate or enhance the evaporative loss of water or solvent(s). The RF energy can be applied across the liquid at constant, intermittent, or gradient intensities to achieve the desired rate and degree of drying. Similarly, other methods of drying may be  
20           applied at constant, intermittent or gradient intensities to achieve the desired drying result.

          Hot melt applied systems are applied in their molten state at an elevated temperature and then cooled to yield the desired solid coating. The hot melt compositions can be heated to a molten state by various methods including but not  
25           limited to conventional melt tanks, microwave heating and RF heating. Once the hot melt composition is melted, it may be applied in a variety of different types of hot melt coatings, including but not limited to spirals and beads, hot blown, slot coat, and co-extrusion. After application, the molten hot melt composition can be passively or actively cooled to return to its solid form. Active cooling may be



accomplished by blowing cool air across the applied material, or by allowing the substrate to make contact with a heat-sink surface.

Powdered applied systems are applied in their "fine" particle state (1-20  $\mu\text{m}$ ) by electrostatic spray or gun. The applied layer is activated by RF energy as in liquid or hot-melt systems.

Once dried and/or cooled, the substrate may be stored until activation of the composition is desired. Many of the applied compositions of the invention are substantially non-tacky and may be applied to a substrate which is then rolled up. Upon unrolling and activating, the substrate may be adhered to one or more other substrates. Those compositions that are tacky may be activated immediately after being applied and dried if necessary. Alternatively, they may be covered with a removable strip or dusted with talc or similar material.

One aspect of the invention also relates to a method for applying a susceptor composition to a substrate, comprising:

(1) formulating the susceptor composition as a liquid dispersion;

(2) applying the liquid dispersion of the susceptor composition to the substrate;

(3) drying the susceptor composition, wherein the drying step includes the step of applying RF energy across the composition, thereby generating heat within the liquid dispersion. In a preferred embodiment, one may roll up the substrate after the susceptor composition has dried.

The susceptor compositions may be applied to any conventional substrates including, without limitation, woven and nonwoven substrates such as polyolefins, such as PP and PE webs, non-wovens, films and the like, cellulose substrates prepared from, for example, wood pulp (such as paper, cardboard and the like), cotton fibers (e.g. textiles such as cloth, sheeting and industrial fabrics), glass, ceramic surfaces, rubber and synthetic polymeric substrates such as polyester or polyolefin substrates prepared from, for example, polypropylene and polyethylene, polyvinyl alcohol, polyhydroxyvalerate butyrate, polylactides, cellulotics, polyamides, polyvinyl chloride, polystyrene, acrylics, synthetic textile products,

etc. and any combination of the aforementioned. Other substrates include metal (e.g. aluminum foil and other metal foils), wood, composites, etc.

## ***VI. Apparatus For Activating the Various Compositions of the Present Invention***

5           Generally, the compositions of the present invention may be heated (i.e., activated) by any system capable of generating an electromagnetic field of sufficient strength and frequency.

10           FIG. 4 illustrates a high level block diagram of an RF heating system 400 that is capable of generating an electromagnetic field for activating the compositions of the present invention. Heating system 400 includes an RF power supply 402 that provides about a 1 kW, 1 to 15 MHz, RF signal 404 to a heat station 406. Heating system 400 also includes an inductor 408 that is coupled to RF power supply 402 through heat station 406. Generally, heat station 406 includes a capacitor connected either in series with or parallel to inductor 408.

15           RF signal 404 provided to heat station 406 by RF power supply 402 creates an alternating current flowing through inductor 408, which creates an electromagnetic field. Heating of a sample 410, which is or includes a composition of the present invention, occurs when sample 410 is placed in proximity to inductor 408. The best heating takes place when sample 410 is placed near the proximal (or "terminal") end 411 of inductor 408, and little or no heating occurs when sample 410 is placed at the distal (or "turn") end 412 of inductor 408. Further, there is a heating gradient from terminal end 411 to turn end 412. In theory and without limitation, the best heating occurs at the terminal end 411 because it is believed that the intensity of the electric field component of the electromagnetic field at terminal end 411 is greater than at the distal end 412.

20           FIG. 5 illustrates a high level block diagram of another embodiment of a heating system 500 that is capable of generating an electromagnetic field for activating the compositions of the present invention. Heating system 500 includes an alternating voltage generator 502 and a probe 504, which is connected to an output terminal 501 of voltage generator 502. Voltage generator 502 alternately

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positively charges and negatively charges probe 504, thereby creating an electromagnetic field 506 centered at probe 504. Heating can occur when sample 410 is placed in proximity to probe 504. How quickly and how much heating occurs depends on the sample itself, the strength of the electromagnetic field at the sample, and the frequency of the alternating voltage 509 produced by voltage generator 502.

Generally, probe 504 is a conductive material, such as, but not limited to copper, aluminum, or stainless steel. Generally, probe 504 can have a variety of shapes, including cylindrical, square, rectangular, triangular, etc. Preferably, probe 504 is square or rectangular. Probe 504 can be hollow or solid, preferably hollow. Generally, probe 504 can be straight or non-straight, such as curved. The preferred characteristics of probe 504 ultimately depends on the application that it is being used for.

In yet another embodiment, which is illustrated in FIG. 6, heating system 500 includes at least two probes 602 and 604 for activating the compositions of the present invention. Probe 602 is connected to output terminal 610, and probe 604 is connected to output terminal 612. Like probe 504, probes 602 and 604 are made from conductive materials as discussed above. Probes 602 and 604 can have a variety of shapes. For example, they can be either straight or curved. Preferably, at least a portion of probe 602 is parallel to a portion of probe 604, although not a required.

In the system shown in FIG. 6, probe 602 has a net positive charge when probe 604 has a net negative charge, and probe 602 has a net negative charge when probe 604 has a net positive charge. When probes 602 and 604 are oppositely charged, a strong electromagnetic field 606 is present between the probes. Thus, sample 410 is preferably heated by placing it in a region above (or equivalently below) the region between probe 602 and probe 604, as illustrated in FIG. 7A and 7B. This region is referred to as an activation region. Preferably, an insulating layer 702 (see FIG. 7A) is placed between sample 410 and probes 602 and 604, although this is not a requirement.

Generally, the vertical distance between sample 410 and probes 602 and 604 ranges from about .01 to 2 inches, more preferably from about .02 to 1 inch, and most preferably from about .025 to .185 inches. Sample 410 can also be heated by placing it between probes 602 and 604. Generally, The center to center distance between probes 602 and 604 ranges from about 0.1 to 3 inches, more preferably from about 0.2 to 2 inches, and most preferably from about 0.25 to 0.75 inches. Additionally, in general, the height and width of a rectangular probe, or the diameter for a cylindrical probe, ranges between about 0.02 and 0.5 inches, and the length generally ranges from about 0.25 inches to 20 feet.

In one embodiment, the distal end 750 of probe 602 is curled to reduce corona effect (see FIG. 7C). For the same reason, the distal end of probe 604 is also curled.

An advantage that the two probe system shown in FIG. 6 has over the system shown in FIG. 4, is that sample 410 heats equally as well at the proximal end of probes 602, 604 as it does at the distal end. Consequently, the system of FIG. 6 does not experience the heating gradient problem that is encountered with the system of FIG. 4.

Generally, the compositions of the present invention may be activated by a frequency of the alternating voltage 509 ranging from about 1 KHz to 5 GHz, more preferably from about 1 MHz to 80 MHz, and most preferably from about 10 to 15 MHz. The peak to peak voltage between probes 602 and 604 generally ranges from 1 to 15 kilo volts (kV). Generally, the duration of RF energy application to the sample 410 (also referred to as dwell time), for most applications, ranges from about 100 milliseconds to 30 seconds. However, there are some applications where the dwell time greatly exceeds 30 seconds. In the case of a composition comprising a thermoset resin, the dwell time ranges from about 1 second to 20 minutes, preferably from about 1 to 10 minutes, and most preferably from about 2.5 to 5.0 minutes to initiate cross linking reactions(s) leading to a high degree of thermoset character.

FIG. 8 illustrates one embodiment of alternating voltage supply 502. The invention, however, is not limited to this or any particular voltage supply, since

any system capable of generating a strong enough electromagnetic field could be utilized to activate the compositions of the present invention. In one embodiment, voltage supply 502 includes direct current (DC) voltage source 802 that is connected to a broadband amplifier 806 through DC power rail 804. The function of DC voltage source 802 is to provide a DC voltage to broadband amplifier 806. The DC voltage produced by DC voltage source 802 can range from 0 volts to 200 volts. The magnitude of the voltage provided to broadband amplifier 806 is dependent upon an output signal 815 from a main controller 814. Output signal 815 of main controller 814 can be controlled manually by a user 821 through user interface 820, or automatically by a production line control system 822.

Broadband amplifier 806 amplifies a low level RF signal 817 generated by frequency controller 816, and thus generates a high level RF power signal 808. Preferably, the frequency of RF signal 817 ranges between 10 MHz and 15 MHz. RF signal 808 is passed through a power sensor 810 and provided to an impedance matching circuit 812 (also referred to herein as "heat station") through an RG393 50 ohm cable 811. Upon RF signal 808 being inputted into impedance matching circuit 812, an electromagnetic field 606 is generated at the probes 602 and 604. This electromagnetic field is used to heat the compositions of the present invention.

While RF signal 808 is applied to impedance matching circuit 812, power sensor 810 continuously feeds a reflected power signal 832 to frequency controller 816 and main controller 814. Power sensor 810 also continuously feeds a forward power signal to main controller 814. Reflected power signal 832 represents the amount of reflected power and forward power signal 830 represents the amount of forward power.

Frequency controller 816 uses reflected power signal 832 to continually adjust the frequency of RF signal 817 so as to minimize the amount of reflected power. Main controller 814 uses forward power signal 830 and reflected power signal 832 to maintain the power level set by user 821 through user interface 820 or set by production line control system 822. Main controller maintains the correct power level by adjusting the level of DC voltage supplied by DC voltage

source 802 and by adjusting the output level of RF signal 817 generated by frequency controller 816.

As sample 410 changes during a heating process, the impedance on the probes 602 and 604 change, which causes a change in the forward and reflected power. Frequency controller 816 will detect this change in reflected power because it is receiving reflected power signal 832 from power sensor 810. Frequency controller 816 changes the frequency of RF signal 817 so as to minimize reflected power, thereby achieving an optimum impedance match and insuring a repetitive power transfer from heating system 800 to sample 410.

DC voltage source 802, sensor 810, frequency controller 416, and main controller 814 are further described in U.S. Patent Application No. 09/113,518, entitled, "RF Power Supply," which is incorporated herein by reference in its entirety. A broadband amplifier suitable for use in heating system 800 is described in U.S. Patent Application No. 09/270,506, filed March 17, 1999, entitled, "High Frequency Power Amplifier," which is incorporated in its entirety herein by reference.

FIG. 9 is a flow chart illustrating a process for heating a composition according to the present invention using heating system 800. The process begins with step 902 when user 821 or production line control system 822 sends a "heat-on" signal to the main controller 814. Upon receiving the "heat-on" signal, main controller 814 begins an initial tuning process for determining the frequency of RF signal 817 that produces the minimum amount of reflected power. The initial tuning process encompasses steps 904-908. In step 904, main controller 814 directs DC voltage source 802 to output a "tune" voltage. The "tune" voltage is the lowest voltage level that can provide a sufficient signal to measure the reflected power over a range of frequencies. The objective is to consume the least amount of energy during the initial tuning process. Typically, the "tune" voltage level is 10% of the full scale voltage, where the full scale voltage is the voltage at which the composition is intended to be heated.

After step 904, control passes to step 906. In step 906, heating system 800 performs course tuning. That is, heating system 800 determines a course

estimate (i.e., rough estimate) of the frequency that produces the minimum reflected power. Hereafter this frequency shall be referred to as the resonant frequency. The course estimate of the resonant frequency can be determined by sampling reflected power over a first predetermined frequency range. After step 5 906, control passes to step 908. In step 908, the heating system 800 performs fine tuning. That is, the heating system 800 determines a fine estimate (i.e., more precise estimate) of the resonant frequency. The fine estimate can be determined by sampling the reflected power over a second frequency range, which includes the course estimate of the resonant frequency. After step 908, control passes to 10 steps 910 and 912 in parallel. In step 910, main controller 814 ramps (i.e., rapidly increases) the voltage output by the DC voltage source 802 such that within approximately 30 milliseconds the voltage increases from the "tuning" voltage level to approximately the full scale voltage level. In step 912, the heating system 800 continuously tracks the resonant frequency until a power off indication is 15 received. The methods for course tuning, fine tuning, and tracking resonant frequency are described in U.S. Patent Application No. 09/113,518.

FIG. 10A further illustrates one embodiment of impedance matching circuit 812. Impedance matching circuit 812 is used to match the impedance of 50 ohms on the input to the variable impedance of the probes 602 and 604 and 20 sample 410. The impedance of the probes 602 and 604 and sample 410 is typically in the order of 200 to 500 ohms. The impedance of the sample has an equivalent circuit of a resistance between 500 Ohms and 50 Kilo Ohms in parallel with a 0.1 picofarad capacitor.

Circuit 812 includes a connector 1001, two capacitors 1002 and 1004, 25 and an inductor 1006. Capacitor 1002 is a variable capacitor, which is adjustable from 10 to 50 picofarades (pf) to achieve impedance match to the varying impedance of probes 602 and 604 and sample 410. The capacitance of capacitor 1004 is preferably 100 pf, and the inductance of inductor 1006 is preferably 1.0 micro henries ( $\mu$ H). Capacitor 1004 and inductor 1006 form a parallel resonance 30 circuit that will resonate typically at a frequency between 12.5 and 14.5 MHz. Capacitor 1004 and inductor 1006 are water cooled with a flow rate of

approximately half a gallon per minute. Probe 602 is connected to a node 1020 of circuit 812, and probe 604 is connected to a node 1022 of circuit 812. The high power RF input 411 (typically less than 1 kilowatt) from a 50 ohm source generator is connected to connector 1001.

5           A process for setting the capacitance of variable capacitor 1002 will now be described. The process begins by applying a low level RF signal (typically 10 watts) to input 1001 of circuit 812. The frequency of the applied RF signal is adjusted until the amount of reflected power is minimized. The capacitance of capacitor 1002 is then adjusted to optimize the reflected power minima. To  
10           achieve the least amount of reflected power that is practical to achieve, which is about two percent reflected power (or 1.25 voltage standing wave ratio (VSWR)), the frequency of the applied RF signal and the capacitance of capacitor 1002 are adjusted in an iterative process. Once the process is completed, sample 410 is placed in proximity to probes 602 and 604. At this point it may be necessary to  
15           adjust the frequency of operation and capacitor 1002 in order to achieve an optimum reflected power. Once optimum reflected power is achieved, the power level of the input RF signal is increased. As the input RF power level is increased the resonant frequency of the matching circuit and probes 602 and 604 and sample 410 will change requiring a change of operating frequency to continue to minimize  
20           the reflected power.

FIG. 10B illustrates another embodiment of impedance matching circuit 812. In this embodiment, impedance matching circuit 812 includes a connector 1051, a 1:1 balun transformer 1052, two variable capacitors 1054 and 1056, and one inductor 1060. Capacitors 1054 and 1056 are adjustable from about 3 to 25  
25           picofarades (pf) to achieve impedance match to the varying impedance of probes 1082 and 1084 and sample 410. The inductance of inductor 1060 is preferably about 5 micro henries (mH). Capacitor 1058 is not an actual circuit element of the impedance matching circuit 812. Capacitor 1058 represents the capacitance associated with the inductor system, which consists of the inductor 1060, the leads  
30           (not shown) connecting the inductor 1060 to the capacitors 1054 and 1056, and the leads (not shown) connecting the probes 1082 and 1084 to the inductor 1060.



The capacitance of capacitor 1058 is preferably less than about 15 pf. One advantage the impedance matching circuit 812 shown in FIG. 10B has over the impedance matching circuit shown in FIG. 10A, is that the impedance matching circuit 812 shown in FIG. 10B provides a balanced signal on the probes 1082 and 1084 relative to ground.

## ***VII. Method of Bonding Substrates***

The compositions of the present invention may be employed in a variety of bonding methods, including but not limited to adhesive bonding, thermal bonding and mechanical bonding.

Adhesive bonding is accomplished when a susceptor composition is interposed between two substrates that are to be joined (adherands) and activated by RF energy to undergo adhesive attachment to each of the adherands.

In the case of thermoplastic adhesive compositions such as hot melts, RF energy causes the composition to melt and wet-out onto adherands that are in close contact. Upon cooling, the composition returns to a solid state with sufficient cohesive strength and adhesion to each of the adherands to form a good bond. The degree of heating and melting of the adhesive composition is controlled by the intensity and duration of the applied RF energy and the formulation of the adhesive composition. Such control is required to prevent undesired results stemming from under-heating or over-heating the adhesive composition. For example, under-heating can result in a weak bond due to insufficient wet-out of the adhesive onto the adherands. Also, over-heating can result in undesirable bond, with thermal distortion or destruction of the adherands, as well as thermal degradation of the thermoplastic composition.

In the case of thermoset adhesive compositions, RF energy causes the composition to become cured, resulting in sufficient increase in cohesive strength and adhesion to adherands to form a strong bond. As in the case of thermoplastic compositions, the degree of heating and curing of thermoset compositions is controlled by the intensity and duration of the applied RF energy. Such control

is required to prevent undesired results from under-heating or over-heating. For example, under-heating can result in a weak bond due to insufficient cross-linking. Over-heating can cause effects such as thermal distortion or destruction of the adherands, as well as thermal degradation and excessive shrinkage of the thermosetting composition.

Thermal bonding is accomplished when the composition is used to generate sufficient heat to cause one or more adherands to become thermally fused to each other.

One example of thermal bonding involves saturating a porous thermoplastic material, such as a non-woven polypropylene web, with an RF-heatable composition, and then interposing the saturated web of material between two adherands and RF-heating the composition to cause the saturated web and adjacent adherands to melt and fuse to each other.

Another example of thermal bonding involves saturating a porous, first thermoplastic adherand with an RF-heatable composition, and then placing the first adherand against a second thermoplastic adherand and RF-heating the composition to cause the first and second adherands to melt and fuse together.

FIG. 11 shows a method of bonding polyolefin and non-polyolefin materials using a composition that is activated in the presence of RF energy according to the present invention.

In step 1102, adherands that are to be bonded or adhered are chosen. Once the materials or layers are chosen, an appropriate composition is prepared in step 1104. For example, if nonwoven PP layers are chosen to be bonded, a susceptor, which includes an ionomer as described herein, is combined with a polar carrier. The type of composition may depend on whether a transparent, translucent, or lightly colored adhesive obtained by the method of the present invention is needed for a particular application. After the composition is prepared in step 1104, control can pass to step 1106, 1109, or 1110.

In step 1106, a second carrier, such as an insoluble porous carrier (e.g., nonwoven PP), is saturated with the prepared composition. In step 1108, the saturated insoluble porous carrier is then placed in between the layers chosen to

be bonded. RF energy is applied in step 1120. The RF energy applied in step 1120 can be applied for 100 milliseconds to several minutes. The application of RF energy allows for the precision heating of the layers to be bonded, without the unwanted side effects of non-uniform bonding, or damage to the bonded layers.

5 In step 1110, one or both of the layers to be bonded are coated with the composition. In step 1112, the composition is allowed to dry or the hot melt to congeal depending on the type of composition created. Alternatively, a heat source (e.g. an oven or lamp) and fan may be used to dry the coating or RF energy may be applied to drive off any water or other solvents. According to step 1114,  
10 the layers to be bonded are placed together, such that the coated surfaces are in contact. Uniform pressure placed on the contacted layers helps enhance the bonding or adhesion process activated by the applied RF energy (step 1120). Such uniform pressure may be applied while the composition is being activated or immediately thereafter by use of conventional nip rollers.

15 In step 1109, a film of the composition is created. Such a film can be created according to film making processes well known in the art. The film made in step 1109 can then be sandwiched between the two materials to be bonded in step 1111. RF power is then applied in accordance with step 1120.

20 In a further embodiment, two or more adherands may be bonded or adhered by a process comprising: applying a first composition onto a first adherand; applying a second composition onto a second adherand; contacting the first composition with the second composition; applying RF energy to the first and second compositions to heat the compositions, thereby causing the first and second adherands to become adhered or bonded; wherein one of the compositions  
25 comprises a susceptor and the other of the susceptors is a polar carrier, and the susceptor and/or the carrier are present in amounts effective to allow the first and second compositions to be heated by RF energy.

30 In this embodiment of the invention, the susceptor and carrier components of the composition are applied separately to the adherands prior to placing the adherands together. FIG. 52 shows a susceptor-coated adherand 5201 assembled to an adherand 5203 coated with the polar carrier. After coating one or both of

the adherands, one may apply a temporary release liner 5205 to the coated side to allow the coated adherand to be rolled up or stacked. Alternatively, one may dry one or both coatings.

After nipping the two coated adherands in the assembly stage, the assembly is passed through an RF field 5207 for activation. The RF energy causes the susceptor and carrier to heat with the resulting adhesion between the two adherands. The final nip rollers 5209 press and bonds the two adherands, while cooling the bond line.

FIG. 53 shows the replacement of the pre-applied polar carrier on the adherand with a polar carrier spray coated onto the adherand just prior to the assembly nip rollers 5206. A polar carrier is applied (e.g. sprayed or otherwise as described herein) by a spray applicator 5302 onto adherand 5201. When assembled with the susceptor coated adherand 5203 and exposed to RF energy, the interfaced composition activates to form a bond.

### ***VIII. Additional Probe Embodiments***

Additional embodiments of probes 602 and 604 are described below with reference to FIGs. 12 to 17. These additional embodiments are in no way limiting and merely provide additional examples of possible configurations of the probes.

In FIG. 12, probes 602 and 604 are each curvilinear and oppositely charged. In this particular example, probes 602 and 604 are sinusoidally or "S" shaped, but any similar arrangement is possible. Probes 602 and 604 are made from conductive materials, as described above, preferably, but not limited to, copper, aluminum, or stainless steel. Probe 602 includes a proximal region 1206, and activation region 1208 and a distal region 1210. Similarly, probe 604 includes a proximal region 1212, an activation region 1214, and a distal region 1216. In proximal regions 1206 and 1212, probes 602 and 604 are spaced apart in order to prevent arcing. The amount of spacing depends on the size of probes 602 and 604, and in one example, probes of 0.125 inch square cross-section should be spaced at least 1.1875 inches apart. Similarly, distal regions 1210 and 1216 are

spaced apart to prevent arcing, the amount of such spacing is similarly dependent upon the size of the probes. In activation regions 1208 and 1214, probes 602 and 604 are in proximity to one another in order to create an electromagnetic field between the probes. How close probes 602 and 604 must be to one another again depends on the size of the probes and the magnitude of the charge on them. In one example, probes 602 and 604 have about a 0.125 inch square cross-section and preferably spaced between 0.25 and 0.75 inches apart. It is preferable the space between probes 602 and 604 remains substantially equal throughout the activation region, but it is not necessary. An activation zone 1222 is defined in activation regions 1208 and 1214 between an outermost end 1218 of probe 602 and an outermost end 1220 of probe 604. Activation zone 1222 is indicated in dashed lines in FIG. 12. Activation zone 1222 defines the area of sample 410 that can be heated/activated by the system when the substrates being joined are moving in the direction indicated. If the substrates are stationary with respect to the probes, the activation zone is defined by the area in between the probes.

In another embodiment, probes 602 and 604 may be repeated in order to provide a larger activation zone. Such an arrangement is shown in FIGs. 13A, 13B and 13C. For example, in FIG. 13A, a pattern of one probe 602 and two probes 604 is provided. This arrangement may include any number of probes 602 and 604, as long as oppositely charged probes are placed next to one another. This arrangement works equally well with multiple sets of curvilinear probes, as shown in FIG. 13B.

FIG. 13C illustrates one embodiment of what is termed an "interdigitated probe system." The interdigitated probe system 1301 is advantageous because it provides an extended activation zone, as shown by the dotted rectangle 1350. Interdigitated probe system 1301 includes a first element 1302 and a second element 1304.

The first element 1302 includes a first conductor 1310 and one or more second conductors 1322 connected to the first conductor 1310. Preferably, conductors 1322 are coplanar and uniformly spaced apart, but this is not a

requirement. Additionally, in one configuration of element 1302, each conductor 1322 forms a right angle with conductor 1310, but this is also not a requirement.

Similarly, the second element 1304 includes a first conductor 1312 and one or more second conductors 1320 connected to the first conductor 1312. Preferably, conductors 1320 are coplanar and uniformly spaced apart, but this is not a requirement. Additionally, in one configuration of element 1304, each conductor 1320 forms a right angle with conductor 1312, but this is also not a requirement.

In one embodiment, first element 1302 and second element 1304 are orientated such that conductors 1320 are coplanar with conductors 1322 and each conductor 1320 is adjacent to at least one conductor 1322. First element 1302 and second element 1304 are not limited to any particular type of conductive material. However, conductors 1310, 1312, 1320, and 1322 are preferably copper, and more particularly, copper tubes. In one embodiment, the copper tubes have a one-eighth of an inch diameter.

In one embodiment, the length of conductors 1310 and 1312 is about 40 inches, and the length of conductors 1320 and 1322 is about 2 inches. However, conductors 1310, 1312, 1320, and 1322 are not limited to any particular length. Typically, the length of conductors 1310 and 1312 ranges between about 3 inches and 80 inches, and the length of conductors 1320 and 1322 ranges between about 1 inch and 70 inches.

FIG. 14 shows another embodiment of a probe system for activating a multi-sided sample 1402. In this embodiment, sample 1402 is mounted on a block 1404. Sample 1402 may be mounted on any similar device which allows each side of sample 1402 to be exposed to moving probe blocks 1406. This particular example shows a three-sided sample exposed to three moving probe blocks 1406, however, the sample may include more sides and be exposed to an equivalent amount of moving probe blocks. Probe blocks 1406 include probes 602 and 604 mounted in an electrically insulating material such as, but not limited to, polytetrafluoroethylene (TEFLON™). Probes 602 and 604 are mounted on pressure plates 1408 of probe blocks 1406. In this particular example, three

probes are used in each probe block 1406, two negatively charged probes 604 and one positively charged probe 602. However, more or less probes can be used, depending on the size of the probe blocks, as long as adjacent probes are oppositely charged. Probes 602 and 604 are coupled to an alternating voltage supply 502, via output terminals 610 and 612 as generally shown in FIG. 6. Probe blocks 1406 are moved into proximity of sample 1402 mounted on block 1404, preferably between 0.125 and 0.375 inch, thereby activating the compositions of the present invention, as previously described. Alternatively, probe blocks 1406 could be placed at the appropriate interval and block 1404 with sample 1402 could be moved into position to be activated. While FIG. 14 shows the probe blocks 1406 as having a regular shape, one skilled in the art will recognize that the probe blocks could be any three dimensional shaped object.

FIG. 15 shows another embodiment for activating a multi-sided sample 1502 using a stationary probe system. In this embodiment, probes 602 and 604 are mounted on multiple sides of a single probe block 1504, similar to the manner in which probes 602 and 604 were mounted in probe blocks 1406, described above. In this particular example, probes 602 are mounted on three sides of a generally square probe block 1504, but probes 602 and 604 could be mounted on multiple sides of any polygonal block or three dimensional object. Sample 1502 is brought into proximity of probe block 1504 by pressure plates 1506, thereby activating the compositions of the present invention, as previously described. In this particular example, two negatively charged probes 604 and one positively charged probe 602 are shown on each side of probe block 1504, however, it will be recognized that any number of probes could be utilized, depending on the application, as long as adjacent probes are oppositely charged. Probes 602 and 604 are coupled to an alternating voltage source 502 via output terminals 610 and 612, as generally depicted in FIG. 6.

FIGs. 16A and 16B show yet another embodiment of a probe system for activating a sample material including compositions of the present invention. In FIGs. 16A and 16B, sample 1602 is draped over a conveyor rod 1604 and generally moves along the circumference of the conveyor rod. Conveyor rod 1604

is constructed of electrically non-conductive material. A probe system 1606 is disposed in proximity to a portion of the circumference of conveyor rod 1604, e.g., 0.02 to 1.5 inches, and more preferably within 0.125 to 0.375 inch, and is shaped to conform to the shape of conveyor rod 1604, as best seen in FIG. 16B. Probe system 1606 includes adjacent, alternately charged probes 602 and 604 for activating sample 1602. Probes 602 and 604 are coupled to an alternating voltage source 502, as generally depicted in FIG. 6.

The probe systems described above all activate a single side of the sample material. However, probe systems could be placed on both sides of the material in each of the above-described embodiments, provided that the polarity of the probes is such that the electromagnetic fields do not cancel each other out. A particular example of an activation system for activating both sides of the material is shown in FIG. 17. Rather than using a probe system, two oppositely charged conductive plates 1702 (positively charge) and 1704 (negatively charged) are disposed on opposite sides of sample material 1706. Plates 1702 and 1704 are preferably constructed of copper, but may be constructed of any suitable conductive material, such as the aforementioned conductive materials of probes 602 and 604. Sample material 1706 may be stationary or moving when exposed to the activation region between plates 1702 and 1704. Plates 1702 and 1704 are preferably spaced between 0.02 and 24 inches, more preferably between 0.02 and 15 inches, and most preferably between 0.05 and 0.375 inches. Plates 1702 and 1704 are coupled to an alternating voltage source 502 via output terminals 610 and 612, as generally depicted in FIG. 6.

***IX. Applicator System for Applying a Composition of the Present Invention to a Substrate/Adherand***

FIG. 18 illustrates one embodiment of an application system 1800 for applying a composition according to the present invention to an adherand 1810. The manufacturing system includes an applicator 1815. Applicator 1815 applies a hot melt or liquid dispersion or powder of the composition 1812 to one side of adherand 1810. Composition 1812 may be applied via a hot melt by applying heat



to the composition 1812 so that it reaches its melting point and can be applied to an adherand. In a hot melt application heat is applied to the composition 1812 in the applicator 1815, and the composition 1812 is applied to the adherand at a temperature between 200 and 325 degrees Fahrenheit, preferably 250 degrees Fahrenheit.

Composition 1812 may also be formulated as a liquid dispersion. The composition 1812 can then be applied to the adherand at room temperature. Once the liquid dispersion composition 1812 is applied to the adherand, the coated material 1810 is passed through a heating system 1820. Heating system 1820 acts to dry the composition 1812. Heating system 1820 can be any conventional heating system, like an oven, or heating system 1820 can be an RF heating system, such as heating system 500 described above. Other drying means that may be employed include, for example, a heat lamp with or without a fan to remove volatiles, or microwave heating system.

Composition 1812 can be applied in powder form by conventional electrostatic gun/spray.

In one embodiment, the coated adherand 1810 is rolled onto a roller 1830 after composition 1812 is sufficiently dried. Alternatively, the coated adherand 1810 can be cut into pieces and stacked. The coated susceptor 1810 can be used at a later point in time in the bonding process described above. The bonding process can occur anytime within a few seconds up to many months after the adherand 1810 has been coated with composition 1812.

#### ***X. Systems for Adhering or Bonding Two Adherands.***

FIG. 19 illustrates one embodiment of a system for bonding or adhering various adherands or layers. The system utilizes RF heating system 400, including power supply 402, cable 404, heat station 406, and coil 408, and clamp 1902. The adherands to be bonded by RF heating 400, shown as layers 1910, pass through or in proximity to coil 408. Layers 1910 can either be coated with a suitable susceptor composition, can sandwich a film made from a susceptor composition

or can sandwich an insoluble, porous carrier (such as a thermoplastic carrier web) that is saturated with a susceptor composition as described above. A clamp 1902 provides uniform pressure to the adherands to be bonded or adhered. Alternatively, coil 408 can be implemented to provide a uniform pressure to the adherands to be bonded or adhered. Precision bonding or adhering takes place as the layers 1910 are exposed to the electromagnetic field generated when an alternating current flows through coil 408. The electromagnetic field has sufficient RF energy to activate the bonding composition. Preferably, layers 1910 are exposed to the electromagnetic field for at least 100 milliseconds to several seconds or minutes. In the case of thermoset compositions, in general, longer times are needed, e.g. from 1 second to several minutes or hours.

FIGS. 20A and 20B illustrates a static bonding system 2000 for bonding or adhering adherands 2090 and 2092 (see FIG. 20B). Bonding system 2000 is referred to as a static because the adherands to be bonded do not substantially move while they are being exposed to the electromagnetic field that activates an RF activatable composition which is located between the adherands.

Referring now to FIG. 20A, bonding system 2000 includes a power supply, such as voltage supply 502, for generating an alternating voltage between output terminal 612 and output terminal 610. Connected to output terminal 612 is a probe 2006, and connected to output terminal 610 is a probe 2008. The characteristics of probe 2006 and probe 2008 are described above with reference to probes 602 and 604. In one embodiment, probe 2006 and 2008 are rectangular hollow tubes made from a conductive material, preferably copper. Preferably, the height (H) and width (W) of each probe is about equal, and the length (L) is generally larger than the height and width. For example, in one embodiment, the height and width of each probe is about 1/8 of an inch, whereas the length of each probe is about 10 inches. In general, the height and width of a rectangular probe, or the diameter for a cylindrical probe, ranges between about 0.02 and 0.5 inches, and the length generally ranges from about 0.25 inches to 20 feet.

System 2000 is not limited to two probes. A third probe (not shown) could be placed adjacent to probe 2006 such that probe 2006 will then be between

the new probe and probe 2008. With this configuration, the new probe would be connected to the output terminal that probe 2008 is connected to, which in this case is terminal 610. An exemplary three probed system is illustrated in FIG. 13A. One skilled in the art should recognize that any number of probes could be used, provided that no two adjacent probes are connected to the same output terminal of voltage supply 502.

In one embodiment, probes 2006 and 2008 are placed in an electrically insulating block 2010. Insulating block 2010 is composed of an electrically insulating material, such as, but not limited to polytetrafluoroethylene (TEFLON™). An optional electrically insulating layer 2012 (see FIG. 20B) may be placed on top of probes 2006 and 2008. Preferably, electrically insulating layer is made from polytetrafluoroethylene or other like material which resists adhesion of the substrates or adherands thereto.

An alternative electrically insulating block 2050 is illustrated in FIG. 20C. FIG. 20C shows a cross-sectional view of probes 2006 and 2008 housed within the insulating block 2050. Insulating block 2050 is formed from two elements, insulating element 2052 and insulating element 2054.

Insulating element 2052 has two U shaped recesses 2056 and 2058 for receiving probes 2006 and 2008, respectively. In one embodiment, a low dielectric encapsulate 2060 is placed with the probes in the recesses. Insulating element 2054 has two protrusions 2062 and 2064 for mating with the recesses 2056 and 2058 of insulating element 2052. Preferably, both insulating element 2052 and insulating element 2054 consist primarily of polytetrafluoroethylene (TEFLON™).

Referring now to FIG. 20B, to bond adherand 2090 to adherand 2092, adherand 2090 and/or adherand 2092 is coated with a suitable composition 2091, or a film of the composition 2091 is sandwiched between adherand 2090 and adherand 2092, or an insoluble porous carrier is saturated with composition 2091 and placed between adherand 2090 and adherand 2092. Adherands 2090 and 2092 are then placed over probes 2006 and 2008 such that composition 2091 is between the adherands and over the region between probe 2006 and probe 2008,

as shown. Power supply 502 is then activated, which creates an alternating voltage between terminals 612 and 610, which creates an electromagnetic field between probes 2006 and 2008. The composition 2091 is exposed to the electromagnetic field for a predetermined amount of time. The predetermined amount of time can range between about 100 milliseconds to about one second, several minutes, or hours depending on the composition and/or the strength of the electromagnetic field. The electromagnetic field causes composition 2091 to heat. When composition 2091 reaches a given temperature, the composition will begin to melt and flow, causing an impedance change on the matching circuit 812. The impedance change can be detected by a change in reflected power signal 832. This change in reflected power signal 832 can be used to control the intensity of the RF energy. Other methods of detecting when composition 2091 melts is to detect displacement of a pressure plate 2020 with a feed back loop. After the predetermined amount of time has expired or while the composition is exposed to the electromagnetic field, the adherand can be pressed together using pressure plate 2020, pressure roller (not shown), or any other pressure delivery apparatus or means, thereby assuring a good bond.

The resulting bond can be an adhesive bond, mechanical bond, thermal bond, or any combination of aforementioned bonds. For example, composition 2091 may have adhesive properties to create an adhesive bond between adherands 2090 and 2092, and/or composition 2091 may be used as a source of thermal energy for welding the adherands together.

An advantage of the present invention is that non-electrically conductive materials can be stacked on top of an adherand without affecting the bonding process. Only composition 2091 is directly heated when the layers are exposed to RF energy having the preferred frequency range of 10 to 15 MHz. Thus, by selectively heating only the composition 2091, multiple layers may be assembled prior to forming the bond between adherands 2090 and 2092. This allows the assembly of complex laminates prior to bonding.

Another advantage of the present invention is that RF energy can be re-applied to the bonded product and the two (or more) adherands 2090 and 2092

can be disassembled. This is known as de-activating the composition 2091. In fact, the composition 2091 can be activated and de-activated a number of times.

FIGS. 38 and 39 illustrate two exemplary manufacturing systems in which static bonding system 2000 could be utilized. FIG. 38 illustrates a step and repeat manufacturing system. There are many applications in general manufacturing where adherands are joined or bonded together using an adhesive. In a conventional step and repeat joining (or bonding) system there is a gluing station immediately followed by a joining station. The gluing station applies an adhesive to an adherand. After the adhesive is applied, the adherand moves immediately to a joining station where it is brought together with the other adherand to which it is to be joined. The joining station then nips the adherands together to form a bond.

The adhesive compositions according to the present invention allow the adhesive to be applied to the adherand(s) prior to the adherand(s) entering the manufacturing line. For example, the adhesive compositions according to the present invention may be applied at the part supplier's facility with on-demand bonding occurring for, example, days, weeks, or months later, by RF activation.

Referring now to FIG. 38, a step and repeat manufacturing process as applied to a continuous production line 3802 with base adherand 3806 and top adherand 3808 being supplied to bonding system 2000 on a conveyor system 3804. In one embodiment, base adherand 3806 is pre-coated with an adhesive composition 3805 according to the present invention. Base adherand 3806 could have been coated minutes, days, weeks, or months prior to base adherand 3806 entering continuous production line 3802. Base adherand 3806 travels along the conveyor 3804 and top adherand 3808 is assembled to base adherand 3806 by hand or automatic system (not shown). The assembled adherands 3810 are placed onto a pressure plate 2010 in which probes 2006 and 2008 are embedded. The bonding process begins when an electromagnetic field is created between probes 2006 and 2008 by power supply 502. The electromagnetic field activates the adhesive composition 3805, which then creates a bond between adherands 3806 and 3808. Pressure plate 2020 is used to nip the bond during and/or after RF

activation. After the bond is nipped, the assembly 3810 is removed from bonding system 2000 and placed back on the conveyor 3804.

FIG. 39 illustrates an index table bonding system. Index table bonding systems are used in many manufacturing industries to automate the bonding process. Examples include the bonding of labels onto bottles. The index table process allows for setting up multiple stations where different processes in the assembly process are performed. The time the index table stops at each station is the same, thus it is dependent upon the slowest process. An advantage of using an adhesive composition according to the present invention includes the pre-application to one or both of the parts to be bonded prior to loading the parts onto the index table. Other advantages are fast activation and curing time. Consequently, by removing the adhesive application from the index table, one less station is used and a higher production throughput is achieved.

Referring now to FIG. 39, an index table bonding system is described. The index table bonding system includes an index table 3902, which is generally round and rotates either clockwise or counter-clockwise. Base parts 3904(1)-(N) having a pre-applied adhesive composition 3906 are placed onto index table 3902. When index table 3902 moves base part 3904(1) to the next station (station 2), a top part 3908 is placed onto base part 3904 to form assembly 3910. Assembly 3910 then moves to station 3 where it is exposed to an RF field, which activates adhesive composition 3906. In station 3, the RF field is generated by probes (not shown) positioned so that adhesive composition 3906 is activated. The probes may be configured to either contact the assembly 3910 and apply some pressure to aid in the bonding process. Alternatively, the probes could be configured so there is no contact with the assembly 3910. After activation of the adhesive 3906, the assembly 3910 moves to station 4 for a nip or cure portion of the bonding process. After station 4, the assembly 3909 moves to station 5 where it is unloaded from the index table 3902.

FIG. 21 illustrates a dynamic bonding system 2100 (also referred to as an in-line bonding system) for bonding or adhering adherands. Bonding system 2100 is referred to as dynamic because the adherands to be adhered, adherands 2110

and 2112, continuously move through an electromagnetic field; which is generated by heating system 2140. In one embodiment, adherand 2110 is pre-coated with a composition 2104 according to the system shown in FIG. 18.

5        Bonding system 2100 includes a roll 2102 of coated adherand 2110 and plurality of rollers 2120, 2122, 2124, 2126, and 2128 for, among other things, continuously guiding coated adherand 2110 and adherand 2112 through an electromagnetic field generated by heating system 2140. In one embodiment, coated adherand 2110 and adherand 2112 move through the electromagnetic field at a rate of about 0.01 to 2000 feet per minute, most preferably, about 1000 feet  
10        per minute (ft/minute).

      The bonding process begins when coated adherand 2110 is fed onto roller 2120. Coated adherand 2110 is then passed over roller 2122. A pressure activated construction bond may be formed by passing the two adherands 2110 and 2112 between roller 2122 and nip roller 2124. A construction bond may be  
15        required in this process to maintain the proper location of coated adherand 2110 and adherand 2112 prior to and/or during activation. Preferably, the composition 2104 is formulated to provide a pressure sensitive tack when a construction bond is needed. Coated adherand 2110 and adherand 2112 are not limited to any particular thickness. As should be readily apparent to one skilled in the art, the  
20        system can be designed to accommodate any reasonable thickness of adherand.

      In this embodiment, the invention relates to a method for dynamically bonding a first adherand to a second adherand, comprising:

      (1) creating an article of manufacture comprising the first adherand, the second adherand, and a composition, the composition being placed between the  
25        first adherand and the second adherand, wherein the composition can be activated in the presence of an RF field;

      (2) moving the article of manufacture along a predetermined path;

      (3) generating along a portion of the predetermined path an RF field having sufficient energy to activate the composition, wherein the composition is  
30        activated by its less than one second exposure to the RF field.

In a preferred embodiment, the article passes through the RF field at a rate of at least about one-thousand feet per minute. In a more preferred embodiment, the article passes through the RF field at a rate of about 1000 feet per minute.

Referring now to FIG. 22, after the construction bond is formed, the construction bonded coated adherand 2110 and adherand 2112 are passed through an RF field 2230, which is generated by heating system 2140. FIG. 22 further illustrates heating system 2140 for use in dynamic bonding system 2100.

Heating system 2140 includes a power supply, such as power supply 502, for generating an alternating voltage between terminal 612 and terminal 610. Connected to terminal 612 is a probe 2210, and connected to terminal 610 is a probe 2220. The characteristics of probes 2210 and 2220 are described above with reference to probes 602 and 604 and probes 2006 and 2008. In one embodiment, probe 2210 has a distal section 2211, a center section 2212 and a proximal section 2213. Similarly, in one embodiment probe 2220 has a distal section 2221, a center section 2222 and a proximal section 2223. Preferably, center section 2212 is parallel with center section 2222, and they both have a length of about 48 inches when the adherands 2110 and 2112 are traveling at about 1000 feet/minute in the direction indicated by arrow 2130. This configuration results in about a preferred 240 millisecond dwell time. Dwell time refers to the maximum amount of time that any given point on adherands 2110 and 2112 is positioned beneath (or over) probes 2210 and 2220 (i.e., within the activation region). If the speed of the adherands 2110 and 2112 is increased, the preferred dwell time can remain constant by increasing the length of probes 2210 and 2212. For example, if it is desired for the adherands 2110 and 2112 to move at a rate of about 2000 feet/min over probes 2210 and 2220, and the preferred dwell time is about 100 milliseconds, then the minimum length of probes 2210 and 2220 would be about 40 inches. Although a preferred dwell time is 600 milliseconds, the dwell time can be increased to several minutes if desired by increasing the length of probes 2210 and 2220, e.g., from about the 20 inches to 20 feet, and/or decreasing the speed at which adherands 2112 and 2110 travel



over probes 2210 and 2220. Shorter probes are also contemplated, for example from about 0.25 inches to about 20 inches.

Preferably, probes 2210 and 2220 are positioned with respect to coated adherand 2110 such that the composition that coats coated adherand 2110 is  
5 beneath (or above) an activation region. The activation region is the area between the center section 2212 and center section 2222.

The frequency of the alternating voltage generated by power supply 502 can range from the low Kilohertz to high Gigahertz range. In one embodiment the frequency ranges between about 1 MHz to about 5 GHz, most preferably about  
10 10 MHz and 15 MHz. The peak to peak level of the voltage generated by power supply 502 may range from about 500 V to 20 kV, most preferably about 1 to 15 kV. The composition 2104 will remain activated as long as the RF energy is delivered.

After the adherands 2110 and 2112 pass over (or under) probes 2210 and  
15 2220 they are nipped by non-destructive nip rollers 2126 and 2128, which assure that a good bond is created between adherand 2110 and adherand 2112. For optimal performance, the nip rollers 2126 and 2128 apply pressure immediately after re-flow temperatures are achieved within the adhesive material. Additionally, nip roller 2126 and/or nip roller 2128 may be cooled to remove thermal energy  
20 from the adherands. Upon cooling, the composition forms a strong bond between the adherands 2110 and 2112. The bonded adherands can then be subsequently processed in accordance with a particular application.

There are a number of benefits of the above system. First, the system provides a finished bond in less than about one second of activation. Second, the  
25 activation process does not produce harmful emissions or by-products that may interfere with the bonding of two thin films. Third, the activation only occurs in the activation region.

FIGS. 23-27 illustrate alternative designs for heating system 2140. As shown in FIG. 23, curved probes 2310 and 2320 can be used in place of straight  
30 probes 2210 and 2220. An advantage of curved probes 2310 and 2320 is that the width 2390 of the activation region is greater than the distance 2311 between

probes 2310 and 2320, whereas the width of the activation region provided by probes 2210 and 2220 equals the distance between center section 2212 of probe 2210 and center section 2222 of probe 2220.

5 The heating system shown in FIG. 24 includes probe 2410 in addition to probes 2210 and 2220. Probe 2410 is positioned between probes 2210 and 2220. Probe 2410 is parallel with probes 2210 and 2220. Preferably, the distance (d) between probe 2410 and 2210 is equal to the distance (d) between probe 2410 and probe 2220. Probes 2210 and 2220 are both connected to the same output terminal of voltage supply 502, whereas probe 2410 is connected to the other  
10 output terminal. An advantage of the probe design illustrated in FIG. 24, is that it provides a larger activation region. The width 2420 of the activation region is greater than the distance (d) between any two of the probes. Based on the above description, one skilled in the art will recognize that any number of probes can be used in heating system 2140, provided that no two adjacent probes are connected  
15 to the same output terminal of voltage supply 502.

The heating system shown in FIG. 25 is similar in concept to the one shown in FIG. 24. A curved probe 2510 is placed between curved probes 2310 and 2320. Curved probes 2310 and 2320 are both connected to the same output terminal of voltage supply 502, whereas probe 2510 is connected to the other  
20 output terminal. Again, an advantage of the heating system shown in FIG. 25 is that it can provide a larger activation region than the similar heating system shown in FIG. 23.

FIG. 26 illustrates another heating system. The heating system shown in FIG. 26 includes two plates 2610 and 2620. Plate 2610 is positioned above  
25 adherand 2110 and plate 2620 is positioned below adherand 2112. Thus, composition 2104 travels between plates 2610 and 2620. Plate 2610 is connected to output terminal 610 of voltage supply 502, and plate 2620 is connected to output terminal 612 of voltage supply 502. When voltage supply 502 is turned on, it generates an electromagnetic field between plates 2610 and 2620, which is used  
30 to activate composition 2104. FIG. 27 illustrates another perspective of plates 2610 and 2620. As is apparent from FIG. 27, the width of the activation region

for this design is simply the width (W) of the plates. The center to center distance (d) between plate 2610 and plate 2620 can range from 0.02 inches to 20 inches. In one embodiment, the distance ranges between 0.25 inches and 1.5 inches. The length (L) of course depends on the desired dwell time and the rate at which any given point on adherand 2110 or 2112 travels between any two points along the length of one of the plates.

## ***XI. Exemplary Specific Applications of the Present Invention***

The susceptor compositions may be employed for many purposes including bonding, cutting, and coating. Thus, the susceptor compositions may be employed for packaging applications, e.g. to bond or adhere cases or cartons as described in U.S. 5,018,337, but with the additional step of RF activation. Applications for the RF cured thermoset compositions, which are illustrative only and not to be considered limiting of the scope of the present invention, include:

Coatings for conventional and spray applications on plastics, metals, wood etc.

Corrosion resistance coatings.

Industrial and protective coatings.

Top coats.

Automotive coatings.

Lamination of composites.

Laminating adhesives.

Bonding of structural composites.

Inks and decorative coatings.

Barrier coatings.

Additional applications are listed below, but are likewise illustrative and not limiting of the scope of the present invention.

**A.     *Manufacture of Flexible Packaging***

FIGS. 28A and 28B illustrate one embodiment of a system for the manufacture of flexible packaging. Flexible packages are used for, among other things, packaging foods. The system includes a system 2802 (see FIG. 28A) for manufacturing an RF activated adhesive film 2815 and a bonding system 2804 (see FIG. 28B) for bonding the adhesive film 2815 to another film 2850.

Referring now to FIG. 28A, film manufacturing system 2802 includes an extruding system 2810, a casting wheel 2814 a heating system 2820, a stretching system 2830, and an optional film roller 2840. In one embodiment, extruding system 2810 includes three extruders 2811, 2812, and 2813. An RF activated adhesive composition according to the present invention is first formulated into an extrudable resin (for example, ethylene vinyl acetate or other polymer based material is added to the adhesive composition) and then provided to extruder 2813 in a pellet or liquid form. Polypropylene or other like similar substance, such as but not limited to ethylene vinyl acetate (EVA), is provided to extruder 2811, and a sealing material is provided to extruder 2812. The output of extruders 2811-2813 are cast into a film 2815 by casting wheel 2814.

FIG. 29 illustrates film 2815. As shown in FIG. 29, film 2815 includes a first layer 2902 consisting of the sealing material, a second layer 2904, e.g., OPP and/or EVA and/or other similar substance, and a third layer 2906 consisting of the RF activated adhesive. Because film 2815 includes an adhesive composition according to the present invention, film 2815 can be RF activated.

Referring back to FIG. 28A, film 2815 is provided to heating system 2820. In one embodiment, heating system 2820 includes heater rollers 2821 and 2822. The function of heating system is to heat the film to a temperature that allows the film to be stretched. After being processed by heating system 2820, film 2815 is stretched by stretching system 2830. In one embodiment, stretching system 2830 includes a plurality of stretch rollers 2831, 2832, 2833, 2834, and 2835 and a transverse stretcher 2837. Stretching system 2830 stretches film 2815 both length

and width wise. After being stretched, film 2815 may be rolled up using film roller 2840. Alternatively, film 2815 can be cut and stacked after being stretched.

Referring now to FIG. 28B, bonding system 2804 is used to bond film 2815 with film 2850. In one embodiment, film 2850 is a 70 gauge oriented polypropylene (OPP) film. Film 2850 is passed over a print wheel 2855 and then through oven 2857. A pair of nip rollers 2860 and 2861 press film 2815 with film 2850 to form a construction bond and thus form a single multi-layer film 2870. FIG. 30 illustrates one embodiment of film 2870.

As shown in FIG. 30, film 2870 includes layer 2902 consisting of the sealing material, layer 2904 that includes thermoplastics and/or elastomers, for example, OPP and/or EVA and/or other similar substance, third layer 2906 consisting of the RF activated adhesive, a fourth layer 3002 consisting of the ink applied by print wheel 2855, and a fifth layer 3004 consisting of film 2850.

Referring back to FIG. 28B, an RF heating system 2875 creates an RF field that is used to heat adhesive layer 2906. Heating system 2875 defines an activation region. The activation region is an area in which the RF field generated by heating system 2875 is strong enough to activate adhesive layer 2906. Film 2870 can travel through the activation region in as quickly as about 100 milliseconds. Shortly after passing through the activation region, film 2870 is nipped by nip rollers 2880 and 2881 and then rolled by film roller 2885. FIGS. 16A and 16B illustrate one embodiment of the probe portion of heating system 2875. Other heating systems could be used, such as those described above with respect to FIGS. 20 and 21.

FIG. 31 illustrates an alternative system 3100 for manufacturing an RF activated adhesive film for use in the flexible packaging industry. System 3100 is similar to system 2802, except that system 3100 does not include extruder 2813. In place of extruder 2813, system 3100 includes an adhesive applicator 3101 and a heating system 3102. An adhesive composition according to the present invention is formulated into a liquid dispersion and applied to film 2815 by adhesive applicator 3101. In one embodiment adhesive applicator 3101 includes a gravure application tool (not shown). Heating system 3102 can be a

conventional heating system, such as an oven, or it can be an RF heating system, such as heating system 600 or any of the other heating systems described herein.

### ***B. Food Packaging and Cap Sealing***

Conventionally, metallic foils are used as susceptors of electromagnetic energy to generate heat for package sealing. Typical examples include tamper evident bottle seals (i.e., cap sealing) and food packaging. While the conventional systems are effective in sealing the packages, the use of metallic foils eliminates the manufacturer's ability to perform post sealing inspection, such as metal detection, x-ray, and the like. Additionally, there may be a recycling benefit and a cost saving to the system by eliminating the metallic foil.

One solution is to replace the metallic foil with a composition of the present invention. The composition may or may not have adhesive properties. FIG. 32 illustrates a conventional aseptic package construction. A conventional aseptic package includes an outer polyethylene layer 3202, a paper layer 3204, a second polyethylene layer 3206, a layer of metallic foil 3208, a third polyethylene layer, an inner polyethylene layer 3212, and a container 3214 that holds the food or beverage. Inner polyethylene layer 3212 is the layer that contacts with the container 3214, and is used to seal the container during the food packaging process. The sealing is achieved through induction heating of the metallic foil layer 3208 causing the inner polypropylene layer 3212 to melt and bond to the container 3214.

FIG. 33 illustrates one embodiment of a packaging construction that does not use metallic foils. The packaging construction includes the outer polyethylene layer 3202, the paper layer 3204, the second polyethylene layer 3206, a susceptor composition according to the present invention 3302, a barrier layer 3310, an inner layer 3212, and a container 3214 that holds the food or beverage. Inner layer 3212 is the layer that contacts with the container 3214, and is used to seal the container 3214 during the food packaging process. Inner layer 3212 can be a polyethylene or EVA layer. In one embodiment, barrier layer 3310 is an EVOH

barrier layer. The sealing is achieved through RF heating of susceptor composition 3302, which causes the inner layer 3212 to melt and bond to the container 3214. The advantage of replacing metallic foil 3208 with susceptor composition 3302 is that now the container 3214 can be inspected after it is sealed by using a metal detector or x-ray machine, and there are recycling advantages as well.

A conventional cap sealing construction is illustrated in FIG. 34. FIG. 34 illustrates a polyethylene bottle 3402, a seal 3401, and a bottle cap 3414. Seal 3401 includes several layers of substrate, including a polyethylene layer 3404, a metallic foil layer 3406, another polyethylene layer 3408, a wax layer 3410, and a paper layer 3412. Seal 3401 is adhered to bottle 3402 by heating foil through induction, which causes layer 3404 to weld to bottle 3402. As discussed above, it is desirable to remove metallic foil layer 3406.

FIG. 35 illustrates an improved seal 3501 for bottle 3402. Seal 3501 is identical to seal 3401 (see FIG. 34), except that the metallic foil 3406 has been replaced with a composition 3502 according to the present invention. As discussed above, the advantage of removing metallic foil 3406 is that now bottle 3402 can be inspected after it is sealed by using a metal detector or x-ray machine, and can be more easily recycled.

Another use of the compositions described herein is to attach a flexible bag 3602 containing dry food to an outer box 3604, as illustrated in FIG. 36. In one embodiment, flexible bag 3602 includes three layers, 3610, 3611, and 3612, and outer box 3604 is a paper product, such as a paper board. To bond flexible bag 3602 to outer box 3604, an adhesive composition 3620 according to the present invention is placed between outer box 3604 and layer 3610. Adhesive composition 3620 is then exposed to an RF field that causes the composition 3620 to melt and flow and bond layer 3610 to outer box 3604. In one embodiment, layer 3610 is a polyethylene layer, layer 3611 is an EVOH barrier layer, and layer 3612 is an EVA food contact layer. In another embodiment (see FIG. 37), outer box 3604 is coated with a polyethylene layer (or other like layer) 3730. This configuration creates an improved bond.

**C.     *Printing Applications***

The susceptor compositions of the present invention may also be applied together with one or more inks to provide writing, a design or graphic, e.g. as is described in U.S. Patent No. 4,595,611. Particular application of this aspect of the invention is in the preparation of ink-printed substrates such as ovenable food containers. Examples of pigments that can be combined with the susceptor composition include titanium dioxide, iron oxide pigments, carbon black and organic pigments such as isoindoline yellow. In a preferred embodiment, the susceptor is a sulfonated polyester. Alternatively, a sulfonated polyester-cationic dye salt may be employed as disclosed in U.S. Patent No. 5,240,780. The substrate may be printed once or multiple times to achieve the desired result. Once printed, the substrate may be further coated with a clear unpigmented composition which may comprise the susceptor composition of the invention. The same composition used to print may be used to further coat, but without the added pigments. The susceptor compositions may be RF activated after each printing/coating step, or after all of the coatings are applied. Finally, the substrate may be coated with a clear polyester sealing resin.

An extension the printing application is high speed ink-jet used in printers/copiers. Inks formulated as liquids (H-P/Cannon) or solid (Tetronic) composition can contain the susceptor compositions of this invention in amounts effective that can be activated by RF energy for rapid drying and fixing. Current ink formulations are too "slow in drying" or need excessive heat energy.

**D.     *Bookbinding and Mailers***

The susceptor compositions of the present invention may be used to bond paper substrates used in printing and/or copying. An advantage of the present invention is that a substrate to be printed on (such as a paper substrate) can be coated with a susceptor composition described herein prior to printing on the substrate. For example, FIG. 43 illustrates a process for assembling a book,



magazine, or periodical, or the like. In step 4302, a portion of one side of a substrate is coated with a susceptor composition that functions as an adhesive. Any one of the various methods for coating a substrate described herein can be used to coat the substrate. FIG. 44 illustrates a preferred portion of a substrate to be coated with the susceptor composition. As shown in FIG. 44, a thin strip of the susceptor composition 4404 coats one edge of the substrate 4402. The portion of the substrate that is not coated is the portion where ink will be printed. Preferably, the susceptor composition 4404 is formulated such that it is tack free, however, this is not a requirement.

After the substrate 4402 has been coated, the substrate may be processed into rolls, stacks and the like and stored for later use (step 4304). In step 4306, the coated substrate is fed into a printing means that prints ink onto the substrate. The printing means can be a conventional printer or conventional photocopying machine. Further, the substrate can be fed into the printing means as a continuous substrate or as cut pieces. For this example, we will assume that cut pieces of the substrate are fed into the printing means. In step 4308, after the printing means prints ink onto a substrate, the substrate is stacked with the other substrates that have already been fed into the printing means as shown in FIG. 45. The stack is placed in an electromagnetic field. The electromagnetic field causes the susceptor composition to melt and flow. The stack is then nipped to assure a good bond (step 4312).

In one embodiment, prior to placing the stack in the electromagnetic field, the substrate stack is pressure bonded by applying upward and/or downward pressure on the stack. In another embodiment, the ink that is printed on the substrates includes a susceptor composition. In this way, the ink can be dried rapidly by passing the substrate through an electromagnetic field.

In another embodiment, mailers or envelopes can be constructed. Referring to FIG. 46, a portion of one side of substrate 4602 is coated with a susceptor adhesive composition 4604. Preferably, the susceptor adhesive composition 4604 is formulated so that it is tack-free. The substrate 4602 includes a fold line 4610. The coated substrate 4602 can be fed into a printing

means that prints ink onto the substrate. After the ink is printed thereon, the substrate is folded along the fold line 4610 so that the top portion 4612 of the substrate 4602 contacts the bottom portion 4614 of the substrate (see FIG. 47). At this point, the substrate is passed through the electromagnetic field so as to melt and flow the susceptor composition 4604, thereby bonding the top portion 4612 of the substrate with the bottom portion 4614 when the susceptor composition 4604 solidifies.

### *E. Security Devices*

As would be apparent to one skilled in the relevant art(s), the adhesive of the present invention can be used to seal containers, casings, housings and the like (hereafter "container"). In particular, the adhesive of the present invention is preferably used to seal containers that a manufacturer does not want accessed by others. A manufacturer may want to prevent a third party from opening certain containers for security, safety or quality control reasons. However, the inside of the container must still be accessible to the manufacturer or qualified repair facility. By exposing the seal to an electromagnetic field, the manufacturer can disassemble the container.

For example, a manufacturer may want to prevent an article intended for one-time use from being reused. As such, the adhesive of the present invention can be used, for example, to seal the shell or casing of a disposable camera. The manufacturers of such disposable cameras often do not want to have the shells reloaded and reused by the consumer or a competitor company. If the adhesive of the present invention is used to seal the camera shell, then when the film developer opens the camera body to remove and process the film, mating sections of the camera shell attached by the adhesive would break or deform such that the camera body could not be reused. As such, the adhesive of the present invention would prevent tampering with and unauthorized reloading of disposable camera shells.

FIG. 48 shows an example of a container 4800 sealed with a susceptor composition of the present invention. Container 4800 includes a first portion 4804 and a second portion 4808. In one embodiment, first portion 4804 is a container base and second portion 4808 is a lid. Container 4800 can be made from a variety of materials, including, for example, polypropylene, polystyrene, polyolefin, wood or wood products, rubber, plastics, glass, ceramics, paper, cardboard, natural or synthetic textile products, aluminum or other foils, metals, or any combination of these materials. An adhesive composition 4812, made in accordance with the present invention, is applied to a surface of container 4800. In the example of FIG. 48, adhesive composition 4812 is applied to a first mating surface of first portion 4804. Second portion 4808 is then placed on top of first portion 4804, so that a second mating surface of second portion 4808 comes in contact with adhesive composition 4812. A suitable electromagnetic field, as described herein, is then applied to adhesive composition 4812 to join the first and second mating surfaces of first and second portions 4804 and 4808.

To open container 4800, suitable RF energy must again be applied to container 4800 to cause adhesive composition 4812 to reflow. If a person attempts to open container 4800 without applying the suitable electromagnetic field, the container 4800 is designed to preferably break or catastrophically fail and so that it cannot be reused.

FIG. 49 shows another example of a device 4900 sealed or otherwise joined together with a composition of the present invention. Device 4900 includes a first portion or substrate 4904 and a second portion or substrate 4908. Device 4900 can be made of a variety of materials, as discussed above with respect to container 4800, shown in FIG. 48. In this embodiment, first substrate 4904 includes a male portion 4912 forming the first mating surface. Male portion 4912 includes a narrowed section 4916 and a wider section 4920. A corresponding female portion 4924 forming a second mating surface is formed in second portion 4908 and is configured to accommodate or receive wider section 4920 of male portion 4912. Second portion 4908 may also be configured to accommodate a portion of narrowed section 4916.

An adhesive composition 4928, made in accordance with the present invention, is applied to the second mating surface of female portion 4924 of second portion 4908. First portion 4904 is then assembled so that the first mating surface comes in contact with adhesive composition 4928 on second portion 4908 while the adhesive composition is within the electromagnetic field. First portion 4904 is locked into second portion 4908 once the application of electromagnetic field is discontinued, causing adhesive composition 4928 to solidify. To disassemble device 4900, an electromagnetic field must again be applied to adhesive 4928 to cause it to reflow and allow the portions 4904 and 4908 to separate. If someone attempts to disassemble device 4900 without application of a suitable electromagnetic field, narrowed section 4916 of male portion 4912 will break or otherwise catastrophically fail resulting in device 4900 being unusable. As such, this embodiment will prevent authorized disassembly and reuse of device 4900.

FIG. 50 shows another example of a device 5000 sealed or otherwise joined together with a composition of the present invention. Device 5000 is similar to device 4900 described above with respect to FIG. 49, except that an electronic circuit path 5004 is added to male portion 4912 such that it is disposed through narrowed section 4916. As such, should portions 4904 and 4908 of device 5000 be disassembled without application of a suitable electromagnetic field, electronic circuit path 5004 will be cut during failure of narrowed section 4916, resulting in further failure of device 5000.

FIG. 51 shows still another example of a cross-section of a container 5100 that has been sealed with the adhesive of the present invention. Container 5100 includes a first portion 5104 and a second portion 5108. Container 5100 can be made of a variety of materials, as discussed above with respect to container 4800, shown in FIG. 48. First portion 5104 includes a protrusion 5112 which forms a first mating surface. In the embodiment shown in FIG. 51, protrusion 5112 extends around the entire circumference of container 5100. However, it would be apparent to one skilled in the relevant art that one or more discrete protrusions 5112 could be used instead of or in addition to the continuous protrusion 5112.

Second portion 5108 includes a recess 5116 which forms a second mating surface corresponding to the first mating surface of protrusion 5112. Protrusion 5112 and corresponding recess 5116 are formed slightly inward of the periphery of container 5100 to so that when first and second portions 5104 and 5108 are joined, the mating surfaces and an adhesive composition 5120 therebetween cannot be accessed, thereby further reducing the risk of a person prying apart or otherwise disassembling container 5100. Adhesive composition 5120, made in accordance with the present invention, is applied to the second mating surface of recess 5116. First and second portions 5104 and 5108 can be joined together by application of suitable electromagnetic field and similarly disassembled by re-application of the electromagnetic field.

The invention relates to an apparatus, comprising:

a first portion having a first mating surface;

a second portion, having a second mating surface;

a composition disposed between the first mating surface and the second mating surface, wherein the composition comprises a susceptor and a polar carrier wherein the susceptor and/or the polar carrier are present in amounts effective to allow the composition to be heated by RF energy, and wherein the composition adheres the first mating surface to the second mating surface such that application of a force to separate the first mating surface and the second mating surface results in breakage of the apparatus unless the composition is in a melted state.

In this apparatus, the composition may be disposed on the first mating surface and the second mating surface such that the composition is not accessible when the first and second mating surfaces are joined. In another embodiment, the first mating surface may comprise a protrusion disposed on the first portion. In another embodiment, the second mating surface may comprise a recess formed in the second portion. In a further embodiment, the apparatus may further comprise an electronic circuit path disposed in the protrusion. In another embodiment, the first portion and the second portion are disassembled upon application of an electromagnetic energy to the composition.

***F. Thermal Destruction***

The susceptor composition of the present invention can not only be used to coat a substrate and bond adherands, but also can be used to cut a substrate. A substrate can be cut using a susceptor composition described above by first  
5 applying the susceptor composition to at least one side of the substrate. Next, an electromagnetic field is applied to the susceptor composition causing the susceptor composition to heat. The thermal energy generated by the susceptor composition heats the substrate, particularly the section of the substrate that is in contact with the susceptor composition. The substrate is heated until a section of the substrate  
10 melts resulting in the substrate being cut.

In this embodiment, the invention relates to a method for cutting a substrate, comprising:

applying a composition to a portion of the substrate, wherein the composition comprises a susceptor and polar carrier wherein the susceptor and/or  
15 the polar carrier are present in amounts effective to allow the composition to be heated by RF energy, and wherein the portion of the substrate defines a first section of the substrate and a second section of the substrate;

melting the portion of the substrate, wherein the melting step includes the step of heating the composition, wherein the step of heating the composition  
20 includes the step of applying RF energy to the composition;

after the portion of the substrate has begun to melt, applying a force to the substrate to separate the first section from the second section.

***G. Seam Sealing***

The susceptor compositions of the present invention may be used to seal  
25 the seams of products made from cloth. Conventional cloth materials manufactured from man made or natural fibers are sewn together to form cloth products, such as clothing, bags, tents, awnings, covers, and the like. Typically, the seams of cloth products such as tents, awnings, bags, etc. need to be sealed to

prevent leakage of liquids through the small holes in the products created by a sewing needle and thread during a stitching process. The susceptor compositions of the present invention can be used to seal these seams.

FIG. 62 illustrates how a susceptor composition of the present invention can be used to seal the seams of cloth products. FIG. 62 illustrates a seam sealing system 6200 for sewing a first cloth material 6202 to a second cloth material 6204 and for sealing the seam. In one embodiment, a susceptor composition 6206 according to the present invention is placed between the first and second cloth materials 6202 and 6204. In another embodiment, either one or both of the cloth materials 6202 and 6206 are coated with the composition in the location where the seam will exist.

The system includes a pressure plate 6208 and a reciprocating needle 6212, through which a thread 6210 can be threaded, for joining the first cloth material 6202 with the second cloth material 6204. The seam sealing system 6200 also includes an RF heating system according to the present invention for activating the composition 6206. The RF heating system includes a reciprocating pressure foot 6214 and at least two probes (not shown) placed within and near the surface of the pressure plate 6208. The probes (not shown) are connected to the power supply 502 for generating an RF field at the probes. Alternatively, the probes can be located within the pressure foot 6214 as opposed to the pressure plate 6208.

The cloth materials 6202 and 6204 and the composition 6206 are pulled past the reciprocating needle 6212 and then past the reciprocating pressure foot 6214. The reciprocating needle 6212 and thread 6210 stitch the first material 6202 to the second material 6204, thereby joining the materials together at a seam. This stitching process creates small holes in the materials 6202 and 6204. The RF field generated at the probes within the pressure plate 6208 activates the composition 6206, which causes the composition 6206 to flow and thereby fill or cover the small holes created by the needle 6212 during the stitching process. The reciprocating pressure foot 6214 functions to evenly flow the activated composition 6206, thereby facilitating the composition in the filling/covering of

the holes created by the needle 6212. In this manner, the susceptor compositions of the present invention can be used to seal seams.

### ***XII. Kits***

The invention also provides kits for use in the preparation of the bonding composition according to the present invention. Kits according to the present invention comprise one or more containers, wherein a first container contains a susceptor composition of the invention. Additional kits of the invention comprise one or more containers wherein a first container contains a susceptor as described above and a second container contains one or more adhesive compounds and/or carriers, such as water, glycerine, N-methyl pyrrolidone (NMP), dimethylformamide (DMF), dimethylacetamide (DMAC), dimethylsulfoxide (DMSO), tetrahydrofuran (THF), polyvinyl pyrrolidone (PVP), polyvinylpyrrolidone/vinyl acetate copolymer (PVP/VA), and branched polyesters. The kits of the invention may be used to produce one or more of the bonding compositions of the present invention for use in a variety of applications as described below.

The invention also provides for kits comprising at least two containers, wherein one of the containers comprises a susceptor and another of the containers comprises a polar carrier, wherein when the susceptor and the carrier are applied to substrates and the applied susceptor and carrier are interfaced, a composition is formed that is heatable by RF energy.

### ***XIII. Experimental Set-up***

FIG. 40 shows an example experimental set-up utilized to test the susceptor compositions described above with respect to example 4. An RF signal is generated by a signal generator 4001. Signal generator 4001 can be an HP 8165A signal generator (available from Hewlett Packard Corporation). The RF signal is coupled to the input side of RF power amplifier 4002 (available from



ENI). The RF power is fed from the output side of RF power amplifier 4002 to the input side of an impedance matching circuit 4003 that functions to match the output impedance to the combined load impedance of coil 4004 and test sample 4005. Impedance matching circuit 4003 can be designed according to known electronics principles as would be apparent to those of skill in the art. See, e.g.,  
5 "The Art of Electronics," by P. Horowitz and W. Hill, Second Ed., Cambridge University Press (1994), especially Chapter 40, incorporated by reference herein. The RF power of load coil 4004 was inductively coupled to test sample 4005. The frequency of signal generator 4001 was tuned to result in resonance at load  
10 coil 4004. This frequency was detected by a single turn, 2 inch diameter probe loop 4007, which was located just below and in proximity to load coil 4004. Resonance was indicated by a maximum resulting voltage drop across probe loop 4007, and was displayed on an oscilloscope 4006, such as a model number OS7020A oscilloscope available from Goldstar. Frequency tuning was performed  
15 at sufficiently low RF powers in order to avoid heating of test sample 4005. Once the frequency of signal generator 4001 was tuned to resonance, the RF power delivered to load coil 4004 was increased to a desired power level by increasing the output level of signal generator 4001. The front panel of RF power amplifier 4002 displayed the measured RF power level delivered to test sample 4005.

20 FIG. 41 illustrates another experimental heating system 4100. Heating system 4100 includes a signal generator 4102. Signal generator 4102 can be an HP 8165A signal generator (available from Hewlett Packard Corporation). Signal generator 4102 is used to generate a low level radio frequency signal having a frequency between 10 MHz and 15 MHz. Signal generator has a control panel  
25 4103 that allows a user to manually select the frequency of the generated radio frequency signal. The output level of the signal is also controllable from control panel 4103, or from a controller 4114. The output level of the generated RF signal can vary from 0 Volts to 1 Volt peak to peak into 50 ohms, or 0 dBm.

30 Controller 4114 is interfaced to signal generator through a general purpose interface board (GPIB) (not shown). In one embodiment, controller 4114 is a personal computer (PC) running the Windows® operating system. A visual C++

program that provides a user interface for controlling the output level of signal generator 4102 is configured to run on controller 4114.

The low level RF signal generated by signal generator 4102 is provided to the input of a broadband RF amplifier 4106 using a coaxial cable 4104. Preferably, broadband RF amplifier 4106 is the A1000 broadband amplifier sold by ENI of Rochester, NY, and coaxial cable 4104 is a standard RG58 coaxial cable. Broadband Amplifier 4106 amplifies the low level RF signal by 60 dB, thereby providing a 1 Kilowatt output into a 50 ohm load for a 1 milliwatt (0 dBm) input. If the low level RF input signal provided to amplifier 4106 consists of a timed pulse, amplifier 4106 will amplify the pulse to produce a high level pulse output.

Connected to the output of broadband amplifier 4106 is a directional coupler 4110. A suitable directional coupler can be purchased from Connecticut Microwave Corporation of Cheshire, Connecticut. Directional coupler 4110 is connected to the output of amplifier 4106 through an RF cable 4107, such as an RG393 RF cable. The output of directional coupler 4110 is connected to an impedance matching circuit 4122 using RG393 RF cable 4112.

The function of impedance matching circuit 4122 is to match a 50 ohm input impedance to a variable impedance of probes 602 and 604 and the sample 410. Typical impedances of probes 602 and 604 in combination with sample 410 range from 200 ohms up to 500 ohms.

Directional coupler 4110 has a reflected power output port 4111 that is connected to an oscilloscope 4118. Preferably, oscilloscope 4118 is a TDS210 digital real time oscilloscope available from Tektronix, Inc. Directional coupler 4110 provides a signal representing the amount of reflected power to oscilloscope 4118, which then displays the magnitude of the reflected power.

The process for heating sample 410 using heating system 4100 will now be described. Initially, an operator interacts with a user interface on controller 4114 to activate signal generator 4102 so that it produces a 50 millivolt RF signal. The reflected power is then observed on oscilloscope 4118. The frequency of the 50 millivolt RF signal and matching circuit 4122 are adjusted such that the

reflected power is minimized. Once the frequency and the matching circuit are adjusted such that the reflected power is minimized, the signal generator is turned off and sample 410 is placed close to probes 602 and 604.

Next, controller 4114 is used to turn on signal generator 4102 so that it once again produces a 50 millivolt RF signal. At this point, the frequency and matching circuit are adjusted again until the reflected power is minimized. On achieving the minimum reflected power, signal generator 4102 is turned off. Next, operator uses controller to direct signal generator to produce an RF signal with a voltage ranging from 100 millivolts to 1000 millivolts and with a pulse time of between 20 milliseconds and 1000 milliseconds. This low level RF signal is amplified by broadband amplifier 4106. The amplified signal is then provided to impedance matching circuit 4122 and an a RF pulsed electromagnetic field is produced at probes 602 and 604. The presence of the pulsed electromagnetic field causes sample 410 to heat.

FIG. 42 illustrates probes 4202 and 4204, which were the probes utilized to test the compositions described herein. The present invention is not limited to this or any particular probe design. Probe 4202 and probe 4204 are both 1/8 inch square copper tubes. Probe 4202 and probe 4204 both rest on a block 4250 of non-electrically conductive material, preferably, but not limited to, TEFLON™. More specifically, block 4250 has 1/8 inch square slots milled therein so that probes 4202 and 4204 are recessed into block 4250.

Probe 4202 has a proximal section 4209, a center section 4210, a transition section 4211, and a distal section 4212. Similarly probe 4204 has a proximal section 4213, a center section 4214, a transition section 4215, and a distal section 4216. Center section 4210 is parallel with center section 4212. The center to center distance between center section 4210 and center section 4212 is on half of an inch.

Proximal section 4209 diverges away from probe 4204. Similarly, proximal section 4213 diverges away from probe 4202. The center to center distance between the proximal end of proximal section 4209 and the proximal end of proximal section 4213 is about at least one and three sixteenths of an inch.

Distal section 4212 is parallel with distal section 4216 and parallel with center section 4210. The center to center distance between distal section 4212 and distal section 4216 is about at least one and three sixteenths of an inch. Transition section 4211 is between center section 4210 and distal section 4212. Similarly, transition section 4215 is between center section 4214 and distal section 4216.

The reason the distance between the proximal end of proximal section 4209 and the proximal end of proximal section 4213 is about at least one and three sixteenth of an inch is to prevent arcing at the ends of probe 4202 and 4204. For that same reason the distance between distal section 4212 and distal section 4216 is about at least one and three sixteenth of an inch.

#### *XIV. Examples*

Without further elaboration, it is believed that one skilled in the art can, using the preceding description, utilize the present invention to its fullest extent. The following preferred specific embodiments are, therefore, to be construed as merely illustrative and not limitative of the remainder of the disclosure in any way.

##### *Example 1*

Various susceptor compositions were screened for use at frequencies from about 4 MHz to 15 MHz, and at power levels from about 0.5 kW to 1 kW. RF frequencies of less than about 15 MHz are much less costly to produce and operate than RF frequencies of greater than 15 MHz. The best results consistently occurred at the upper ends of the experimental frequency and power ranges (e.g., 15 MHz and 1 kW). See FIG. 6 for a schematic diagram of the experimental set-up and equipment used for the various tests described herein. According to the present invention, a preferred composition comprises a uniform solution of PVP, NMP, and  $\text{SnCl}_2$ . A suitable susceptor composition comprises  $\text{SnCl}_2$  present in a concentration of from about 1 % to about 50 %, NMP in a concentration of from about 25% to about 75%, and PVP in a concentration of from about 1 % to

about 35%. These three components are soluble in one another. These components were mixed together to form a uniform solution that was able to be heated from about 75° F to a boiling point of about 280°F in several seconds. Acceptable results can also be obtained, for example, by substituting similar concentrations of PVP/vinyl acetate copolymer for PVP, and substituting similar concentrations of lithium perchlorate for  $\text{SnCl}_2$ . In addition, other suitable compositions include a mixture comprising ethylene/vinyl acetate copolymer in a concentration of from about 75% to about 99% and ethylene/acrylic acid copolymer in a concentration of from about 1% to about 25%, a mixture comprising  $\text{LiC}_2\text{H}_3\text{O}_2$  in a concentration of from about 1% to about 25%, ethylene/vinyl acetate copolymer in a concentration of from about 50% to about 98%, and styrenated ethylene/acrylic acid copolymer in a concentration of from about 2% to about 25%, and a mixture comprising PVP/vinyl acetate copolymer in a concentration of from about 5% to about 35%,  $\text{SnCl}_2$  in a concentration of from about 5% to about 49%, and NMP in a concentration of from about 1% to about 90%. Other composition constituent concentrations will be apparent to those of skill in the art based on the present description.

In this example, a preferred susceptor composition comprising  $\text{SnCl}_2$  in a concentration of about 33%, NMP in a concentration of about 50%, and PVP in a concentration of about 17% was prepared to bond various combinations of thin polyolefin layers of polypropylene (PP) and polyethylene (PE). This example susceptor composition resulted in a uniform dispersion of salt ions in a polymeric adhesive.

The experiment was conducted by saturating a second carrier, a thin layer of an insoluble porous carrier (in this example non-woven PP), with a small amount of the susceptor composition. The example salt-based susceptor composition provides a continuous matrix of salt ions in a polar organic medium throughout the insoluble porous carrier. As shown schematically in FIG. 3, the insoluble porous carrier 302 was sandwiched in-between two layers of PP or PE, layers 304 and 306, respectively, and then transversely heated by the application of RF energy. By RF heating at about 14-15 MHz for about 1-2 seconds at about

0.8-1 kW of power output, sufficient bonding occurred between the non-woven PP carrier and the layers of PP and PE. In this example, the strength of the bonded region was at least as strong or stronger than the PP or PE substrates themselves. The polyolefin layers to be bonded were chosen from combinations of (1) PP non-woven and (2) PE film. The results of this example are shown below in Table 1.

**Table 1**

insoluble porous carrier (saturated with PVP, NMP and $\text{SnCl}_2$ )	bonding results
PP non-woven/ PP non-woven	bonding within 1-2 seconds
PE film/ PE film	bonding within 1-2 seconds
PP non-woven/ PE film	bonding within 1-2 seconds

For each combination, the saturated insoluble porous carrier bonded the outer layers in about 1-2 seconds. There was evidence of melting in the outer layers, with some minor substrate distortion and tiny melt holes. By using the saturated PP non-woven carrier, a uniform matrix of the adhesive and susceptor components resulted in intimate contact between the adhesive component and both outer layers.

**Example 2**

The susceptor composition utilized in this example comprised  $\text{SnCl}_2$  in a concentration of about 33 %, dissolved in a mixture of NMP in a concentration of about 50 % and PVP in a concentration of about 17 %. Various PP and PE substrate surfaces were coated with the RF susceptor composition, including: (1) PP non-woven and (2) PE film. The susceptor composition was hand drawn onto each surface as a wet layer that would eventually dry, leaving a coating which was dry to the touch. RF heating tests were performed on the coated substrates. In each case, two like samples were placed together with the coated surfaces in contact with one another. The contacted surfaces were placed in a load coil that

was designed to clamp the surfaces firmly together and transversely heat a 0.25 inch x 8 inch strip of the susceptor composition. The operating frequency was about 14 MHz, and the power delivered to the coil was about 1 kW. The tests were split into two parts: using a wet susceptor composition and using a vacuum dried susceptor composition. The results are shown in Table 2.

Vacuum drying was employed in this experiment as an extreme experimental condition for comparison purposes, but is not expected to represent a commercial embodiment.

**Table 2**  
**Coated Substrates in Contact, Under 1.2 kW RF at 14 MHz**

<b>WET/DRY Substrates</b>	<b>non-woven PP/ non-woven PP</b>	<b>PE film/ PE film</b>
<b>WET</b>	successfully bonded together within 1-2 seconds	evidence of melting, slight bonding at edges of the coat of susceptor material
<b>VACUUM DRIED</b>	No evidence of heating after 1 minute.	No evidence of heating after 1 minute.

The results show that the wet susceptor composition generates enough heat within 1-2 seconds at 14 MHz and 1 kW to melt the PP non-woven or PE film in transverse heating of thin hand drawn films. Bonding is successful between layers of PP non-woven. As layers of PP non-woven are brought together, the susceptor composition is displaced into the open space between the fiber of the PP non-woven layers, allowing the two layers of PP non-woven to come together and make intimate contact, enabling bonding during re-flow of the layers.

Complete bonding was not demonstrated between layers of PE film. As layers of PE film were brought together, the susceptor mixture behaved as a hydrostatic middle layer or boundary, preventing intimate contact between the two outer polyolefin layers. It was observed that the material in the layers of PE was

more likely to partition away from the susceptor composition than to cross the susceptor composition layer during melting and re-flow. It was also observed that as the susceptor composition was vacuum dried, it lost its ability to be RF heated effectively. Results were likely due to one or more of the following factors: the precipitation of ions back into an inactive salt as the solvent volatilizes to form the dry coat; a decrease in translational mobility of any ions still supported by the dry coating, thus preventing RF heating from occurring; and, in the case of PE films, an insufficient intermolecular contact due to the smoothness of the films. According to the present invention, this problem can be solved by introducing an additive, such as a surfactant, nonvolatile solvent or, plasticizer to the composition, to achieve better attachment.

### *Example 3*

In this example, an RF activated susceptor composition was prepared from an EASTMAN AQ branched polyester (available from the Eastman Chemical Corporation) and an aqueous solution of  $\text{SnCl}_2$ . Various layers of PP non-woven and PE film were tested. The susceptor composition that was used in this example comprised  $\text{SnCl}_2$  dissolved in distilled water. This solution was blended with a branched polyester adhesive component, EASTMAN AQ35S. Suitable concentrations of the branched polyester ranged from about 25% to about 75%.

In a series of experiments, the susceptor composition was used to adhere all combinations of: (1) PP non-woven and (2) PE film substrates. In each experimental combination, the composition was first coated onto the two substrate surfaces and dried under ambient conditions similar to those used in commercial practice. The two substrates were then pressed together in the work coil with the two susceptor composition coated surfaces in contact with each other. The coated surfaces were not tacky enough at this point to result in contact adhesion between the substrates. All combinations of substrates were successfully adhered to one another by RF heating for a period of about 1 second at about 14 MHz and about 1 kW. The substrates were adhered to each other by the RF-activated



susceptor composition, instead of being bonded by re-flow of the substrate. No apparent melting or distortion of the substrate occurred. This example demonstrates that a susceptor composition coating can be dry to the touch and still be activated by RF heating.

5      *Example 3a*

Analogous to Example 3 above, the active ingredients Eastman AQ35S and  $\text{SnCl}_2$  (in constituent concentrations consistent with the parameters described above) were dissolved in NMP to form a susceptor composition. The composition was coated on a PP non-woven web and was allowed to air dry. The slightly tacky web was placed between polyolefin substrates and the assemblies were RF heated in the RF work station at 14.65 MHz and about 0.8 kW for 5 seconds. Good adhesion was obtained in each case.

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*Example 3b*

A susceptor composition capable of inductive activation was prepared with an aqueous dispersion of a sulfopolyester, EASTEK 1300 Polymer (available from Eastman Chemical Company), by addition of  $\text{SnCl}_2$ . A precipitate was recovered from this mixture and a film from this precipitate was obtained by pressing between hot platens at about 200°F and 1000 psi for a short time. This film was slightly wet and was sandwiched between a PE film and a PP non-woven web. The assembly was RF heated in the RF work station at 14.63 MHz and about 0.8 kW for 1 second. Good adhesion without substrate deformation was achieved.

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*Example 3c*

Analogous to Example 3b, additional experiments were performed in which slightly wet-to-touch thin films of the susceptor compositions of example 3b were sandwiched between two stacks consisting of multiple layers of non-

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woven PP or multiple layers of non-woven PP laminated to PE film. Three different types of sandwiched assemblies were tested, including: (1) 8 layers of non-woven PP / susceptor composition / 8 layers of non-woven PP, (2) 2 layers of PP non-woven laminated to PE film / susceptor composition / 2 layers of PP non-woven laminated to PE film (with PP non-woven facing PE film at the stack interface), and (3) 4 layers of PP non-woven/ susceptor composition / 4 layers of PP non-woven. In each case, the assemblies were RF heated in the RF work station at 14.63 MHz and about 0.8 kW for 1 second. In all cases, good adhesion occurred between the multilayer stacks without causing distortion to the stacks.

In this experiment, each multilayered stack was pre-assembled using a conventional contact adhesive, and the two stacks were later adhered to one another using the susceptor composition. However, it is contemplated in the practice of the invention that each multilayer stack can be preassembled using a susceptor composition to either simultaneously or sequentially bond or adhere the various layers of each stack.

#### ***Example 4***

Based on the success of the susceptor compositions tested in Examples 1-3b, other susceptor compositions were made and tested. In this example, sample compositions (in constituent concentrations consistent with the parameters described above) were tested in half-filled test tubes nearly centered within the coil of the RF equipment described in FIG. 40. Various settings for voltage input and frequency of the current were investigated. Table 3 summarizes the results of these experiments. The effectiveness of RF heating is shown by the time required for the samples to boil or to rise to the indicated temperature.

**Table 3**

**Selected Test Tube Experiments with Potential Susceptors**

	Materials	Frequency, MHz	Input, mV	Field, V	Time to boil or temperature rise
	<b>Solid Salts</b>				
5	SnCl <sub>2</sub> x 2H <sub>2</sub> O	13.73	300	7.8	Poor heating
	SnCl <sub>2</sub> x 2H <sub>2</sub> O	13.73	600	15	39 sec., 122°F
	SnCl <sub>2</sub>	13.74	800	~20	Poor heating
	LiClO <sub>4</sub> x3H <sub>2</sub> O	13.74	800	~20	10 sec.
	<b>Aqueous Solutions</b>				
10	Distilled Water	13.75	800		80 sec., 117°F
	LiClO <sub>4</sub>	13.73	800	~20	3 sec.
	SnCl <sub>2</sub>		800	~20	5 sec.
	NaCl	13.67	800	~2.2	3 sec.
	NaCl	5.833	320	~2.2	20 sec.
15	NaCl	3.719	10	58	30 sec.
	Li-acetate	13.73	1000		10 sec.
	<b>Nonaqueous Solutions</b>				
	NMP	13.74	800		60 sec., 89°F
	NMP/SnCl <sub>2</sub>	13.74	800		47 sec., 350°F
20	NMP/PVP/SnCl <sub>2</sub>	13.73	685	20	8 sec., 142°F
	NMP/PVP/LiClO <sub>4</sub> x 3H <sub>2</sub> O	13.73	643	20	6 sec., 135°F
	NMP/PVP/Li-acetate x 2H <sub>2</sub> O	13.73	600	18	18 sec.
	<b>Liquid Samples</b>				
	Uni-REZ 2115	13.75	1000		75 sec., 126°F
25	MICHEM 4983	13.77	10	40	30 sec., 178°F
	MICHEM ACRYLIC I	13.73	800		4 sec.

These tests show that, as expected, aqueous solutions of various susceptor, such as salts, coupled very well with the RF energy. All tested susceptor compositions came to a boil within 3 - 30 seconds. As discussed above, SnCl<sub>2</sub> dissolved in NMP also coupled very effectively. Although a boiling time of 47 seconds is shown in this experiment, the temperature of 350 °F reached by this mixture is substantially higher than is required for heat bonding polyolefins.

In the section on nonaqueous solutions in Table 3, it can be seen that while NMP is only a weak susceptor in its own right, it couples very effectively with the

RF energy, when a variety of salts are dissolved in it. It was also observed that the solution's ability to solubilize salts seems to be enhanced when PVP is dissolved in it. Since the compositions shown in Table 3 were not optimized, the RF heating capability of the solutions appear to be very good.

The last section of Table 3 shows the RF heating capability of some liquid polymers. They range from very mild coupling ability to very powerful coupling ability in the case of MICHEM ACRYLIC 1 (available from Michelman Corporation), a styrenated ethylene-acrylic acid polymer.

### Example 5

This experiment tested the compatibility of various film forming and adhesive polymers with modifying resins and additives and with inorganic or organic susceptors (in constituent concentrations consistent with the parameters described above). A series of experiments were conducted with low-density polyethylene (LDPE) as the substrate, as summarized in Table 4.

**Table 4**

#### Bonding Feasibility Experiments With a Low Density Polyethylene Substrate

Sample	Frequency, MHz	Input, mV	Time	Adhesion	Substrate
ELVAX 40W + UNI-REZ 2641	14.55	500	1 min.	None	LDPE
ELVAX 40W + UNI-REZ 2641 + Li-acetate	The Li-acetate did not mix with the polymers under pressing conditions				
ELVAX 40W + N,N-ethylenebis-stearamide @80/20	14.54	500	1 min.	None	LDPE
ELVAX 40W + N,N-EbSA @80/20 + MICHEM ACRYLIC 1	14.54	650	1 min.	Slight	LDPE
MICHEM ACRYLIC 1	14.54	650	30 sec.	Slight	LDPE
PRIMACOR 3460 + Li-acetate	The Li-acetate did not mix with the polymers under pressing conditions				

**Table 4**

**Bonding Feasibility Experiments With a Low Density Polyethylene Substrate**

	Sample	Frequency, MHz	Input, mV	Time	Adhesion	Substrate
5	N,N-EbSA +ELVAX 40W + Poly (ethylene- maleic anhydride) + Li- acetate +MICHEM PRIME 4983	14.6	850	30 sec.	Partial	LDPE
	ELVAX 40W + Li-acetate +MICHEM PRIME 4983	14.6	850	30 sec.	Partial	LDPE
10	ELVAX 40W + MICHEM PRIME 4983	14.6	850	15 sec.	Partial, good	LDPE
	UNI-REZ 2641	14.6	850	30 sec.	Partial, good	LDPE
	ELVAX 40W + UNI-REZ 2641 + MICHEM PRIME 4983	14.6	850	30 sec.	Partial, good	LDPE
15	Polyethylene (AS) + AlliedSignal GRADE A-C + MICHEM PRIME 4983	14.6	850	15 sec.	Partial, good	LDPE
	Polyethylene (AS) + MICHEM PRIME 4990	14.6	850	30 sec.	Partial, good	LDPE
20	ELVAX 40W + Fe <sub>2</sub> O <sub>3</sub>	14.6	850	30 sec.	None	LDPE
	ELVAX 40W + MICHEM PRIME 4990	14.6	850	15 sec.	Good	LDPE
	ELVAX 410 + N,N-EbSA + PEMA	14.6	850	30 sec.	None	LDPE
25	ELVAX 410 + N,N-EbSA + PEMA + KEN-REACT LICA 44	14.6	850	30 sec.	None	LDPE
	ELVAX 410 + N,N-EbSA + PEMA +MICHEM ACRYLIC 1	14.6	850	10 sec.	Partial, good	LDPE
30	ELVAX 40W + Li- acetate/MICHEM ACRYLIC 1 (paste)	14.6	850	15 sec.	Good	LDPE
	ELVAX 40W + MICHEM ACRYLIC 1	14.6	850	15 sec.	Good	LDPE

**Table 4**

**Bonding Feasibility Experiments With a Low Density Polyethylene Substrate**

Sample	Frequency, MHz	Input, mV	Time	Adhesion	Substrate
ELVALOY EP 4043 MICHEM ACRYLIC 1	14.6	850	30 sec.	None	LDPE
ELVAX 40W + UNI-REZ 2641 + STEROTEX HM wax + MICHEM ACRYLIC 1	14.6	850	30 sec.	None	LDPE
SnCl <sub>2</sub> + NMP + PVP K29-30 + 0.1% SURFYNOL 104PA	14.54 14.54	850 850	10 sec. 30 sec.	None None	Mylar/dry Mylar/dry

Several susceptors which had been very effective in test tube runs did not always lead to good bonding or adhesion in these trials. MICHEM ACRYLIC 1 is a good example. Likely reasons include: (a) the susceptors were only effective wet and lost their coupling ability when used in a dry film, or (b) the susceptor itself was not a good adhesive and formed a barrier to melted PE bonding to itself. There was an indication that the second reason prevailed when it was shown that the susceptors performed better when blended with PE or EVA which could act as hot melt adhesives for the substrates. Better results were obtained with MICHEM PRIME 4983 and 4990, MICHEM ACRYLIC 1, and UNI-REZ 2641 in combination with either PE or EVA. However, N,N-ethylene-bisstearamide, which was described in the Degrand reference, was not very effective in these experiments. A number of trials provided partial adhesion, as noted in Table 4, which were likely caused by the inability of the film holder to clamp the substrates tightly and flat. Although the shortest successful heating times were on the order of 10 to 15 seconds, which would be too long for a commercial operation, the results are positive, in that both the adhesive compositions and the operation of the film station can be further optimized without undue experimentation.

**Example 6**

Another series of experiments were performed with other polyolefin substrates, including a PP non-woven. These trials are summarized below in Table 5. Very good bonds were obtained with several compositions at dwell times down to about 1 second. While this may be too long for some commercial applications, it is highly encouraging for trials that were not optimized with regard to either the susceptor composition or the test equipment.

**Table 5**

**Bonding Experiments With Polyolefin Substrates**

Sample	Frequency, MHz	Input, mV	Time	Adhesion	Substrate
ELVAX 40W (Du Pont 40% vinyl acetate to polyethylene)	14.61	850	30 sec.	None	PP nonwoven
ELVAX 40W + MICHEM PRIME 4990	14.61	850	15 sec.	Good	PP nonwoven
ELVAX 40W + UNIREZ 2641 + STEROTEX HM wax + MICHEM ACRYLIC 1	14.61	850	30 sec.	None	PP nonwoven
ELVAX 40W + UNIREZ 2641 + MICHEM ACRYLIC 1	14.61	850	15 sec.	Good	PP nonwoven
SnCl <sub>2</sub> + NMP + PVP K29-30 + 0.1% SURFYNOL104PA	14.61	850	1 sec.	Good	PP nonwoven
(Dried 90 min)	14.61	850	5 sec.	Good	PP nonwoven
(Dried 15 hrs)	14.61	850	25 sec.	None	PP nonwoven
(Dried 15 hrs + NMP)	14.61	850	2 sec.	Good	PP nonwoven
SnCl <sub>2</sub> + NMP + PVP K29-30 + 0.1% SURFYNOL 104PA	14.61	850	1 sec.	None	PE/PE

Sample	Frequency, MHz	Input, mV	Time	Adhesion	Substrate
SnCl <sub>2</sub> + NMP + PVP K29-30 + 0.1% SURFYNOL 104PA	14.61 14.61	850 850	1 sec. 2 sec.	Slight Good	PP/PP n/w PP/PP n/w
ELVAX 40W + UNI-REZ 2641 + Indium tin oxide	14.61	850	30 sec.	None	PP nonwoven
SURFADONE LP-300 + SnCl <sub>2</sub>	14.61	850	30 sec.	None	PP nonwoven
PVP/VA S-630 + SnCl <sub>2</sub> + NMP	14.61 14.61	850 850	1 sec. 1 sec.	Good Slight	PP nonwoven PE/PE
PVP/VA S-630 + SnCl <sub>2</sub> + NMP + Fumed silica	14.61 14.61 14.61	850 850 850	1 sec. 5 sec. 2 sec.	Good Slight Slight	PP nonwoven PE/PE PP/PP n/w
ELVAX 40W + UNI-REZ 2641 + Li-acetate, pressed	14.61	850	30 sec.	None	PP nonwoven
Li-acetate x 2H <sub>2</sub> O	14.61	850	30 sec.	None	No melting
Mg(NO <sub>3</sub> ) <sub>2</sub> x 6H <sub>2</sub> O	14.61	850	30 sec.	None	No melting
MgAc x 4H <sub>2</sub> O	14.61	850	30 sec.	None	No melting
Stearic acid + Cetyl alcohol + Mg(NO <sub>3</sub> ) <sub>2</sub> x 6H <sub>2</sub> O	14.61 14.61	850 850	2 sec. 10 sec.	None Good	PP nonwoven PP nonwoven
EVA AC-400 + SURFADONE LP-300 + SnCl <sub>2</sub>	14.61	850	2, 5 and 10 sec.	None	PP nonwoven

Some of the better results were obtained with compositions containing PVP or PVP/VA and SnCl<sub>2</sub> salt dissolved in NMP (in constituent concentrations consistent with those discussed above). It was shown, however, that thorough drying of the susceptor composition eliminated its ability to couple with the RF field. It appears that the mobility, provided by the presence of at least a small amount of NMP solvent, is important for efficient coupling to the applied RF field. This mobility function can be provided by the selection of an appropriate nonvolatile plasticizer, such as epoxidized oils, polyhydric alcohols, substituted amides, sulfonamides, aryl and alkyl aryl phosphates, polyesters and a wide variety



of esters, including benzoates, phthalates, adipates, azelates, citrates, 2-ethylbutyrates and hexoates, glycerides, glycollates, myristates, palmitates, succinates, stearates, etc. Plasticizers are used to solvate a material, and thus improve its molecular mobility if it has become too rigid.

5 In general, ethylene co-polymers with functionality providing (a) enhanced compatibility and (b) ionic or highly polar constituents are effective in bonding or adhering substrates, together with salts that are either soluble or readily dispersed in the polymer matrix. There is also evidence that in some compositions mobility of the dipoles must be assured. This was achieved in the presence of such high-boiling solvents as NMP. It can also be extrapolated that other high boiling solvents or non-volatile plasticizers can achieve the same effects with more reproducible results.

10 These example susceptor compositions utilize a combination of polar components and hydrated salts in a polymer matrix plasticized with high boiling and high dielectric constant additives that are activatable at a relatively low frequency of about 15 MHz.

15 The methods and experiments set forth above will allow those of skill in the art to determine without undue experimentation that a particular mixture would be suitable for bonding or adhering substrates according to the present invention.

### 20 *Example 7*

This example demonstrates RF-heatable thermoplastic compositions derived from the combination of various ion-containing polymers with glycerin. These compositions are shown to be significantly more susceptible to RF heating than either the component ion-containing polymers or glycerin are by themselves.

### Example 7a

Several compositions comprising 70 wt% sulfonated polyesters in 30 wt% glycerin were prepared. Each sample was prepared by first mixing 14 grams of sulfopolyester material with 6 grams of glycerin in a 60 milliliter glass jar. The open topped jar was heated in a convection oven at 165C for 1 hour. After thirty minutes, the composition was removed from the oven and hand stirred for 1 minute and then immediately returned to the oven. After an additional 30 minutes of heating at 165C, the composition was removed from the oven and hand stirred for 1 minute. While the composition was still molten, it was hand-drawn into a 1 inch wide by 3 inch long by 0.006 inch thick coating on the surface of a 0.004 inch thick sheet of transparency film (PP2500 series 3M transparency film) that was supported on a 180° F 10 inch x 10 inch Corning model PC620 hot plate. Immediately after the composition was coated to the film, the coated film was removed from the hot plate and allowed to cool to room temperature. The samples were then evaluated for film properties and RF heating.

The RF equipment setup used for testing this example and examples 8-16 consisted of the RF probes described in Figure 42 and the RF equipment described in Figure 41. Unless otherwise noted, in each case a 1-inch x 3-inch sample (410) was placed over the RF probes as shown in Figure 41. The distance from the surface of the probes to the sample was about 0.016 inches. The sample was heated at about 1 KW input power into the tuned heat station 4122 (or impedance matching circuit 4122) at about 13.5 MHz for the time required to cause observable heating and melting in the activation region of the RF probes.

Table 6

BRANCHED SULFONATED POLYESTERS ... (Eastman A Polyesters, Available from Eastman Chemical Company, Kingsport, TN, USA)			
Experiment #	Composition Description	Film Properties	Time to Melt (s)
1	70 wt% AQ1045 30 wt% glycerin	Clear, tacky, flexible.	0.25
2	70 wt% AQ1350	Clear, tacky,	0.25

	30 wt% glycerin	flexible.	
3	70 wt% AQ1950 30 wt% glycerin	Clear, tacky, flexible.	0.25
4	70 wt% AQ14000 30 wt% glycerin	Clear, tack, flexible.	0.25
<b>LINEAR SULFONATED POLYESTERS ...</b> (Eastman A Polyesters, Available from Eastman Chemical Company, Kingsport, TN, USA)			
Experiment #	Composition Description	Film Properties	Time to Melt (s)
5	70 wt% AQ35S 30 wt% glycerin	White, tack-free, flexible.	0.5
6	70 wt% AQ38S 30 wt% glycerin	White, tack-free, flexible.	0.5
7	70 wt% AQ55S 30 wt% glycerin	Clear, tack-free, flexible.	0.2

#### *Example 7b*

Several compositions comprising 70 wt% ethylene acrylic acid copolymers in 30 wt% glycerin were prepared. Each sample was prepared by first mixing 52 grams of ethylene acrylic acid copolymer material (a 25 wt% solids emulsion) with 5.57 grams of glycerin in a 60 milliliter glass jar. The combined materials were then mixed for 10 minutes to result in an emulsion. The resulting emulsion was then cast onto a sheet of 0.004 inch thick transparency film (PP2500 series 3M transparency film) at room temperature. The cast emulsion was then allowed to dry-down under a heat lamp to form a film. The samples were then evaluated for film properties and RF heating.

Table 7

ETHYLENE ACRYLIC ACID COPOLYMERS (Acid Form) ... (MICHEM 4983P, Available from Michelman Incorporated, Cincinnati, OH, USA)			
Experiment #	Composition Description	Film Properties	Time to Melt (s)
1	100 wt% MICHEM 4983P	Clear, colorless, brittle, tack-free.	28
2	70 wt% MICHEM 4983P 30 wt% glycerin	Clear, colorless, less brittle, tack-free.	0.5
3	50 wt% MICHEM 4983P 50 wt% glycerin	Clear, colorless, flexible, tack-free.	0.4
ETHYLENE ACRYLIC ACID COPOLYMERS (Sodium Salt Form) ... (MICHEM 48525P, Available from Michelman Incorporated, Cincinnati, OH, USA)			
Experiment #	Composition Description	Film Properties	Time to Melt (s)
4	100 wt% MICHEM 48525P	Clear, colorless, brittle, tack-free.	No Heating in 1 minute.
5	70 wt% MICHEM 48525P 30 wt% glycerin	Clear, colorless, flexible, tack-free, rubbery.	0.5
6	50 wt% MICHEM 48525P 50 wt% glycerin	Clear, colorless, flexible, tack-free, rubbery.	0.2 - 0.4

*Example 7c*

Several compositions comprising 70 wt% vinyl acetate acrylic copolymers in 30 wt% glycerin were prepared. Each sample was prepared by first mixing 46.67 grams of vinyl acetate acrylic copolymer material (a 55 wt% solids emulsion) with 3 grams of glycerin in a 60 milliliter glass jar. The combined materials were then mixed for 10 minutes to result in an emulsion. The resulting emulsion was then cast onto a sheet of 0.004 inch thick transparency film (PP2500 series 3M transparency film) at room temperature. The cast emulsion was then allowed to dry-down under a heat lamp to form a film. The samples were then evaluated for film properties and RF heating.

Table 8

VINYL ACETATE ACRYLIC COPOLYMERS ... (ROVACE HP3442, Available from Rohm and Haas, Philadelphia, PA, USA)			
Experiment #	Composition Description	Film properties	Time to Melt (s)
1	100 wt% HP3442	Clear, colorless, flexible, tack-free.	No Melting in 1 minute
2	90 wt% HP3442 10 wt% glycerin	Clear, colorless, flexible, very tacky, with good cohesion.	0.3

*Example 7d*

This example demonstrates how the addition of glycerin as well as adjustments in pH to gelatin solutions can affect the properties of derived gels.

Several compositions were prepared as solutions of a commercially available gelatin (Eastman 45Y56-853-3V0-6CS available from Eastman Gelatine Corporation). All compositions had water. Some solutions had glycerin added to them. Some solutions had their pH adjusted by the addition of 10N NaOH or 6N HCl. The compositions were prepared as follows:

Composition #1 was prepared by adding 70 grams of gelatin to 280 grams of water and stirring and heating the resulting mixture at about 65°C for 1 hour to obtain a solution. The solution had a pH of 6.18 at 65°C.

Composition #2 was prepared by stirring 6 grams of glycerin into 70 grams of composition #1. The solution had a pH of 5.8 at 65°C.

Composition #3 was prepared by stirring drops of 10N NaOH (about 25 drops) into 125 mls of composition #1, until the resulting solution had a pH of 10.1 at 65°C.

Composition #4 was prepared by stirring 8.51 grams of glycerin into 99.3 grams of composition #3 to result in a solution with a pH of 10.1 at 65°C.

Composition #5 was prepared by stirring drops of 6N hydrochloric acid (about 90 drops) into 125 grams of composition #1, until the resulting solution had a pH of 1.9 at 65°C.

Composition #6 was prepared by stirring 5.361 grams of glycerin into 62.57 grams of composition #5 to result in a solution with a pH of 1.9 at 65°C.

Each gelatin solution was cast onto a sheet of transparency film (3M PP2500 Transparency Film) and allowed to set-up at room temperature to form a gel film. The gels differed in their film properties and in their RF-heating properties as described in Table 9.

Gelatin films (susceptors) may not act as a good adhesive on low energy surfaces, such as PE, PP, etc. However, they are expected to perform effectively as adhesives on polar substrates, such as paper, Kraft paper, linear boards, wood, etc.

**Table 9**

<b>GELATINS ...</b> (Eastman 45Y56-853-3V0-6CS gelatin, Available from Eastman Gelatin, USA)			
<b>Experiment #</b>	<b>Composition Description</b>	<b>Film properties</b>	<b>Time to Melt (s)</b>
1	gelatin pH 5.8 at 65°C	Brittle w/ poor adhesion to substrate.	No Heating in 1 minute.
2	70 wt% gelatin 30 wt% glycerin pH 5.8 at 65°C	Flexible w/ good adhesion to substrate.	10
3	gelatin pH 10.1 at 65°C	Brittle w/ poor adhesion to substrate.	No Heating in 1 minute.
4	70 wt% gelatin 30 wt% glycerin pH 10.1 at 65°C	flexible w/ good adhesion to substrate.	4
5	gelatin pH 1.9 at 65°C	Brittle with poor adhesion to substrate.	17
6	70 wt% gelatin 30 wt% glycerin pH 1.9 at 65°C	Flexible w/ good attachment to substrate.	< 1

**Example 8**

Several compositions were prepared by mixing various polar materials with a representative ionomer (Eastman AQ35S Sulfopolyester). In each case, the

compositions are demonstrated to be more susceptible to RF heating than the component ionomer or polar material by themselves.

Each composition is comprised of 70 wt% AQ35S in 30 wt% polar material. Each sample was prepared by first mixing 46.67 grams of AQ35D (a 30 wt% solids emulsion) with 6 grams of polar carrier in a 60 milliliter glass jar. The combined materials were then mixed for 10 minutes to result in an emulsion. The resulting emulsion was then cast onto a sheet of 0.004 inch thick transparency film (PP2500 series 3M transparency film) at room temperature. The cast emulsion was then allowed to dry-down under a heat lamp to form a film. The samples were then evaluated for film properties and RF heating.

**Table 10**

VARIOUS POLAR MATERIALS USED IN COMPOSITIONS COMPRISING: 70 wt% EASTMAN AQ35S / 30 wt% POLAR MATERIAL.			
Experiment #	Composition Description	Film properties	Time to Melt (s)
1	70 wt% EASTMAN AQ35S 30 wt% <b>Ethylene Glycol</b> (The DOW Chemical Company, Midland, MI, USA)	Clear, tack-free, flexible.	1
2	70 wt% EASTMAN AQ35S 30 wt% <b>1,2-propylene glycol</b> (The DOW Chemical Company, Midland, MI, USA)	White, slightly tacky, flexible.	0.150
3	70 wt% EASTMAN AQ35S 30 wt% <b>polyethylene glycol 200</b> (Union Carbide Chemicals and Plastics Company Inc., Danbury, CT, USA)	Clear, yellow, tacky, flexible.	0.4
4	70 wt% EASTMAN AQ35S 30 wt% <b>polyethylene glycol 8000</b> (Union Carbide Chemicals and Plastics Company Inc., Danbury, CT, USA)	Cloudy, orange, tack-free, w/ some undissolved polyethylene glycol, flexible.	12
5	70 wt% EASTMAN AQ35S 30 wt% <b>hexylene glycol</b> (Shell Chemical Company, Houston, TX, USA)	White, tack-free, flexible.	1.5
6	70 wt% EASTMAN AQ35S 30 wt% <b>diethylene glycol</b>	Clear, slightly tacky, flexible.	.25

	(The DOW Chemical Company, Midland, MI, USA)		
7	70 wt% EASTMAN AQ35S 30 wt% <b>glycerin</b> (The Procter and Gamble Company, Cincinnati, USA)	Clear, tack- free, flexible.	<0.5
8	70 wt% EASTMAN AQ35S 30 wt% <b>sorbitol</b> (Sigma, St. Louis, MO, USA)	Slightly cloudy, slightly tacky, flexible.	2
9	70 wt% EASTMAN AQ35S 30 wt% <b>NPC-ST-30</b> , Colloidal silica in ethylene glycol monopropyl ether. (Nissan Chemical Company, Japan; New York Office, Tarrytown, NY, USA)	Clear, yellow, slightly tacky, flexible.	10
10	70 wt% EASTMAN AQ35S 30 wt% <b>EG-ST</b> , Colloidal silica in ethylene glycol. (Nissan Chemical Company, Japan; New York Office, Tarrytown, NY, USA)	Slightly cloudy, slightly tacky, flexible.	0.2
11	100 wt% EASTMAN AQ35S NO ADDED POLAR MATERIALS (CONTROL)	Clear, tack- free, flexible.	NO HEAT in 1 minute.
12	70 wt% EASTMAN AQ35S 30 wt% <b>N-methylpyrrolidone</b> (Aldrich Chemical Co., Inc. Milwaukee, WI)	Clear, tack- free, flexible.	2.8
13	70 wt% EASTMAN AQ35S 30 wt% <b>dimethyl formamide</b> (Aldrich Chemical Co., Inc. Milwaukee, WI)	Clear, tack- free, flexible.	0.3
14	70 wt% EASTMAN AQ35S 30 wt% <b>formamide</b> (Aldrich Chemical Co., Inc. Milwaukee, WI)	Clear, slightly tacky, flexible.	0.2
15	70 wt% EASTMAN AQ35S 30 wt% <b>dimethyl sulfoxide</b> (Aldrich Chemical Co., Inc. Milwaukee, WI)	Clear, slightly tacky, flexible.	0.15

25

30



**Example 9**

Thermoset polymers are a class of polymeric systems formed by chemical (usually covalent bonding) reaction of lower molecular weight functional building blocks. For instance, epoxy thermoset polymers are formed by the reaction of oxirane groups of epoxy compounds with other functional groups such as hydroxyl, carboxyl, amine etc. In the case of urethanes, isocyanate groups are reacted with functional groups such as amines, hydroxyls etc. Chemical reactions of the functional groups of the building blocks typically need energy source such as heat, radiation and presence of catalyst. The reaction product resulting from such an interaction leads to crosslinking between the functional groups of the building blocks which in turn gives a cured polymeric system with many desirable properties such as improved heat, chemical and solvent resistance, enhanced strength and mechanical properties etc. A key feature of thermoset systems is the fact that once the crosslinks are formed in the cured state it is very difficult to reverse it.

A convenient way to study the crosslinking reaction in a thermoset system is to follow the gelling reaction. At the start of the crosslinking reaction, viscosity of the initial reaction mixture is low. In the presence of appropriate catalyst and energy source, chemical crosslinking starts to take place with increase in molecular weight and viscosity. After a critical stage of the crosslinking reaction has taken place, the system sets up to an insoluble (in a solvent such as MEK in which the starting compounds are soluble) gel. Physico-chemically, chemical bonds are being formed leading to a network structure of the cured system. It has been shown that many of the properties of a thermoset system (such as glass transition temperature, solvent and chemical resistance, mechanical properties etc) can be readily correlated to the gel content of the system.

The degree of cross linking of various thermoset systems was assessed by measuring the gel content of formulation after exposure to RF field to different time and energy levels. Increase in gel content of a given composition after RF

exposure (compared to the gel content of the same composition after air drying for several hours) is taken as a measure of cure of the thermoset system.

Typical gel measurements were carried out as follows:

5 A sample of the formulation is applied to a glass slide. The sample is air dried for a 1-2 hours so the applied layer is dry to touch. Sample weight is noted as "A" after taking into account the tare weight of glass slide. Then it is exposed to RF source (in the case of control experiments, the sample is put in a conventional laboratory oven at a set temperature and time). The cured sample is cooled down to ambient temperature. The glass slide containing the cured sample  
10 is dipped in 40ml of MEK for 10 minutes. The slide is taken out and air-dried prior to weighing. Sample weight is noted as B.

*% Gel content is calculated as  $(B/A) \times 100$*

15 It is worth noting that gel content as measured by the above procedure gives only the initial cure state of the thermoset system. Typically, crosslinking reaction progress further upon aging leading to a higher cured state of thermoset system.

In the following experiments, the following materials were used:

EPON 828: Diglycidylether of bisphenol-A from Shell Chemicals.

20 ANCAMINE 2441 catalyst, a modified polyamine from Air Products & Chemicals Inc.

EpiRez dispersion, a bisphenol A based epoxy dispersion from Shell Chemicals.

Epicure 8536-MY60, an amine curing agent from Shell Chemicals.

MAINCOTE HYDUR, a self-reactive acrylic emulsion from Rohm & Haas.

Aropol 7241, an isophthalic polyester (unpromoted) from Ashland Chemical.

25 KELSOL 5293, a water dispersible polyester from Reichhold.

CYMEL 385, butylated urea formaldehyde resin from Cytec Industries.

DESMODUR-W, an aliphatic diisocyanate {CAS #5142-30-1, (4-isocyanatocyclohexyl) methane} from Bayer

FORMREZ 11-36, a polyester diol from Witco

T-12 catalyst, dibutyltindilaurate from Air Products & Chemicals inc.  
Eastman A 35 D sulfonated branched polyester from Eastman Chemicals.

A. *Epoxy Resins:*

5 Epoxy resins are typically cured to a thermoset state by application of heat  
in the presence of catalysts such as amines, acids, anhydrides etc. By proper  
selection of epoxy resin, catalyst (amine, acid etc.) and optionally a polar carrier  
such as water, glycerin and similar high dielectric constant liquids, it is possible to  
formulate RF cured thermoset epoxy systems of potential interest in diverse  
10 applications, such as: adhesives and coatings for conventional and spray  
applications on plastics, metals, wood, etc; corrosion resistant coatings; industrial  
and protective coatings; top coats; automotive coatings; lamination of composites;  
laminating adhesives; bonding of structural composites; inks and decorative  
coatings; barrier coatings; etc.

15 The effect of time and temperature on some thermally cured epoxy resin  
systems using typical cure conditions is shown in this example. The composition  
included:

EPON 828 resin	3 parts
ANCAMINE 2441 catalyst	0.3 parts

20 The above composition was air dried without any heat and the gel content  
measured. It was found to be zero showing that the resin is not cross-linked to a  
cured system.

The above composition was heated to 130 deg C for 5 minutes and the gel  
content of the sample was found to be 11%. This shows that there is some  
crosslinking occurring under this condition.

25 The above composition was heated to 130 deg C for 15 minutes and the  
gel content was found to be 48%. As expected, longer exposure to higher  
temperature increases crosslink density and gel content.

The above composition was heated to 120 deg C for 20 minutes and the  
gel content was found to be 42.5%. This shows that longer exposure time at a

lower temperature compared to previous experiment did not increase gel content. From this observation, one can conclude that temperature has a more significant influence on crosslink density and gel content of the system.

The above composition was exposed to 120 deg C for 30 minutes and the gel content was found to be 73.5%.

The main conclusion from the above experiments is that a fairly long time (order 30 minutes) is needed to reach a high gel content thermoset epoxy resin system, cured by conventional thermal energy.

In the next series of experiments, similar epoxy compositions were evaluated when exposed to RF (14.7 MHz) for various lengths of time and energy. The composition included:

EPON 828	2.1 Parts
ANCAMINE 2441	0.5 parts

The above composition was air-dried and the gel content of the dried sample was found to be 10.7%. This shows that there is a very small level of gel in the air-dried (1 hour) sample.

10% Glycerin was added to the above composition and the sample air-dried for 1 hour and its gel content was found to be 4.3%. This data shows that glycerin tends to solubilize the gel under air dry condition.

The above composition (without glycerin) was applied onto a glass slide and was exposed to 500 mv for 2.5 minutes and its gel content was found to be 8.5%. This shows that there is not much activation under this level of RF energy.

The 10% glycerin composition was applied to a glass slide and the sample was exposed to 500 mV for 2.5 minutes. Gel content of the sample was found to be 77.3%. This result shows that addition of glycerin enhances the RF susceptibility of the resin and high level of crosslinking is achieved.

These experiments clearly show that epoxy resins can be activated in a very short period of time (compared to thermal curing conditions), especially in the presence of a polar carrier such as glycerin.

In the next series of experiments, another type of epoxy and curing agent was tested. The composition included:

EPI-REZ 3520-WY-55

2 parts

Epicure 8536-MY60

1 parts

5 The above composition was applied to a glass slide and activated under 100 mV for 5 minutes. No heat was noted and the gel content was found to be 19.3%

The same composition was activated under 500 mV for 5 minutes. Gel content was found to be 52.5%. This shows that higher power compared to first experiment is needed for crosslinking to take pace.

10 10% Glycerin was added to the above composition and the sample, after drying on a glass slide, was activated for 5 minutes under 500 mV. The gel content was found to be 52.4%, which is very similar to what was obtained without any glycerin. This result shows that the presence of glycerin or other carrier is not necessary for RF activation in all cases, especially if the resin system is water based such as the Epi Rez resin.

15 **B. Acrylic system:**

In this series of experiments, the use of RF activation for an acrylic class of resin is demonstrated.

20 MAINCOTE HYDUR 30, a water based acrylic emulsion with carboxyl and unsaturation functionalities from Rohm and Haas was tested. The sample was air-dried and its gel content was found to be 37%. This result shows that the unsaturation in the acrylic resin results in some crosslinking due to air oxidation, as seen in drying oils and alkyd resins.

25 10% glycerin was added to MAINCOTE HYDUR 30 and the gel content of the air dried sample was found to be 4.6%. This result shows that glycerin acts as good solvent for the air-dried sample.

MAINCOTE HYDUR 30 was applied to a glass slide and the sample exposed to 500 mV for 2.5 minutes. The gel content was found to be 61.5%. This clearly shows that RF field activates the acrylic resin leading to high levels of crosslinking.

10% Glycerin/MAINCOTE HYDUR 30 was exposed to 500 mV for 2.5 minutes. The gel content was found to be 92.3%. This result shows that presence of glycerin promotes RF coupling with the resin.

MAINCOTE HYDUR 30 was exposed to 700 mV for 2.5 minutes and the gel content was found to be 81.8%. This shows that increased RF power promotes crosslinking of acrylic resin.

10% Glycerin/MAINCOTE HYDUR 30 was exposed to 700 mV for 2.5 minutes and the gel content of the sample was found to be 100%. This result shows the beneficial role of glycerin in promoting RF activation of acrylic resin.

The next experiment is a comparative example showing thermal curing of acrylic resin. MAINCOTE HYDUR was heated to 100 deg C for 5 minutes and the gel content was found to be 93%. Note that gel content of RF activated sample is higher even though it was exposed only for half the duration to energy.

This series of examples show that functionalized acrylic polymers can be activated under RF energy.

### C. *Polyester Resin*

In this series of experiments, the RF response of polyester/ vinyl ester resins was studied.

Aropol 7241, an isophthalic polyester resin from Ashland Chemical, was applied to a glass slide and the dried sample was exposed to RF field at 500 mV and 5 minutes. The gel content was found to be 51.3%.

This result shows that RF energy can activate an isophthalic polyester resin.

KELSOL 5293, a polyester dispersion from Reichhold Chemicals, was tested in this example. The composition included:

KELSOL 5293	2 parts
CYMEL 385 crosslinker	0.6 parts

The composition was exposed to RF for 2.5 minutes at 500 mV. The gel content was found to be 8.5%.

10% glycerin was added to the composition and exposed to RF for 2.5 minutes at 500 mV. The gel content was found to be 21.9%. This shows that addition of glycerin promotes RF activation.

5 The above composition (without glycerin) was exposed to 700 mV for 2.5 minutes and the gel content was found to be 73.7%. This result shows that exposure to higher RF field leads to higher gel content.

The composition comprising 10% glycerin was exposed to 700 mV for 2.5 minutes and the gel content was found to be 59.5%.

10 These experiments show that RF energy can be used to activate polyester type resins.

#### *D. Urethanes*

A linear polyurethane composition based on DESMODUR-W (an aliphatic diisocyanate from Bayer) and FORMREZ 11-36 (a polyester diol from Witco) was evaluated. The composition included:

15	DESMODUR W	0.75 parts
	Formerez 11-36	3.2 parts
	T-12 catalyst from Air Products & Chemicals	1-2 drops

20 A glass slide containing the above composition was exposed to 700 mV RF field for 5 minutes and the gel content was measured to be 11.4%. (Note: At 500 mV, the gel content was zero for 2.5 and 5 minute exposures without glycerin and 1.3% and 8.3% for 2.5 and 5 minute exposures with 10% glycerin)

10% Glycerin was added to the above composition and the RF activation repeated under the same conditions (700 mV and 5 minutes). The gel content was found to be 27%. This level of gel content is quite good for a linear polyurethane.

25 This result shows that addition of glycerin promotes urethane reaction and gel formation. It is very likely that hydroxyl groups present in the glycerin molecule is acting as reactive polyol in the formation of urethane. It may be possible to increase the gel content by increasing the ratio of isocyanate in the formulation relative to polyol. It may also be possible to increase the gel content

of the composition by partially replacing the diisocyanate (DESMODUR W) and polyester diol (Formerez 11-36) with multifunctional isocyanate such as polymeric MDI (methylene bisdiphenyldisocyanate) and triols. Use of multifunctional isocyanate and polyol should significantly increase gel content close to 100%. Use of isocyanate terminated prepolymer of higher molecular weight (8,000-10,000) in the urethane reaction may also increase gel content of the system.

**Example 10. Effect of "susceptor" addition on RF activation of acrylic and polyesters.**

The effect of adding 4-styrene sulfonic acid, Na salt, vinyl sulfonic acid, Na salt and A 35 D sulfonated polyester from Eastman Chemicals on RF activation of acrylic and polyester resins was evaluated. A first composition included:

MAINCOTE HYDUR-	95 parts
4-styrene sulfonic acid, Na salt	5 parts

The above composition was evaluated as described in previous examples at 700 mV and 2.5 minutes. The gel content was found to be 45.5%. Gel content of the sample without 4-styrene sulfonic acid, Na salt, under the same conditions was found to be 81.8% (see above).

10% Glycerin was added to the above composition and the sample evaluated under 700 mV and 2.5 minutes exposure conditions. The gel content was found to be 66.7%. Gel content of the sample without susceptor was 100% (see above).

This result shows that styrene sulfonate, Na salt does not promote the RF activation of acrylic resin, with and without glycerin.

A second composition included:

KELSOL 5243 polyester	2 parts
CYMEL	0.6 parts
Vinyl sulfonic acid, Na salt	
At 25% in water	0.5 parts



The above composition was evaluated as before at 700 mV and 2.5 minutes. The gel content was found to be 67.2%. Similar composition without susceptor had a gel content of 73.7% (see above). This result shows that addition of vinyl sulfonic, Na salt, does not promote RF activation of polyester resin.

5           10% Glycerin was added to the above composition. The resultant composition was evaluated as before under 700 mV and 1 second RF field. The sample became too hot and burst into flames. The result shows that glycerin does activate under high field and it is possible to get high degree of crosslinking reaction under very short times, say less than 1 second.

10           A third composition included:

                  MAINCOTE HYDUR acrylic                   1part

                  Eastman AD 35 D polyester susceptor           1 part

                  The above composition was evaluated as before and the gel content was found to be 79.2% at 700 mV and 2.6 minutes. The same composition without the  
15           susceptor had a gel content of 81.8% at 700 mV and 2.5 minutes exposure (see above). In this case addition of a susceptor does not have any effect on RF activation of acrylic polymer.

                  10% Glycerin was added to the third composition which was exposed to 700 mV for 2 minutes. This exposure led to a very violent reaction. This shows  
20           that susceptor was too active.

                  The third composition was exposed to 500 mV for 5 minutes. Gel content was found to be 68.4%. A comparable sample without the addition of susceptor was found to give a gel content of 61.5% after 2.5 minutes exposure (see above). The result shows that the susceptor had very little effect.

25           The third composition comprising 10% glycerin was evaluated at 500 mV and 5 minutes. The gel content was found to be 69.7%. The same sample with out susceptor had a gel content of 92.3% at 500 mV and 2.5 minute exposure (see above). The result shows that the addition of susceptor had a negative effect on RF activation.

It appears addition of known susceptors to the various thermoset resin compositions has very little impact on RF activation of the resins. In some cases, it seems to have a negative impact.

In a few cases, the heat generation is quite violent suggesting that proper tuning of frequency/ power/ time and other variables will lead to conditions that would allow very short cure times.

#### ***Example 14***

The use of the carboxyl containing diol dimethylol butanoic acid was tested as a susceptor. The composition included:

10	Formerez 11-36	3.2 parts
	DESMODUR W	0.75 parts
	Dimethylol Butanoic acid	0.28 parts
	T-12 catalyst	1-2 drops

15 No significant activation took place when this composition was exposed to 500 mV level (2.5 and 5 minute exposure) . At 5 minute exposure under 700 mV , the glass slide broke and no data could be gathered. As noted above, under similar conditions without the susceptor, a gel content of 11.4% was obtained in the absence of glycerin and 27% in the presence of glycerin.

20 It may be useful to add the acid diol in N-methyl pyrrolidone or another polar solvent and neutralize with a tertiary amine to protonate the acid. Further use of urethane prepolymer containing carboxyl or sulfonate groups in the presence of a tertiary amine (to protonate the acid) may be a better susceptor candidate for the urethane reaction.

#### ***Example 15***

25 This example demonstrates a method of selectively activating the compositions of the invention within a multi-layer stack of materials.

The composition comprised 70 wt% Eastman AQ35S sulfopolyester in 30 wt% glycerin. The composition was applied and dried down from an aqueous dispersion to form a continuous 0.003 inch thick film on one side of a bilaminate polyolefin material. The bilaminate polyolefin material comprised a single layer of polypropylene (PP) non-woven material bonded to a single layer of polyethylene (PE) film. The composition was coated onto the PP side of the bilaminate material. The coated bilaminate material was then interposed between two multi-layer stacks of un-coated bilaminate material to form a composite sandwich of materials. Each multi-layer stack had two layers of the un-coated bilaminate polyolefin material. The composite sandwich (410) was then placed directly over the RF probes and compressed under a TEFLON™ block at 30 psi. The composition was RF heated by applying approximately 1 kW of forward power into the tuned heat station 4122 for 200 milliseconds at approximately 13.5 MHz. After applying the RF energy to the composite sandwich, the pressure was removed and the sandwich was evaluated by slowly pulling the layers apart by hand. Every layer was easily pulled apart, with no observed bonding, except for the two layers that were in direct contact with the bonding composition. The two layers that were in direct contact with the bonding composition were firmly adhered by the bonding composition. As a control experiment, the experiment was repeated, except that no RF energy was applied to the composite sandwich. This experiment resulted in no observable bonding between any of the layers, including the two layers that were in direct contact with the bonding composition. It should be understood that in this experiment, the bonding composition was pre-applied to one surface of one of the layers of the composite sandwich. The bonding composition could be applied to more than one surface of more than one layer. Bonding would occur between any layers that are each in contact with a given layer of bonding composition.

### ***Example 16***

This example demonstrates the method of interfacing a carrier layer onto the surface of a susceptor layer to achieve an RF heatable composition. First, a 0.003 inch layer of a sulfonated polyester copolymer, Eastman AQ35S (Supplied by Eastman Chemical Company, Kingsport, TN) was coated out of an aqueous dispersion onto the polypropylene (PP) non-woven side of a bilaminate web consisting of a layer of PP non-woven bonded to a layer of polyethylene (PE) film. The coating was thoroughly dried down under a heat lamp and fan. A sandwich was made by placing a sample of the coated web against the PP side of a second piece of the same web material which was not coated, such that the coating was between the two webs. The sandwich was placed directly over the RF probes (410) of the RF set-up described in Figure 41. The distance between the RF probes and the sandwich was about 0.010 inch. The sandwich layers were pressed firmly together against the RF probes with 35 psi of applied pressure. About 1 kW of 13.5 MHz RF energy was applied for 500 milliseconds and resulted in no noticeable heating or bonding between the webs. Then the sandwich layers were separated and the susceptor coating was moistened with distilled water. The sandwich was re-assembled and RF energy was applied to it for 500 milliseconds as described above, resulting in very good bonding of the webs. As a control experiment, a sandwich consisting of two webs of the un-coated web material was prepared by moistening the PP side of each web and bringing the water moistened surfaces together. RF energy was applied for 500 milliseconds to the sandwich in the same way described above, resulting in no noticeable heating.

### ***Example 17***

This example demonstrates the effect of varying the concentration of the polar carrier in blends of the polar carrier and an ionomer. The polar carrier of this example is glycerin. Glycerin has a dielectric constant,  $\epsilon$ , of 42.5 at 25°C. The

ionomer of this example is a commercially available sulfonated polyester ionomer (Eastman AQ55S).

Several compositions were prepared as hot-melt blends of AQ55S and glycerin. The wt. % concentration of glycerin in the compositions was varied from 10% to 70. The compositions were prepared as follows:

Each composition was prepared to have a total mass of 50 grams. For each composition, the respective amounts of AQ55S pellets and glycerin were initially weighed into a resin flask and mixed to achieve thorough wetting of the resin pellets with the glycerin. The flask was then fit with a condenser column and sealed stir-assembly, and partially immersed into a 335F hot oil bath to achieve controlled heating and melting of the mixture. After the pellets became molten and swollen with the glycerin, the mixture was stirred and blended into a uniform composition.

Each composition was then applied in its molten state as a 0.003 inch thick x 1 inch wide x 5 inch long, continuous layer along the center line of a 4 inch wide x 0.0035 inch thick sheet of transparency film (3M PP2500 Transparency Film) and allowed to set-up at room temperature. Several such draw downs were made for each composition. A twin blade sample cutter was used to cut strips from the draw downs, by cutting across and perpendicular to the 5 inch long center line of each of the draw downs. This produced 1 inch wide x 4 inch long strips of acetate film, each with a 1 inch x 1 inch x 0.003 inch thick coating of composition in the center and 1½ inch long tails on each end. The resulting coatings differed in their relative RF-heating properties as well as their relative heat resistance to bond failure in a given shear loading condition.

RF-heating of each composition was evaluated as follows. For each composition, several sandwiches were prepared. Each sandwich was made by placing the polypropylene (PP) non-woven side of a 1 inch wide x 4 inch long strip of a bilaminate web against the coated side of the coated acetate test strip. The bilaminate web was composed of a layer of PP non-woven bonded to a layer of polyethylene (PE) film. Each sandwich was placed directly over the RF probes (410) of the RF set-up described in Figure 41, such that the uncoated side of the

acetate test strip was placed toward the probes. The sandwich layers were pressed firmly together against a layer of 0.010 inch thick layer of TEFLON™ and acetate that separated the RF probes and sandwich. A single pulse of 0.5 kW, 13.5 MHz RF energy was applied for a controlled duration to each sandwich. For each composition several sandwiches were activated, each at an incrementally longer duration. This gave a range of RF heating results. Threshold RF activation was determined from each range of results as the minimum duration that result in sufficient melting and wetting of the adhesive coating to the web to be observed by the naked eye. Threshold RF activation by the specific RF set-up (generally indicated in Figure 41) resulted in a narrow band of heating that was biased toward and parallel to the “high” probe of the probe assembly (602 or 604). This was because an “unbalanced” impedance matching network was used in the set-up.

Resistance to shear load bond failure was evaluated as follows. For each composition, bonded specimens were prepared. The specimens each consisted of a sandwich of a 1 inch x 4 inch x 0.0035 inch thick layer of acetate pressed against and hot-melt bonded to the coated side of a coated acetate test strip. (The coated acetate test strips were prepared as described earlier in this example.) Each hot-melt bond was facilitated by pressing the sandwich on a 275F hot plate surface under a 0.5 Kg load for 30 seconds, and then removing the sandwich and allowing it to cool and solidify into a bonded specimen. Each sandwich had a pair of “tails” of unbonded acetate on each side of a centered 1 inch x 1 inch bonded area of the sandwich. One tail from each of the two pairs and on opposite sides of the sandwich was cut off. This resulted in the final bonding specimen, consisting of two 1 inch x 3 inch layers of 0.0035 inch thick acetate bonded together across a 1 inch by 1 inch overlap by an interposed 0.003 inch thick layer of the composition being tested. The specimens were then placed under a shear load of 0.5 Kg in a temperature controlled chamber at 100F. The time required to result in total bond failure (disassembly of the specimen) at 100F was measured for each specimen and is referred to herein as “Shear Holding Time”.

The following observations were made:

- (1) As the percentage of glycerin was increased from 10% to 70%, a sharp increase in relative rates of RF heating began to occur at about 10% glycerin. (See Figure 54.)
- (2) As the percentage of glycerin was decreased from 70% to 10% a sharp increase in relative heat resistance began to occur at about 30% glycerin. (See Figure 55.)

### *Example 18*

This example demonstrates the effect of varying the concentration of the polar carrier in blends of the polar carrier and an alternative sulfonated polyester ionomer to the AQ55S of Example 17. The polar carrier of this example is glycerin. Glycerin has a dielectric constant of 42.5 at 25°C. The ionomer of this example is a commercially available sulfonated polyester ionomer (Eastman AQ35S)

Several compositions were prepared as hot-melt blends AQ35S and glycerin. The wt. % concentration of glycerin in the compositions was varied from 10% to 70%.

The compositions were prepared as follows:

Each composition was prepared to have a total mass of 50 grams. For each composition, the respective amounts of AQ35S pellets and glycerin were initially weighed into a resin flask and mixed to achieve thorough wetting of the resin pellets with the glycerin. The flask was then fit with a condenser column and sealed stir-assembly, and partially immersed into a 335F hot oil bath to achieve controlled heating and melting of the mixture. After the pellets became molten and swollen with the glycerin, the mixture was stirred and blended into a uniform composition.

RF-heating and resistance to shear load bond failure was evaluated for each composition as described in Example 17.

The following observations were made:

(1) As the percentage of glycerin was increased from 10% to 30%, a sharp increase in relative rates of RF heating began to occur at about 10% glycerin. (See Figure 56.)

5 (2) As the percentage of glycerin was decreased from 30% to 20% a sharp increase in relative heat resistance began to occur at about 30% glycerin. (See Figure 57.) These results agreed closely with the results of Example 17.

### *Example 19*

10 This example demonstrates the effects of dielectric constant and concentration of various polar carriers on the ability to achieve significantly improved RF activation times in compositions comprising blends of ionomers and polar carriers, as compared to compositions comprising the ionomer without sufficient presence of polar carrier.

15 The polar carriers and respective measured dielectric constants of this example are:

(1) Propylene carbonate;  $\epsilon = 62.67$  at  $25^{\circ}\text{C}$ .

(2) Glycerin;  $\epsilon = 42.5$  at  $25^{\circ}\text{C}$ .

(3) N-methyl-2-pyrrolidone;  $\epsilon = 32.2$  at  $20^{\circ}\text{C}$ .

(4) 1,2-propyleneglycol  $\epsilon = 32$  at  $25^{\circ}\text{C}$ .

20 (5) Polyethylene glycol 200;  $\epsilon = 17.70$  at  $23.5^{\circ}\text{C}$ .

(6) Benzoflex 9-88 (dipropylene glycol benzoate);  $\epsilon = 12.28$  at  $25^{\circ}\text{C}$ .

25 The ionomer of this example is a commercially available 30% solids aqueous dispersion of sulfonated polyester ionomer (Eastman AQ35D). Several compositions were prepared as aqueous mixtures of AQ35D and each of the polar carriers. The wt. % concentration of polar carrier in each of the compositions was varied from 0% up to 50%, where total weight is based on total weight of ionomer solids combined with total weight of polar carrier.



The compositions were prepared as follows:

Each composition was prepared to have a total mass of 50 grams. For each composition, the respective amounts of AQ35D ionomer dispersion and glycerin were initially weighed into a jar and mixed for about 10 minutes. The jars were sealed with tops until castings were made.

Each composition was then applied as a liquid at room temperature into castings onto a 0.0035 inch thick sheet of transparency film (3M PP2500 Transparency Film) and allowed to dry down into 0.003 inch thick coatings. The resulting coatings differed in their relative RF-heating properties. RF activation was evaluated as described in Example 17.

The following observations were made:

As the percentage of each polar carrier was increased from 0% to 50%, a sharp increase in relative rates of RF heating began to occur at about 10% glycerin (except for the composition that was prepared from Benzoflex 9-88, which experienced a relatively slow and gradual increase). (See Figure 58.)

While Benzoflex 9-88 gave a compatible composition with the AQ35S polymer, it resulted in a significantly less RF-active composition than any of the compositions that were prepared from more polar materials with relatively high dielectric constants. (See Figure 58.)

#### *Example 20*

This example demonstrates the effect of varying the concentration of a microcrystalline wax in the composition, X% (80% AQ55S / 20% Glycerin) / Y% wax. The microcrystalline wax in this example was PARICIN 220 [N-(2-hydroxyethyl)-12-hydroxystearamide].

The compositions were prepared as follows:

Each composition was prepared to have a total mass of 50 grams. A 300 gram batch of 80% AQ55S / 20% glycerin was prepared. 240 grams of AQ55S pellets and 60 grams of glycerin were initially weighed into a resin flask and mixed to achieve thorough wetting of the resin pellets with the glycerin. The flask was

then fit with a condenser column and sealed stir-assembly, and partially immersed into a 335F hot oil bath to achieve controlled heating and melting of the mixture. After the pellets became molten and swollen with the glycerin, the mixture was stirred and blended into a uniform composition. After a total of 4 hours of heating, the flask was removed from the hot oil bath. Several glass jars were each filled with 20 grams of the molten composition. Incrementally increasing amounts of PARICIN 220 were weighed into the hot contents of each jar, to result in a concentration series of X% (80%AQ55S / 20%Glycerin) / Y% PARICIN 220, where Y = 0, 1, 2, 3, 4, 5, 10, 15, 20, 25 and 30, and X = 100 - Y. Each open jar was placed in an oven at 300F for 30 minutes and allowed to become molten. The molten contents were then hand stirred with wooden stir sticks for 2 minutes to form a smooth and uniform blend.

Each composition was then applied in its molten state as a 0.003 inch thick x 1 inch wide x 5 inch long, continuous layer along the center line of a 4 inch wide x 0.0035 inch thick sheet of transparency film (3M PP2500 Transparency Film) and allowed to set-up at room temperature. Several such draw downs were made for each composition. A twin blade sample cutter was used to cut strips from the draw downs, by cutting across and perpendicular to the 5 inch long center line of each of the draw downs. This produced 1 inch wide x 4 inch long strips of acetate film, each with a 1 inch x 1 inch x 0.003 inch thick coating of composition in the center and 1½ inch long tails on each end.

The resulting coatings differed in their relative RF-heating properties and melt viscosities.

RF-heating was evaluated for each composition as described in Example 17. The Brookfield viscosity of each composition was measured at 275F, using an S27 spindle.

The following observations were made:

As the wt% of PARICIN 220 was increased from 0 to 10%, there was a slight increase (<5%) in the time required to heat each composition to the same degree as required at 0% PARICIN 220. As the wt% of PARICIN 220 was increased from 10% to 30%, there was a significant increase in the time required

to heat each composition to the same degree as required at 0% PARICIN 220.  
(See Figure 59.)

As the wt% of PARICIN 220 decreased from 10% to 0%, the melt viscosity at 275F increased by a factor of 6 from 6800 cP to 42000 cP.

5      ***Example 21***

10      This example demonstrates the effect of varying the concentration of the polar carrier in blends of the polar carrier and an ionomer, where the ionomer is the sodium salt of an ethylene acrylic acid copolymer. The polar carrier of this example is glycerin. Glycerin has a dielectric constant,  $\epsilon$ , of 42.5 at 25°C. The ionomer of this example is a commercially available aqueous dispersion of the sodium salt of an ethylene acrylic acid copolymer (MICHEM 48525P).

15      Several compositions were prepared as aqueous mixtures of MICHEM 48525P and glycerin. The wt. % concentration of glycerin in each of the compositions was varied from 0% up to 50%, where total weight is based on total weight of ionomer solids combined with total weight of glycerin.

    The compositions were prepared as follows:

20      Each composition was prepared to have a total mass of 50 grams. For each composition, the respective amounts of MICHEM 48525P ionomer dispersion and glycerin were initially weighed into a jar and mixed for about 10 minutes. The jars were sealed with tops until castings were made. Each composition was then applied as a liquid at room temperature into castings onto a 0.0035 inch thick sheet of transparency film (3M PP2500 Transparency Film) and allowed to dry down into 0.003 inch thick coatings. The resulting coatings differed in their relative RF-heating properties. RF activation was evaluated as described in Example 17.

25

    The following observations were made:

    As the percentage of each polar carrier was increased from 0% to 50%, a sharp increase in relative rates of RF heating began to occur at about 10%

glycerin (See Figure 61). This result agrees well with the results of Examples 17, 18 and 19.

### *Example 22*

This example demonstrates the relative heat resistance to bond failure in a given shear loading condition of four separate compositions that are composed of four different sulfonated polyesters respectively (AQ14000, AQ35S, AQ48S and AQ55S) and the same polar material in each case (glycerin). The polar carrier of this example is glycerin. Glycerin has a dielectric constant,  $\epsilon$ , of 42.5 at 25°C. The ionomers of this example are commercially available sulfonated polyester ionomers (Eastman AQ14000, AQ35S, AQ48S and AQ55S).

The four compositions were prepared to have 80 wt% ionomer / 20 wt% glycerin. Each composition was prepared to have a total mass of 50 grams. For each composition, the respective amounts of ionomer pellets and glycerin were initially weighed into a resin flask and mixed to achieve thorough wetting of the resin pellets with the glycerin. The flask was then fit with a condenser column and a sealed stir-assembly, and then partially immersed into a 335F hot oil bath to achieve controlled heating and melting of the mixture. After the pellets became molten and swollen with the glycerin, the mixture was stirred and blended into a uniform composition.

Each composition was then applied in its molten state as a 0.003 inch thick x 1 inch wide x 5 inch long, continuous layer along the center line of a 4 inch wide x 0.0035 inch thick sheet of transparency film (3M PP2500 Transparency Film) and allowed to set-up at room temperature. Several such draw downs were made for each composition. A twin blade sample cutter was used to cut strips from the draw downs, by cutting across and perpendicular to the 5 inch long center line of each of the draw downs. This produced 1 inch wide x 4 inch long strips of acetate film, each with a 1 inch x 1 inch x 0.003 inch thick coating of composition in the center and 1½ inch long tails on each end. The resulting coatings were each evaluated for their relative RF-heating properties as well as their relative heat

resistance to bond failure in a given shear loading condition, as described in Example 17.

The following observations were made for the four compositions:

80% AQ14000 / 20% Glycerin

T<sub>g</sub> of AQ14000 = 7 °C

Threshold RF Activation Time = 130 ms

Shear Holding Time = 1,604 sec

80% AQ35S / 20% Glycerin

T<sub>g</sub> of AQ35S = 35 °C

Threshold RF Activation Time = 310 ms

Shear Holding Time = 68,252 sec

80% AQ48S / 20% Glycerin

T<sub>g</sub> of AQ48S = 48 °C

Threshold RF Activation Time = 90 ms

Shear Holding Time = 40,346 sec

80% AQ55S / 20% Glycerin

T<sub>g</sub> of AQ55S = 55 °C

Threshold RF Activation Time = 100 ms

Shear Holding Time = 1,450,000 sec

### *Example 23*

This example demonstrates a hot melt composition prepared from a sulfonated polyester ionomer (AQ55S) and a polar plasticizer (RIT-CIZER #8). The composition was prepared to have 80 wt% ionomer / 20 wt% RIT-CIZER #8. The composition was prepared to have a total mass of 50 grams. The respective amounts of ionomer pellets and glycerin were initially weighed into a resin flask and mixed to achieve thorough wetting of the resin pellets with the

glycerin. The flask was then fit with a condenser column and a sealed stir-assembly, and then partially immersed into a 335F hot oil bath to achieve controlled heating and melting of the mixture. After the pellets became molten and swollen with the glycerin, the mixture was stirred and blended into a uniform composition. The composition was then applied in its molten state as a 0.016 inch thick x 1 inch wide x 1 inch long, continuous layer along the center line of a 4 inch wide x 0.0035 inch thick sheet of transparency film (3M PP2500 Transparency Film). The resulting coating was evaluated for relative RF-heating as described in Example 17.

The following observations were made for the composition:

The composition was very thick and stiff at 335 °F. It was not possible to measure the Brookfield viscosity at 275 °F. At room temperature, the composition was clear, tough and brittle. There seemed to be very good compatibility between the polymer and RIT-CIZER #8. The threshold RF activation time was measured to be approximately 4 seconds.

#### ***Example 24***

This example demonstrates a composition that comprises an ionomer-type susceptor, a polar material and an adhesive compound. First, several different RF susceptor compositions were prepared by blending various ionomers and polar material. Then, each of the RF susceptor compositions were blended with an adhesive compound.

#### ***Preparation of the RF-susceptor compositions:***

Several different RF susceptor compositions were prepared by blending various commercially available sulfonated polyester ionomers (Eastman AQ35S, AQ48S and AQ55S, AQ1045, AQ1350, AQ14000) with a polar material (glycerin). The RF-susceptor compositions of this example include but are not limited to:

70 wt % AQ35S / 30 wt % Glycerin

70 wt % AQ48S / 30 wt % Glycerin

70 wt % AQ55S / 30 wt % Glycerin

70 wt % AQ1045 / 30 wt % Glycerin

70 wt % AQ1350 / 30 wt % Glycerin

70 wt % AQ14000 / 30 wt % Glycerin.

Each RF-susceptor composition was prepared to have a total batch mass of 300 grams. For each composition, the respective amounts of ionomer and glycerin were initially weighed into a resin flask and mixed to achieve thorough wetting of the resin pellets with the glycerin. The flask was then fit with a condenser column and sealed stir-assembly, and partially immersed into a 335F hot oil bath to achieve controlled heating and melting of the mixture. After the polymer became molten and swollen with the glycerin, the mixture was stirred and blended into a uniform composition. The compositions that comprised linear polymers (AQ35S, AQ48S and AQ55S) were each blended at 335F for 3 hours. The composition comprising AQ1045 was blended at 335F for 1 hour. The compositions comprising AQ1350 and 14000 were each blended at 335F for 1.5 hours. Each of the RF-susceptor compositions was cooled and stored at room temperature for later use.

***Preparation of the compositions comprising blends of RF-susceptor compositions and an adhesive compound:***

Each of the RF susceptor compositions was blended with an adhesive compound. The adhesive compound of this example is a random copolymer of ethylene vinyl acetate (EVA). The commercially available EVA that was used is DuPont Polymer's ELVAX 210, Lot #90204492.

Each composition was prepared to have a total mass of 17 grams. For each composition, 7 grams of ELVAX 210 and 10 grams of the respective RF-susceptor composition was added to a glass jar at room temperature. The open jar was then heated in a convection oven at 335F for 40 minutes. After 40

minutes of heating, the jar was removed from the oven to the surface of a 330F hot plate and stirred by hand for 1 minute to result in a uniform smooth blend.

A total of six compositions were prepared. The RF-susceptor / Adhesive compositions of this example include but are not limited to:

- A. 41 wt % AQ35S / 18 wt % Glycerin / 41 wt % ELVAX 210
- B. 41 wt % AQ48S / 18 wt % Glycerin / 41 wt % ELVAX 210
- C. 41 wt % AQ55S / 18 wt % Glycerin / 41 wt % ELVAX 210
- D. 41 wt % AQ1045 / 18 wt % Glycerin / 41 wt % ELVAX 210
- E. 41 wt % AQ1350 / 18 wt % Glycerin / 41 wt % ELVAX 210
- F. 41 wt % AQ14000 / 18 wt % Glycerin / 41 wt % ELVAX 210

***Evaluation of the blends of RF-susceptor compositions with ELVAX 210.***

Immediately after stirring the composition into a uniform blend, each composition was then applied in its molten state as a 0.003 inch thick x 1 inch wide x 5 inch long, continuous layer along the center line of a 4 inch wide x 0.0035 inch thick sheet of transparency film (3M PP2500 Transparency Film) and allowed to set-up at room temperature. Several such draw downs were made for each composition. A twin blade sample cutter was used to cut strips from the draw downs, by cutting across and perpendicular to the 5 inch long center line of each of the draw downs. This produced 1 inch wide x 4 inch long strips of acetate film, each with a 1 inch x 1 inch x 0.003 inch thick coating of composition in the center and 1½ inch long tails on each end.

The resulting coatings were each evaluated for their relative coat properties as well RF-heating properties, as described in Example 17.

The following observations were made for the six compositions:



**Table 11**

Composition	Coating Properties				
	Toughness	Clarity	Color	Tackiness	RF Activation Time (ms)
A	Soft	Translucent	White	Slight Tack	520
B	Tough	Translucent	White	Tacky	100
C	Very Tough	Translucent	White	Very Slight Tack	280
D	Very Soft	Clear	None	Tacky	430
E	Soft	Clear	None	Tacky	380
F	Soft	Clear	None	Tacky	340

**Example 25**

This example demonstrates compositions comprising an ionomer, a polar material and various low molecular weight polyolefin additives.

First, an RF heatable hot melt composition was prepared by blending 70 wt% AQ35 (a sulfonated polyester, commercially available from Eastman Chemical Company) with 30 wt% glycerin for about 3 hours at 335F. Then, several compositions were prepared by blending small samples of the molten AQ35/glycerin blend, separately with various grades of EPOLENE (low molecular weight polyolefins, commercially available from Eastman Chemical Company).

The polyolefin polymers of this example are Eastman Chemical's: EPOLENE N-10 (lot #11478), EPOLENE N-11 (lot #89352), EPOLENE N-14 (lot #12877), EPOLENE N-15 (lot #491104), EPOLENE N-20 (lot #87023), EPOLENE N-21 (lot #13018), and EPOLENE N-34 (lot #12710). EPOLENE polymers are low molecular-weight polyolefins that can be useful as base polymers for hot-melt adhesives.

Each composition was then applied in its molten state as a 0.003 inch thick x 1 inch wide x 5 inch long, continuous layer along the center line of a 4 inch wide x 0.0035 inch thick sheet of transparency film (3M PP2500 Transparency Film) and allowed to set-up at room temperature. Several such draw downs were made for each composition. A twin blade sample cutter was used to cut strips from the draw downs, by cutting across and perpendicular to the 5 inch long center line of

each of the draw downs. This produced 1 inch wide x 4 inch long strips of acetate film, each with a 1 inch x 1 inch x 0.003 inch thick coating of composition in the center and 1½ inch long tails on each end.

The resulting coatings were each evaluated for their relative RF-heating properties as well as their relative heat resistance to bond failure in a given shear loading condition, as described in Example 17.

Table 12 summarizes the observations that were made for the various compositions:

Table 12

70% AQ35 / 30% Glycerin + 5% EPOLENE				
EPOLENE#	mw	viscosity	rftime	hangtime
N-10	10000	8675	210	3.91
N-11	6000	7450	210	2.99
N-14	4000	7750	220	4.77
N-15	12000	13500	210	2.60
N-20	15000	10020	220	6.01
N-21	6500	6125	210	1.91
N-34	6200	8100	210	2.95
70% AQ35 / 30% Glycerin + 10% EPOLENE				
EPOLENE#	mw	viscosity	rftime	hangtime
N-10	10000	10220	250	3.30
N-11	6000	7975	240	1.94
N-14	4000	8725	250	3.33
N-15	12000	17900	250	1.46
N-20	15000	9450	240	2.28
N-21	6500	7112	240	1.59
N-34	6200	8212	240	1.78
70% AQ35 / 30% Glycerin + X% EPOLENE N-10				
%EPOLENE		viscosity	rftime	hangtime
0		6362	200	4.46
2.5		8337	210	2.10
5		8675	210	3.91
10		10220	250	3.30
15		11570	280	5.03
20		12250	280	5.99
25		14620	300	7.55
30		15250	825	5.35

**Example 26**

This example demonstrates a series of compositions that comprise: 9% polyethylene glycol and 91% (75% AQ55 / 25% glycerin).

5 First, a blend of 75% AQ55 and 25% glycerin was made by blending AQ55 and glycerin for 3 hours at 335F. Then, a series of compositions was prepared in which each composition was prepared as a molten blend of 9% polyethylene glycol (PEG) and 91% (75% AQ55 / 25% glycerin).

10 Each composition was then applied in its molten state as a 0.003 inch thick x 1 inch wide x 5 inch long, continuous layer along the center line of a 4 inch wide x 0.0035 inch thick sheet of transparency film (3M PP2500 Transparency Film) and allowed to set-up at room temperature. Several such draw downs were made for each composition. A twin blade sample cutter was used to cut strips from the draw downs, by cutting across and perpendicular to the 5 inch long center line of each of the draw downs. This produced 1 inch wide x 4 inch long strips of acetate  
15 film, each with a 1 inch x 1 inch x 0.003 inch thick coating of composition in the center and 1½ inch long tails on each end. The resulting coatings were each evaluated for their relative RF-heating properties as well as their relative heat resistance to bond failure in a given shear loading condition, as described in Example 17.

20 Table 13 summarizes the observations that were made for the various compositions:

**Table 13**

PEG # (PEG200 through PEG8000)	Brookfield Viscosity (cP at 275F)	tack 1= very slight tack; 2=slight tack; 3=tacky	Rftime time required to melt sample (ms).	hangtime hrs for 1 sq. inch bond area to fail at 100F under a 0.5 kg shear load.
200	15650	1	130	12.94
300	12500	1	150	11.35
400	14600	1	130	5.34
600	13700	3	140	7.23
900	12250	1	150	5.76
1000	12800	1	150	6.82
1450	11700	1	210	4.85
3350	15070	1	200	5.70
4000	14620	2	250	5.51
4600	16400	2	220	9.34
8000	17320	1	230	6.35

**Example 27**

This example demonstrates a composition comprising 10% IGEPAL (a commercially available additive from Rhodia) and 90% (75% AQ55 / 25% glycerin).

A first composition comprising 75% AQ55 and 25% glycerin was prepared by blending AQ55 and glycerin for 6 hours at 335F. A second composition was prepared by blending IGEPAL CO-880 at 10 wt% with a sample of the first composition.

Each composition was then applied in its molten state as a 0.003 inch thick x 1 inch wide x 5 inch long, continuous layer along the center line of a 4 inch wide x 0.0035 inch thick sheet of transparency film (3M PP2500 Transparency Film) and allowed to set-up at room temperature. Several such draw downs were made for each composition. A twin blade sample cutter was used to cut strips from the draw downs, by cutting across and perpendicular to the 5 inch long center line of each of the draw downs. This produced 1 inch wide x 4 inch long strips of acetate film, each with a 1 inch x 1 inch x 0.003 inch thick coating of composition in the center and 1½ inch long tails on each end.

The resulting coatings were each evaluated for their relative RF-heating

properties as well as their relative heat resistance to bond failure in a given shear loading condition, as described in Example 17.

Table 14 summarizes the observations that were made for the various compositions:

**Table 14**

Composition	Viscosity (cP at 275F)	RF Activation Time Time required to melt the sample (ms).
75% AQ55 / 25% glycerin	28,200	180
90% (AQ55 / glycerin) / 10% IGEPAL CO-880	9,750	360

**Example 28**

This example demonstrates an RF heatable composition comprising 75% AQ48 (a commercially available sulfonated polyester from Eastman Chemical Company) and 25% glycerin.

The composition was prepared by blending 75 wt% AQ48 with 25 wt% glycerin for 4 hours at 335F. The resulting molten composition was fluid and clear. When this composition was cast onto layers of acetate and allowed to cool, the resulting solid draw-downs were clear and had cold-tack. This composition is ideal for applications where parts are to be initially adhered with a green strength bond by the composition and subsequently fused by the heat that is generated from within the composition as it is exposed to RF energy.

The molten composition had a Brookfield viscosity of 5,750 cP at 275F, using an S27 spindle at 20 RPM. The composition was then applied in its molten state as a 0.003 inch thick x 1 inch wide x 5 inch long, continuous layer along the center line of a 4 inch wide x 0.0035 inch thick sheet of transparency film (3M PP2500 Transparency Film) and allowed to set-up at room temperature. Several such draw downs were made. A twin blade sample cutter was used to cut strips from the draw downs, by cutting across and perpendicular to the 5 inch long center line of each of the draw downs. This produced 1 inch wide x 4 inch long

strips of acetate film, each with a 1 inch x 1 inch x 0.003 inch thick coating of composition in the center and 1½ inch long tails on each end. The resulting coatings were each evaluated for their RF-heating properties, as described in Example 17. RF activation was achieved in 160 ms.

5           The composition was then drawn into flat beads (0.10 inches wide by 0.01 inches thick at the maximum thickness—the beads were crowned in the middle and feathered at the edges). Three sandwiches of materials were made. Each sample was made by placing a single bead of the composition between two identical layers of thin-film bilaminate polyolefin material. Each layer of bilaminate material was  
10           composed of two layers - one layer of polypropylene non-woven (PP) and one layer of polyethylene film (PE).

          The first sandwich (sample 1) was assembled such that the bead was in direct contact with the PP side of one of the layers of bilaminate, and the PP side of the other layer of bilaminate. The second sandwich (sample 2) was assembled  
15           such that the bead was in direct contact with the PP side of one of the layers of bilaminate, and the PE side of the other layer of bilaminate. The third sandwich (sample 3) was assembled such that the bead was in direct contact with the PE side of one of the layers of bilaminate, and the PE side of the other layer of bilaminate.

20           In each case, the bead had slight tack and was able to gently hold the layers of the sandwich together. Each sandwich was then activated in a 13.5 MHz RF field for 200 ms at 1000 watts. In each case, melting of the bilaminate layers had occurred. Then the sandwiches were each immersed and washed in MEK for several minutes in order to remove the adhesive from the bond line. In each case,  
25           after washing the adhesive from the sandwich, residual bonding was observed between all layers of the sandwich in the areas where melting had been observed.

          While various embodiments of the present invention have been described above, it should be understood that they have been presented by way of example only, and not limitation. Thus, the breadth and scope of the present invention  
30           should not be limited by any of the above-described exemplary embodiments, but should be defined only in accordance with the following claims and their

equivalents. Additionally, all patents, patent applications and publications mentioned above are incorporated by reference herein.